METABOLIC ENGINEERING OF Escherichia coli TO

PRODUCE 1,3-PROPANEDIOL DEVOID OF A

PLASMID EXPRESSION SYSTEM



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วิศวกรรมเมทาบอลิกของเชื้อเอสเชอริเชีย คอไล เพื่อการผลิต 1,3-โพรเพนไดออล โดยไม่ใช้ระบบการแสดงออก ของยีนด้วยพลาสมิด



วิทยานิพนธ์นี้เป็นส่วนหนึ่งของการศึกษาตามหลักสูตรปริญญาวิทยาศาสตรดุษฎีบัณฑิต สาขาวิชาเทคโนโลยีชีวภาพ มหาวิทยาลัยเทคโนโลยีสุรนารี ปีการศึกษา 2563

METABOLIC ENGINEERING OF *Escherichia coli* TO PRODUCE 1,3-PROPANEDIOL DEVOID OF A PLASMID

EXPRESSION SYSTEM

Suranaree University of Technology has approved this thesis submitted in partial fulfillment of the requirements for the Degree of Doctor of Philosophy.

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นนทพร หว่อง : วิศวกรรมเมทาบอลิกของเชื้อเอสเซอริเซีย กอไล เพื่อการผลิค 1,3-ไพรเพนไดออล โดยไม่ไข้ระบบการแสดงออกของอินด้วยพลาสมิค (METABOLIC ENGINEERING OF *Escherichia coli* TO PRODUCE 1,3-PROPANEDIOL DEVOID OF A PLASMID EXPRESSION SYSTEM) อาจารย์ที่ปรึกษา : รองศาสตราจารย์ ดร.เขมวิทย์ จันด๊ะมา, 138 หน้า

โอเปอรอนกลีเซอรอล ดีโอคราเทส (gdrAB-dbaB123) จากเชื้อ Kiebsiella preumoniae และอิน 1.3-โพรเพนไดออล ออกซิโดรีดักเทส (vqhD) ที่ขึ้นกับ NADPH จาก Escherichia coli อูก แทรกบนโครโมโรมดีเอ็นเอของ E. coli และอูกแสดงออกภายได้ไปรโมเดอร์ IdhA และ gfB ของ E. coli ตามลำดับ โดยเชื้อ E. coli NSK015 (ΔidhA:gdrAB-dhaB123 ΔackA::FRT ΔgfB::yqhD ΔfrdABCD::cat-sacB) ที่ถูกพัฒนาขึ้นผลิต 1.3-โพรเทนไดออล (1.3-PDO) ได้ที่ความเข้มขึ้น 36.8 กรัมต่อลิตร และมีผลผลิตที่ 0.99 โมลด่อ โมลของกลีเซอรอล การทดสอบผลการ ผลิต โดยใช้สารตั้ง ดันร่วมกันระหว่างกลีเรอรอลและแป้งมันสำปะหลังเพื่อผลิตสาร 1.3-โพรเทนไดออล พบว่าเชื้อนี้ สามารถผลิตสาร 1.3-โพรเพนไดออล ได้ที่ความเข้มขึ้น 31.9 กรัมต่อลิตรและ มีผลการผลิตที่ 0.84 โมลต่อ โมลของกลีเซอรอลตามลำดับ งานวิทยานีพนธ์นี้เป็นงานบุกเบิกที่แสดงการผลิต 1.3-โพ รเพนไดออล อย่างมีประสิทธิภาพโดยทำให้เกิดกรรแสดงออกของยินจากสิ่งมีชีวิตอื่นบนจีโนมของ E. coli โดยที่ไม่ได้ใช้ระบบพลาสมิคในการแสดงออกของยินจากสิ่งมีชีวิตอื่นบนจีโนมของ E. coli โดยที่ไม่ได้ใช้ระบบพลาสมิคในการแสดงออกของยินจากสิ่งมีชีวิตอื่นบนจีโลมของ E. coli โดยที่ไม่ได้ใช้ระบบพลาสมิคในการแสดงออกของยินจากสิ่ง ยาปฏิชีวนะ IPTG และ สารอาหารที่ครบถ้วนไม่ถูกนำมาให้ในการผลิต 1.3-โพรเพนไดออล

เพื่อเพิ่มอัคราการผลิค การผลิก 1,3-ไพรเพนไดอออภายใต้สถาวะที่เหมาะสมถูกศึกษาใน E. colt NSK015 ซึ่งพบว่าสภาวะที่ดีที่สุดคือความเร็วในการกาน 300 รอบต่อนาทีและความเข้มข้น ด้าของโดเอนไขม์บี 12 ที่ 7.5 ไมโคร โมอาร์ โดยที่อัตราการผลิคเพิ่มขึ้นจาก 0.34 กรัมต่อลิตรต่อ ชั่วโมง เป็นที่ 0.79 กรัมต่อลิครต่อชั่นโมง ต่อมาเพื่อที่จะปรับปรุงกวามเข้มข้นของ 1,3-โพรเพนได ออลให้สูงขึ้นจึงได้ทำการหมักแบบทึ่งกะโดยที่มีการปรับเปลี่อนรูปแบบการป้อนสารตั้งต้นร่วมของ กลูโดสและกลีเซอรอล ผลการทดลองแสดงให้เห็นว่าการใช้รูปแบบการให้แบบแบ่งเป็นจังหวะ อย่างต่อเนื่องมีประสิทชิภาพอย่างมากในการผลิค 1,3-โพรเพนไดออล ความเข้มข้นของ 1,3-โพรเพนไดออลเพิ่มขึ้นถึง 132.63 % จาก 38.1 กรัมต่อลิตร เป็น 88.6 กรัมต่อลิครกอในเวลา 144 ชั่วโมง ด้วยอัตราการผลิคในการหมักแบบกึ่งกะที่อัตราการผลิต 0.62 กรัมต่อลิตรต่อชั่วไมง

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สาขาวิชาเทคโนโลยีชีวภาพ ปีการศึกษา 2563

NONTHAPORN WONG : METABOLIC ENGINEERING OF *Escherichia coli* TO PRODUCE 1,3-PROPANEDIOL DEVOID OF A PLASMID EXPRESSION SYSTEM. THESIS ADVISOR : ASSOC. PROF. KAEMWICH JANTAMA, Ph.D., 138 PP.

1,3-PROPANEDIOL/E. COLI/ K. PNEUMONIAE/GLYCEROL/CASSAVA

Glycerol dehydratase (*gdrAB-dhaB123*) operon from *Klebsiella pneumoniae* and *Escherichia coli* NADPH-dependent 1,3-propanediol oxidoreductase (*yqhD*) were stably integrated on the chromosomal DNA of *E. coli* and their overexpression was under the control of the native-host *ldhA* and *pflB* promoters, respectively. A developed *E. coli* NSK015 ($\Delta ldhA::gdrAB-dhaB123 \Delta ackA::FRT \Delta pflB::yqhD$ $\Delta frdABCD::cat-sacB$) produced 1,3-propanediol (1,3-PDO) at the level of 36.8 g/L with a yield of 0.99 mol/mol glycerol. Co-substrate of glycerol and cassava starch was also utilized for 1,3-PDO production with the obtained concentration and yield of 31.9 g/L and 0.84 mol/mol glycerol respectively. This represents a pioneer work for efficient 1,3-PDO production in which the overexpression of heterologous genes on the *E. coli* host genome devoid of plasmid expression systems. Plasmids, antibiotics, IPTG, and rich nutrients were omitted during 1,3-PDO production.

To improve productivity, 1,3-PDO production by *E. coli* NSK015 was optimized. After performing optimization, the best condition for 1,3-PDO production was obtained at 300 rpm agitation with a low concentration of coenzyme B12 of 7.5 μ M. The 1,3-PDO productivity was enhanced from 0.34 g/L/h up to 0.79 g/L/h. To further improve 1,3-PDO production, the fed-batch fermentation mode was carried out

by varying different feeding modes of the co-substrate of glucose and glycerol. The results showed that the use of the continuous-pulsed mode was more powerful to produce 1,3-PDO. After 144 h incubation, 1,3-PDO concentration was improved from 38.1 g/L to 88.6 g/L, which is an 132.6 % with the 1,3-PDO productivity at 0.62 g/L/h.



School of Biotechnology

Student's Signature KUNNT MOD

Academic Year 2020

Advisor's Signature // Mmtama

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LIST OF ABBREVIATIONS

ackA	=	Acetate kinase gene
adhE	=	Alcohol dehydrogenase gene
budA	=	α -acetolactate synthase gene
budB	=	α-acetolactate decarboxylase gene
C. beijerinckii	=	Clostridium beijerinckii
C. butyricum	- H	Clostridium butyricum
C. diolis	- H	Clostridiu <mark>m d</mark> iolis
DHAK		Dihydroxyacetone kinase
DHAT	SB	NADH-dependent 1,3-propanediol
		oxidoreductase gene
dhaKLM		Dihydroxyacetone kinase genes
dhaD	้ ^อ ุทยาลัย	Glycerol dehydrogenase gene
dhaR	=	Transcriptional regulator gene
dhaT	=	NADH-dependent 1,3-propanediol
		oxidoreductase gene
dhaB	=	Glycerol dehydratase gene (large subunit)
dhaC	=	Glycerol dehydratase gene (medium subunit)

LIST OF ABBREVIATIONS (Continued)

dhaE	=	Glycerol dehydratase gene (small subunit)
frdABCD	=	Fumarate reductase genes
gldA	=	Glycerol dehydrogenase gene
glpK	=	Glycerol kinase gene
GDHt	=	Glycerol dehydratase
3-HPA	=	3-Hydroxypropionaldehyde
3-HP	- 4	3-Hydroxypropionic acid
HOR	- <i>H</i>	Hypothetical oxidoreductase
K. pneumoniae	=	Klebsiella pneumoniae
K. oxytoca	日常	Klebsiella oxytoca
ldhA		Lactate dehydrogenase gene
L. collinoides		Lactobacillus collinoides
orfX	้จิ๊กยาลัย	Reactivating factor or gdrB (Small unit)
orfZ	=	Reactivating factor or <i>gdrA</i> (Large unit)
orfW	=	Putative gene
1,3-PDO	=	1,3-Propanediol
pdu	=	Gene relevant to diol dehydratase or
		propanediol utilization pathway
pflB	=	Pyruvate formate lyase

LIST OF ABBREVIATIONS (Continued)



CHAPTER I

INTRODUCTION

1.1 Background and scope of the study

Oil and natural gas products are also essential parts of living life and affect the economic growth throughout the world unavoidably. It is responsible for being a core energy source used in several types of transportation, including the generation of electricity. Furthermore, petroleum is also transformed into a wide range of chemicals, which are an initial raw material for a tremendous quantity of household products, for example, toothpaste, shampoo, clothes, cosmetics, pharmaceutical products, tools in the kitchen, electric appliances, laptop, cell phone, plastic bags, pipe, shoes, construction materials, and medical materials. Thus, all these petroleum products touch us everywhere. Its intervention does not make us only comfortable and easy, but also contribute the infinity global development.

As our planet encounters global warming due to the greenhouse effect released from a variety of industries, particularly petrochemical plant. Releasing an enormous amount of CO₂ and CH₄ directly influences the ozone atmosphere as mentioned above, leading to numerous problems, such as an increase in temperature, and climate change. Nevertheless, the fossil fuel resource is not renewable. As it is used, it is going to run out and cannot be replenished. Eventually, it causes the fluctuation of crude oil prices worldwide and impacts other petrochemical sections in a chain reaction. All of these become major factors to drive the scientist to concern and develop the sustainable greenway for replacing the non-renewable resource. Therefore, renewable energy, for example, wind power, waterpower, solar power, geothermal power, biomass fuel, is gaining attractiveness and widely used in many trial places and vehicles.

Biodiesel is one of renewable energy and is employed in vehicles for the replacement of petroleum diesel. It is capable of production from vegetable oils or animal fats by the reaction of transesterification. In this process, 10% w/v glycerol is always generated as the main byproduct (Kaur et al., 2012). In 2015, the global biodiesel production is approximately produced 30.1 billion liters (REN21, 2016). Thus, three-hundred million liters of glycerol are generally generated each year according to the increasing biodiesel demand. However, glycerol has diverse benefits and used as an ingredient in many related products, such as cosmetics, pharmaceuticals, food, and chemical synthesizing process (Hájek and Skopal, 2010). Importantly, it also serves as a carbon substrate for conversion to the other value-added chemicals and 1,3-PDO is one of those chemicals.

1,3-propanediol or 1,3-PDO used as a monomer for producing polytrimethylene terephthalate or PTT, which is one of polyester that further used in the industries of apparel and carpet mostly. Conventionally, 1,3-PDO is predominantly synthesized from the petrochemical industry. As the increasing greenhouse effects from the fuel combustion in plants, and the usage expedition of alternative white energy, these factors support the target chemical production through the renewably biological process by using raw material from nature, agricultural residuals, and industrial wastes. As a result, a large amount of glycerol derived from rising biodiesel demand is gaining more attractive for adding value to glycerol, especially crude glycerol. Therefore, the utilization of crude glycerol is worthwhile for investment and can reduce the cost of biodiesel production, including addressing the surplus of by-product properly (Della Pina et al., 2011; Werpy and Petersen, 2004).

Although most of the 1,3-PDO biological production has been commercialized by DuPont and well known that they use a best developed recombinant E. coli for producing 1,3-PDO, it seemed recently that the new companies of bio-based 1,3-PDO production have been continuously established in France and China with the company of METabolic EXplorer S.A (METAX) and Zhangjiagang Glory Biomaterial, respectively. This indicates that 1,3-PDO is attracted from many scientists so far by using both native producers and non-native producers. For example, co-cultivation of two *E. coli* strains containing *dhaB12* and *dhaT* produced 1,3-PDO at 41.65 g/L in a 10 L fermenter (Yun et al., 2021). Lee et al. (2018) also constructed the E. coli BL21(DE3) for producing 1,3-PDO from glucose by deleting genes involving conversion of glycerol to dihydroxyacetone phosphate (DHAP) (glpABCD, glpK) and overexpressed the genes related to the conversion of DHAP to glycerol (gpd1, gpp2). The final yield was obtained from glucose at 0.33 mol/mol. However, the result of 1,3-PDO concentration was still low at around 2-3 g/L. Recently, K. pneumonia J2B was metabolically engineered with 33 genetic modifications to produce 1,3-PDO from glucose. The strain could

produce 62 g/L 1,3-PDO with a yield of 1.2 mol/mol under the fed-batch fermentation. (Lama et al., 2020). This study disclosed the method clearly how to engineer *K*. *pneumonia* for producing 1,3-PDO from glucose. This strain takes advantages of using glucose that presents in a huge amount of agricultural resource. However, *K*. *pneumoniae* is capable of pathogenicity that may bring risks to the environment and health when it is used in industrial scales. Unlike *E. coli*, most of the works tried overexpressing heterologous genes on plasmids. The concentration of 1,3-PDO was lower and fluctuated in the industry scale because of the plasmid instability.

1.2 Objectives

E. coli C was selected and engineered to produce 1,3-PDO from glycerol under aerobic conditions. Genes *ldhA*, *ackA*, *pflB* and *frdABCD* encoding lactate dehydrogenase, acetate kinase, pyruvate formate lyase, and fumarate reductase, respectively, were interrupted to eliminate by-products. Genes, *dhaB1*, *dhaB2*, *dhaB3*, *grdA* and *gdrB* originated from *K. pneumoniae* were rearranged in one operon and integrated in front of the *ldhA* host promoter. *E. coli* expressing the functional glycerol dehydratase under the native promoter was then initially evaluated the aldehyde production on the modified indicator plate. The *yqhD* gene was also integrated in frame with the *pflB* promoter to enhance the conversion of 3-hydroxypropionaldehyde (3-HPA) to 1,3-PDO. Finally, 1,3-PDO production by the last developed *E. coli* strain NSK015 was optimized to find the best parameters that enhanced 1,3-PDO production efficacy from co-substrate of glucose and glycerol.



CHAPTER II

LITERATURE REVIEWS

2.1 1,3-propanediol and its beneficial importance

Although, alcohol has been well known for several thousand years, 1,3propanediol (1,3-PDO) is one of the chemical products generated from fermentation and known for several centuries. Generally, 1,3-PDO is a colorless viscous liquid, having specific gravity at 1.054 g/cm³ and lower than that of glycerol at 1.26 g/cm³. It can be used as a component in several products in the market such as cosmetics, lubricant, antifreeze, adhesives, detergents (Kaur et al., 2012). In the past, 1,3-PDO was characterized only in the group of a niche market, but recently 1,3-PDO was identified as a commodity interesting chemical. By considered the global market of 1,3-PDO, it has been forecasted to reach the market price of USD 691 million by 2025. (1,3-Propanediol (PDO) Market, 2020).

Increasing demand of 1,3-PDO in the chemical trade is the result of the final end-product. Especially, polytrimethylene terephthalate (PTT), which has outstanding properties in polyester. This polymer has attracted the attention of many scientists. The major characteristics are derived from elasticity, tolerance, strength, stiffness, and stain resistance, which take advantage of PBT (Polybutylene terephthalate) or PET (Polyethylene terephthalate). These features make PTT become several types of target products and can practically be applied in many fields of industries either materials used for medical product, cloth or thermoplastic (Liu et al., 2010).

Besides, 1,3-PDO is also used as a component for producing a wide range of commodities such as solvent, fragrance, sanitary product, cosmetic, fiber, laminate, and so on (Fig. 2.1). In some cases, 1,3-PDO is used as an ingredient to produce the biocide for example, 2-Bromo-2-nitro-1,3-PDO. This chemical is used for preventing the accumulation of microorganism in the circulating water system (Saxena et al., 2009). 1,3-PDO is thus attracted several scientists to develop its microbial production process for replacing petroleum-based production. However, DuPont is still nowadays a main predominant company that produces 1,3-PDO by using engineered *E. coli* and glucose as a raw material in the manufacture of 1,3-PDO.



Figure 2.1 Applications of 1,3-propanediol (Kaur et al., 2012).

Typically, 1,3-PDO was traditionally produced via chemical processes (Fig. 2.2). In the first reaction, acrolein is hydrated to 3-hydroxypropionaldehyde (3-HPA) under catalyzation solution using the acid catalyst. 3-HPA is then hydrogenated by the expensive catalyst (Fig. 2.2A) such as nickel and transformed to 1,3-PDO at high temperature up to 180°C (Arntz et al., 1994; Lawrence and Sullivan, 1972). For the second chemical synthesis (Fig. 2.2B), ethylene oxide, serves as the substrate, for converting to 1,3-PDO by the reactions of hydroformylation and hydrogenation, respectively (Brown et al., 2000). However, these pathways of 1,3-PDO synthesis are required high temperature, expensive catalyst, or high pressure during the hydroformylation and hydrogenation processes. Importantly, the toxic from both processes is necessary to be eliminated. Thus, they are not suitable and sustainable for an environment. However, the chemical process is decreasing when DuPont developed their recombinant microbial strain for producing 1,3-PDO and using corn sugar as a feedstock. This breakthrough discovery opened the green way of 1,3-PDO production. The production of PTT is then grown gradually and expanding the production base.



Figure 2.2 Chemical reactions of 1,3-propanediol synthesis: (A) Degussa method and

(B) Shell method (Saxena et al., 2009).

2.2 Polytrimethylene terephthalate

Polytrimethylene terephthalate or PTT is a novel interesting polyester. It is synthesized from polycondensation between terephthalic acid and 1,3-PDO (Fig. 2.3). After the discovery of how to synthesize PTT in 1941, this polymer is attracted by many scientists because of its potential properties. By considering the structure of this polymer, three methylene groups result in PTT possesses the zigzag structure, composing the outstanding properties that blend between nylon, PET, and PBT (Fig. 2.4). Therefore, PTT can substitute other polyesters to some extents. The properties of PTT are shown as the uniqueness such as good endurance, strength, high elastic recovery. However, at the beginning of producing PPT, it was not possible for commercialization because the cost of 1,3-PDO production was still high. Until Shell developed the chemical synthesis of 1,3-PDO for further producing PTT. PTT was thus first entered into the market by Shell under the trade name of CORTERRA. PTT is applied in many products. Either yarn, fiber, monofilament, or engineering thermoplastics. All PTT-transformed objects can be used as raw materials for other industries such as medical equipment, electronic parts, film, packaging, and glass, particularly, in the group of carpet, textile, and apparel.

With the special characteristics of the chemical structure of PTT, these make the clothes from PTT have high elasticity, stain resistance but still give the fresh color from intrinsic polymer, cleaning and drying faster, printable, no requirement of coating on polymer. This polymer is therefore quite suitable for producing sport clothes and can be replaced with nylon. Moreover, when 1,3-PDO can be produced based on bioprocess fermentation from the recombinant *E. coli*. This led both markets of 1,3-PDO and PTT to enlarge rapidly. To synthesize PTT, 1,3-PDO is condensed with terephthalic acid or dimethyl terephthalate, catalyzing to PTT (Fig. 2.3)(Houck et al., 2001; Liu et al., 2010).



Figure 2.3 Synthesis of polytrimethylene terephthalate from 1,3-propanediol and



Figure 2.4 The structure of different polyester ; PET, PTT and PBT (Liu et al., 2010).

2.3 Connection between glycerol and 1,3-propanediol

Petroleum is a predominant resource, behind energy and many products that have been used for a long time. Among them, Diesel, gasoline, jet fuel and natural gas are the energy derived from petroleum that employed in transportations. In the European region, around 47.5% of the oil was consumed in the road transportation section (Eurostat statistics Explained, 2020). Compared to Thailand, fuel was consumed in transportation by 39.2%. The remaining sectors were divided into the group of industry, household, commerce, and agriculture, accounting for 36.4, 13.0, 8.0, and 3.4 %, respectively (Department of Alternatives Energy Development and Efficiency, 2019).

The phenomena of climate change in recent years have a large impact on the earth. A long drought, flood, dust pollution, monsoon including pandemic or even moving the settlement of animals are more serious problems and intensify. To reduce fossil fuel, alternative energies are required. Among sustainable fuel, biodiesel is one of them and increasingly taken to substitute the fossil-derived diesel as shown in Fig 2.5. Several plant oils and animals fat or even cooking oil such as palm oil, soybeans, rapeseed, jatropha can be used as feedstock resource of biodiesel production via transesterification (da Silva et al., 2009; Ogunkunle and Ahmed, 2019). By using choices, the emission of greenhouse gas can be possibly reduced. However, in the process of transesterification reaction, a 10% surplus of glycerol is usually generated from the process. This leads the price of glycerol in the market to fall accordingly. Nevertheless, the cost of glycerol removal is also required. These result in the

uneconomical production and a risk to environmental contamination. To manage the waste to become zero and increase the driving force of the utilization of sustainable source concomitantly, glycerol should be changed to other products that have high value such as ethanol, succinic acid, and even 1,3-PDO (Dharmadi et al., 2006; Saxena et al., 2009).

Among the value-added chemicals, the production of 1,3-PDO seemed to be the simplest way derived from glycerol. By using the biological pathway, it composes of only two reaction steps for the transformation of glycerol to 1,3-PDO. Significantly, the attractive point is the rapidly growing market of PTT. With the properties of 1,3-PDO and being a monomer for producing PTT, 1,3-PDO should be tried to produce from glycerol.





Figure 2.5 Primary production of liquid biofuels, EU-28, 1990-2016 (Eurostat, 2018)

2.4 History of 1,3-propanediol

The journey of 1,3-PDO was back into around the late 18th century, when August Freund (1881) found 1,3-PDO from his spontaneous fermentation of soap lyes by *Clostridium pasterianum* (Przystałowska et al., 2015a). Even he tried to remove butyl-alcohol, but the new product still existed with different specific gravity to glycerol. 1,3-PDO in that time was reported in name of trimethylene glycol. Later in 1928, Braak further identified *Bacterium freundii* which is the microorganism that can produce 1,3-PDO. This led to works that isolated more microorganisms capable of producing 1,3-PDO from various animal manures. It found that those microorganisms group of *Citrobacter* sp. including *Citrobacter glycologenes*, *C. album*, *C. intermedium*, *C. decolorans*, *C. diversum*, *C. anindolicum* (Werkman and Gillen, 1932).

In the same period, microorganisms related to 1,3-PDO production were also studied indirectly from several reports and found in the spoiled wine by Pasteur (1866). Voisenet (1918) discovered the connection between bitterness character and acrolein from the presence of *Bacillus* amaracrylus in the wine spoilage. He indicated that the acrolein is generated from glycerol and always presents in the wine thus having a bitter character. In that wine, he also found the formation of 3-HPA and 1,3-PDO (Engels et al., 2016; Serjak et al., 1954; Vollenweider and Lacroix, 2004). This glycerol formation results from the attempt of yeast for preventing the osmotic stress, redox balancing, and maintaining phosphate level during the alcoholic fermentation (Scanes et al., 1998). This makes wine also have glycerol as a byproduct. Other contaminanting microbial strains besides S. cerevisiae can grow and use glycerol as substrate. Small amounts of other by-products were thus generated and found in the spoiled wine. Besides wine, acrolein was also found in cider, brandy, and pear juice by the contamination of *Clostridium perfringens* (Warcollier et al., 1932; 1934). The presence of acrolein in cider or beer also affected the character of alcoholic fermentation to be the lachrymatory or called Pepper. However, the bitterness flavor could be summarized that it was the result of the reaction between polyphenol compounds especially tannins and acrolein (Rentschler and Tanner, 1951). These discoveries led to other numerous investigations either the production pathway of 1,3-PDO and 3-hydroxypropionic acid (3-HP), the enzyme encoding intermediate chemicals or even microorganisms.

2.5 Native 1,3-propanediol producer

Besides microorganisms abovementioned, five main native producers that can produce 1,3-PDO from glycerol have also been usually reported including *Klebsiella* spp., Clostridia spp., Citrobacter spp., Bacillus spp., and Lactobacilli spp. (Ju et al., 2020, Forsberg et al., 1987, Sun et al., 2018). After 1,3-PDO-producing microorganisms have known in the field of alcoholic fermentation, this leads to the isolation and development of microbial strains for producing the essential monomer of PTT extensively by using native producers. For example, Vivek et al. (2016) selected Lactobacillus brevis N1E9.3.3 from the soil contaminated with crude glycerol. Concentration, productivity, and yield obtained from their strain were at 18.6 g/L, 0.78 g/L/h, and 0.89 g/g glycerol, respectively. Ju et al., (2020) used the isolated Lactobacillus reuteri CH3 for 1,3-PDO fermentation. Under the fed-batch fermentation and corn steep liquor as nitrogen source, 1,3-PDO can be derived at 68.32 g/L. Whereas, Maervoet et al. (2016) selected C. werkmanii and performed the additional deletion of *dhaD*, *ldhA*, *adhE* genes. This strain can produce 1,3-PDO with a maximum theoretical yield of 1 mol/mol glycerol in shake flask and 1, 3-PDO was achieved at the concentration of 422.01 mM (~32.11 g/L) by using co-substrate of glycerol and glucose K. oxytoca was also reported with the by-product gene deletions ($\Delta budA$,
$\Delta budB$, $\Delta adhE$, $\Delta ackA$, Δpta , $\Delta poxB$, $\Delta frdA$) in which 1,3-PDO was obtained at 76.2 g/L with the remained by-product of D-lactate at 111.9 g/L (Xin et al., 2017). With the deletion of *ldhA* in *K. oxytoca*, 1,3-PDO was improved up to 83.56 g/L accordingly, but 60.11 g/L 2,3-butanediol was also simultaneously produced (Yang et al., 2007). This indicates that even some by-product genes are removed, other by-products are still generated for balancing redox in the metabolic pathway.

As mentioned previously, K. pneumoniae and C. butyricum are mostly well known that are the major players in the field of 1,3-PDO production. Because they are tolerant to glycerol and 1,3-PDO concentration, and always selected as natural producers that are suitable for 1,3-PDO production in term of both productivity and concentration. Numerous studies reported 1,3-PDO production in high concentration from these microorganisms with and without molecular modifications. For example, the native K. pneumoniae (Mu et al., 2006) can produce 1,3-PDO to 51.3 g/L in fedbatch fermentation by using crude glycerol. Cheng et al. (2004) optimized the aeration level to observe the suitable conditions for producing 1,3-PDO by K. pneumoniae. 1,3-PDO was achieved at 70 g/L in the mix mode of anaerobic and microaerobic conditions. Besides the typical K pneumoniae, gene manipulations are also moderately applied in this microorganism to improve concentration, productivity, and yield. For example, gene deletions of poxB, pta, and ackA (Lin et al., 2016) in K. pneumoniae were performed. Under the fed-batch condition, 1,3-PDO was derived at 76.8 g/L with a yield of 0.66 mol/mol by the strain. This concentration was improved by 15% compared to the

parental strain ($\Delta ldhA$). In addition to the removal of by-product genes for eliminating the competitive routes in *K* pneumoniae, the expression of dhaL was also studied. This strain could produce 1,3-PDO of 20.59 g/L with a yield of 0.76 mol/mol in batch fermentation by using co-substrate of glycerol and mannitol (Lee et al., 2018). Recently, the new strain of K. pneumoniae J2B, was published with 33 gene modifications after developed this strain for several years. In this strain, *ccr*, *ptsG* and *ptsH* genes involving glucose transport system were removed. The diverse by-product genes such as *ldhA*, *adhE*, *frdA*, *pflB*, *mgsA*, *poxB*, *budA*, *budB*, and *budC*, were eliminated to direct carbon flow through 1,3-PDO as much as possible. The *nuoA* and *ndh* were also deleted to enhance reoxidation. Importantly, genes relevant to conversion of glucose to glycerol gpdl and gpp2 were expressed. By using this strain, K. pneumoniae J2B produced 1,3-PDO from glucose up to 62 g/L. This was the successful report of employing glucose as a substrate in K. pneumoniae (Lama et al., 2020).

For *Clostridium* spp., this microorganism relies on the strictly anaerobic conditions and hard to control the operation on the industrial scale, there were many reports about this microorganism. For example, *C. butyricum* AKR102a could produce 1,3-PDO from pure glycerol up to 93.7 g/L with a productivity of 3.3 g/L/h. When this strain was upscaled to 200 L, concentration and productivity were reduced to 61.5 g/L and 2.1 g/L/h, respectively. However, the concentration and productivity were still high and satisfactory (Wilkens et al., 2012). Furthermore, *C. butyricum* was also taken to fermentation in kind of immobilization. Productivity derived from this study was quite

high compared to the free cell (Dolejš et al., 2019). On the other hand, Wischral et al. (2016) overexpressed *gldA* and *dhaKLM* in *C. beijerinckii* DSM791 to increase the supply of NADH in the second reaction during 1,3-PDO production. The result confirmed that the overexpression of these genes supported the glycerol utilization, leading to a high concentration of 1,3-PDO (26.1 g/L) compared to the wild type strain (~20 g/L).

However, the constraint for *K. pneumoniae* is still characterized as an opportunistic pathogen that risks the industrial operator (Gonzalez-Pajuelo et al., 2006). Whereas *Clostridium* spp. is dependent on strictly anaerobic conditions that are difficult to control the parameter in the industry. This indicates that they seem not suitable for use as a host to produce 1,3-PDO in the industrial scale until some limitations are addressed.

2.6 Pathway

Among native producers, 1,3-PDO production pathway in *K. pneumoniae* was studied apparently. It is used as a model for 1,3-PDO production under anaerobic conditions. However, *K. pneumoniae* is a facultative anaerobe. Formation of 1,3-PDO under aerobic or microaerobic conditions can be possible (Liu et al., 2007). The characteristics of oxygen requirement for 1,3-PDO-producing microorganisms in the other genus such as *Citrobacter* sp. and *Lactobacilli* sp. were also reported (Maervoet

et al., 2016; Vivek et al., 2016) except *Clostridium* sp. that is strictly sensitive to oxygen (Raynaud et al., 2003).

For the well-known pathway in K. pneumoniae, glycerol is initially entered into the cell by a glycerol facilitator (GlpF) which is strongly over expressed under aerobic conditions (Huang et al., 2002). Glycerol is then separated into two coupled ways, oxidative and reductive pathways. On the oxidative route, glycerol is directed to the glycolytic pathway. After glycerol come into the cell, it is initially converted to dihydroxyacetone (DHA) by glycerol dehydrogenase (DhaA or DhaD), releasing reducing power NADH. Dihydroxyacetone kinase (DhaKLM) then catalyzes on DHA by taking phosphate from phosphoenolpyruvate or ATP to DHA and adding phosphate to DHA before transformed to dihydroxyacetone phosphate (DHAP). DHAP is further converted to glyceraldehyde-3 phosphate before directing to glycolysis pathway and reduced to several by-products that subsequently re-generated reducing power to the formation of 1,3-PDO in the reductive route (Lin et al., 2016) (Fig. 2.6). In a similar way, this route also presents in other microorganism groups, but byproduction formations are different distinctly, depending on microorganisms. For example, 2,3-butanediol is specially derived from K. pneumoniae and C. freudii, butanol is obtained from *Clostridium pasteurianum*, butyrate and acetate are mainly generated from C. acetobutyricum. Whereas, ethanol, lactate, acetate, are typically found in Klebsiella, Citrobactor freudii, Lactobacillus sp. and E. coli. (Homann et al., 1990; Luers et al., 1997; Magdouli et al., 2016; Sun et al., 2018; Vivek et al., 2016).

On the other hand, in the reductive branch, the entering glycerol is also changed to 3-HPA by coenzymeB12-dependent glycerol dehydratase (DhaB). 3-HPA is then reduced to 1,3-PDO by the NADH-dependent 1,3-PDO oxidoreductase (DhaT) or hypothetical oxidoreductase (HOR). In the second step of the reductive pathway, reducing power, especially NADH from the oxidative pathway is required to reduce 3-HPA to 1,3-PDO. As a result, glycerol metabolism usually couples with oxidative and reductive pathways. Furthermore, in the second reaction of a reductive pathway of *K*. *pneumoniae*, 3-HPA can be oxidized to 3-HP by aldehyde dehydrogenase in which 3-HP is one of another valuable building block biochemicals (Oh et al., 2018; Sun et al., 2018). It is also an interesting chemical due to the much more benefit of 3-HP with a broad range of applications.

In case of providing another source of carbon, such as glucose or xylose, glucose is directed to the oxidative route compared to use only glycerol. Besides, glucose is used as the source providing energy and intermediate in the metabolism and also serves as a source for generating ATP and reducing power (Sun et al., 2018). The glucose-glycerol co-utilization pathway presenting in *K. pneumoniae* is also similar in *L. brevis*. However, the operon encoding coenzyme B12-dependent glycerol dehydratase is the *pdu* operon instead of the *dhaB* operon for *K. pneumoniae* (Vivek et al., 2016).



Figure 2.6 Co-substrate metabolism pathway of glycerol and glucose in K.

pneumoniae (Sun et al., 2018)



Figure 2.7 Organization of genes involving glycerol fermentation derived K.

pneumoniae on plasmid pKM13 (Zheng et al., 2006).



Figure 2.8 Organization of the partial pdu operon of L. collinoides (Sauvageot et al.,

2002)

(A) Pathway of glycerol fermentation by enteric bacteria



Figure 2.9 Glycerol metabolism under anaerobic conditions (A) Typical glycerol pathway for enteric bacteria (B) regulon of glycerol utilization for *Citrobacter freundii* (C) operon of glycerol utilization for *Clostridium pasteurianum*; *dhaD*, glycerol dehydrogenase, *dhaK*, dihydroxyacetone kinase, *dhaB*, *dhaC*, and *dhaE* are the subunits of glycerol dehydratase, *dhaT*, 1,3-propanediol dehydrogenase, *dhaR*, regulatory protein, *orfW*, *orfX*, *orfY*, and *orfZ* are the open reading frames with unknown function. (Daniel et al., 1998).

2.7 The regulon encoding glycerol dehydratase in the different microorganisms

Although K. pneumoniae, K. oxytoca, Lactobacilli spp., and Citrobacter spp. have the same enzyme for catalyzing glycerol into 1,3-PDO which is coenzymeB12dependent glycerol dehydratase (Raynaud et al., 2003). Gene clusters involving regulation of glycerol catabolism in these microorganisms are aligned with three mainly different patterns. In K. pneumoniae and C. freundii, dha regulon is responsible for the same glycerol utilization. Considering the gene orientation as shown in Fig.2.7 and 2.9, genes involving glycerol fermentation are positioned in the opposite direction. Most genes of glycerol dehydratase which composed of *dhaBCDE* and orfZ (or dhaB123, gdrA) are oriented and grouped, whereas orfX (gdrB) encoding small reactivating factor of glycerol dehyatase is on another side and corresponds to the *dhaT* gene encoding of 1,3-PDO oxidoreductase. Compared to C. pasteurianum, genes related to the glycerol dehydratase and 1,3-PDO oxidoreductase are aligned in the same direction (Fig.2.9C). On the other hand, the pdu operon which is in turn responsible for 1,3-PDO pathway was found in the L collinoides, and these genes are transcribed in the one direction as same as the operon of *dhaB* in *C. butyricum* (Fig. 2.8 and 2.10). However, every operon or regulon as stated previously can convert glycerol to 1,3-PDO.



Figure 2.10 Overview of B12-independent glycerol dehydratase in *C. butyricum*. A. represents the arrangement of genes involving 1,3-production pathway. B represents the sequence in the promoter region. C. represents the result of northern blot analysis (Raynaud et al., 2003).

2.8 The modified microorganism for producing 1,3-propanediol

K. pneumoniae and *Clostidium* spp. are considered as the powerful player in the field of 1,3-PDO production even though the regard of the safety of authority is also important. Therefore, the use of microorganisms that is free of risk compared to those of *K. pneumoniae* as a host for 1,3-PDO production is a suitable way in the industry. *E. coli* is one of choice practically employed in the industry, especially DuPont. Several genes were modified in the metabolic pathway and can use glucose effectively instead of glycerol. This substrate takes a high advantage over glycerol in term of price and being a preferred carbon source for *E. coli*. Although, there were several genes were

modified in DuPont's strain to produce 1,3-PDO from glucose, only some relevant major genes were disclosed.

In DuPont's strain, essential genes derived from K pneumoniae, dhaB123 and gdrAB were introduced to the pathway in a non-native producer. Then other genes were modified to force the most of carbon from glucose direct to 1,3-PDO as much as possible as following Genes relevant to converting DHAP to glycerol, which composed of glycerol 3-phosphate phosphatase (GPP2) and glycerol 3-phosphate dehydrogenase (DAR1) derived from Saccharomyces cerevisiae were introduced in the pathway. By overexpression of these two genes, DHAP is converted to glycerol-3phosphate. Furthermore, to prevent the glycerol reversibly catalyzed to DHAP via glycerol-3-phosphate and DHA, respectively, the glpK and gldA were thus removed. The tpiA was eliminated to ensure the conversion of DHAP to glyceraldehyde 3phosphate (GAP). Besides, *tpiA* and *gapA* genes were also downregulated for alleviating the conversion GAP to other intermediates in the central glycolysis pathway. On the other hand, gene responsible for glucose transportation was replaced with galP and glk (Nakamura and Whited, 2003).

In other studies, especially in *E. coli*, most of the works put the effort into *E. coli* by overexpression of coenzyme B12-dependent glycerol dehydratase operon derived from *K. pneumoniae* or *Citrobacter freudii* on plasmid as shown in Table 2.1. However, Yun et al. (2021) recently used B12-independent glycerol dehydratase from *Clostridium butyricum*, which has the advantage in term of no requirement of

coenzyme B12 for catalyzation. This study indicates the one advanced step, due to the possible reduction of industrial cost. However, this study used the two strains of *E. coli* for expressing different enzymes. This is relatively complicated and may allow the instability of plasmids to happen on an industrial scale. The NADH+H⁺ or NADPH+H⁺ dependent 1,3-PDO oxidoreductase gene (*dhaT* and <u>yqhD</u>) that convert 3-HPA to 1,3-PDO was also combined in *E. coli*. No apparent difference of 1,3-PDO concentration between the utilization of *dhaT* and *yqhD* was observed (Table 2.1). Emptage et al. (2003) clearly showed that *yqhD* is more preferable than *dhaT* in *E. coli* to produce 1,3-PDO.

Numerous studies tried to create the recombinant *E. coli* like DuPont by expression of glycerol dehydratase, deletion of by-product, and overexpression of genes relevant to convert glucose to glycerol. For example, Lee et al., 2018 expressed the essential gene for producing 1,3-PDO and eliminated genes relevant to the possible competitive pathways such as *glpK*, *glpABC*, *glpD* for preventing glycerol conversion to DHAP and removing *pta-ackA genes* for reducing acetate, respectively. The disruption of *glPR* and *eda-edd* were also carried out to increase glycerol assimilation rate and protect the carbon flow to the unproductive pathway. However, the yield was still lower than the theoretical yield at 1.0 mol/mol while obtained only 0.47 mol/mol with a concentration around 30 mM (2.22 g/L) in the shake flask. Yang et al. (2018) also performed similarly but attempted to improve the 1,3-PDO concentration by overexpression of *gapN* (glyceraldehyde-3-phosphate dehydrogenase) to increase the

reducing power source. In this strain, 1,3-PDO was produced at 13.47 g/L with a yield of 0.64 mol/mol.

The problems for most of the works found are the shortage of reducing power and the deficiency of cell viability when deletes *ptsG* in *E. coli* (Lee et al., 2018; Yang et al., 2018). In some studies, they usually used only glycerol as a substrate and evaluated the efficiency of strain in only shaking flask scale. Glucose and glycerol were not provided appropriately or in the same ratio as Tong and Cameron (1992) suggested that the theoretical yield (1 mol/mol) can be possible when combining 2 mol glycerol and 1 mol glucose. Importantly, the conditions used to produce 1,3-PDO become limited in kind of micro or anaerobic conditions. These obstacles make us do not exactly determine the maximum possible concentration and yield in E. coli. For example, Yun et al., (2021) added 80 g glycerol with 10 g glucose to produce 1,3-PDO. The yield of 1,3-PDO was reached 0.67 mol/mol. Przystałowska et al. (2015b) also used their recombinant E. coli strain to produce 1,3-PDO with two-stage conditions. The first stage of conditions was set for producing the high cell density, following the typical mode of batch fermentation at 0.4 vvm and 60 rpm. These limitations should be addressed and optimized conditions to find the best condition for producing 1,3-PDO.

2.9 Glycerol dehydratase: mechanism, structure, and coenzyme

As we know that glycerol dehydratase is an important enzyme for the formation of 1,3-PDO and not exist naturally in *E. coli*. To design *E. coli* to produce 1,3-PDO, we

need to provide genes from native producers to catalyze glycerol to 3-HPA. In contrast to other enzymes, glycerol dehydratase requires the coenzyme B12 or adenosyl cobalamin (AdoCbl) as a cofactor. Moreover, there is no genes function to produce vitamin B12 like *K. pneumoniae* or *K. oxytoca* in *E. coli*. This point is accounting as a challenging task for a scientist when we select the non-native producer as a host because genes for synthesizing vitaminB12 are complex and maybe another step for developing strain in the future. CoenzymeB12 or vitaminB12 are thus supplemented to culture for making glycerol dehydratase function properly.

Glycerol dehydratase is an enzyme depending on coenzymeB12. It normally tends to function anaerobically for utilization of glycerol in the native producer. Among the coenymeB12-dependent enzyme, glycerol dehydratase was characterized in group II which functions to rearrangement of a substrate by elimination of hydroxyl group and involves in the reduction of coenzymeB12. In the presence of the glycerol and coenzymeB12, the mechanism of the chemical reaction is going to activate immediately. Glycerol dehydratase is thus catalyzed promptly but also inactivated rapidly. Because binding to substrate results in the conformational change of enzyme, triggering the homolytic cleavage on Co-C bond of coenzymeB12 and leading the covalent bond to break irreversibly. CooenzymeB12 is then becoming the dissociated form, resulting in unknown cobalamin (II) species (or slowly transformed to OH-Cbl) and adenosyl radical (Toraya and Mori, 1999). The adenosyl radical formed serves as an oxidizing agent and requires hydrogen from a substrate, forming 5-deoxyadenosine. A substrate is then rearranging into a product. The hydroxyl group from C-2 is transferred to C-1. Then, the product radical obtained by the transfer of hydrogen back from 5⁻.deoxyadenosine and the hydroxyl group is removed, becoming the aldehyde and H₂O (Daniel et al., 1998; Toraya, 2000).

After the aldehyde product is generated, the enzyme becomes substrate-free form, remaining only the biologically inactive coenzymeB12 or cobalamin species that interacted to enzyme tightly. Glycerol dehydratase is thus irreversibly inactive rapidly from homolytic cleavage of Co-C of coenzymeB12. Importantly, cobalamin species holding in glycerol dehydratase cannot function like at the beginning of a chemical reaction. The reaction is then stop and cannot be continued because of the suicide inactivation of glycerol dehydratase. However, the native producer always has reactivating factors to support glycerol dehydratase to reactivate and function again (Toraya et al., 2016). Based on the assumption of Shibata et al. (2005), the reactivating factors reactivate by the exchange of ATP and ADP. The ADP-containing reactivating factors initially binds to the inactivated glycerol dehydratase, releasing adenine residue and cobalamin (J). ADP on reactivating factors was then replaced with ATP, before finally becomes to ADP-binding reactivating factors to reactivate the inactivate glycerol dehydratase again.

On the other hand, after 5[·]deoxyadenosine residue was released from the coenzyme B12 during catalysis, Cob(II)alamin or coenzyme B12 has become an inactive form. To regenerate the active form of coenzyme B12, adenosyl transferase is catalyzed

on the inactive coenzyme B12 by bringing the adenosyl residue from ATP to add on an inactive form of coenzyme B12.



Figure 2.11 Diol dehydratase-reactivating factor (Shibata et al., 2005). (A)
Reactivation of inactivated holoDD by exchange of a damaged cofactor for intact coenzyme. AdoH, 5[.]-deoxyadenosine, Cbl^{II}, cob(II),cob(II)alamin; Cbl^I, cob(Jalamin; X-Cbl, a damaged cofactor, Dotted lines indicated the reactions outside of DD. (B) Organization of the genes for DD and DDR in the *pdu* operon (C). Postulated mechanism of reactivation of inactivated holoDD by DDR. E, apoDD; RF, DDR; AdoH, 5[.]deoxyadenosine; Cbl, cobalamin.

2.10 Process optimization of 1,3-propanediol production

2.10.1 Mode of operations: anaerobic, microaerobic, and aerobic conditions

Although glycerol dehydratase was believed to function effectively under anaerobic conditions, it is not always true. Liu et al. (2007) performed fed-batch fermentation by using K pneumoniae to ferment under anaerobic and microaerobic conditions. They found that the result of 1,3-PDO concentrations was not distinctly different under both conditions either concentration (75 g/L and 72 g/L), productivity (2.2 and 2.1 g/L/h), or even yield (0.61 and 0.57 mol/mol). However, this study did not indicate the detail of conditions clearly about how to set anaerobic and microaerobic conditions. Oh et al. (2012) also determined the mode of operations by setting the different aeration level against 1,3-PDO production in *K* pneumoniae deficient *ldhA* in the range of 0.5-3.5 vvm. They found the aeration of 2.0 vvm, 1,3-PDO can be achieved satisfactorily at 94.3 g/L, that was higher than those of the low aeration at 0.5 vvm (80.8 g/L). Whereas Ji et al. (2009) employed microaerobic conditions in the first five hours by a combination of between 0.1 vvm and 200 rpm, 1,3-PDO can be derived at 70 g/L. Noticeably that not only K. pneumoniae is a facultative anaerobe, but it is also a natural producer for 1,3-PDO. Hence it tends to efficiently generate 1,3-PDO in both conditions.

Compared to *E. coli*, conditions that applied for the cultivation of microorganism under microaerobic conditions or aerobic conditions were not applied clearly and not fully disclosed. Normally, several studies on *E. coli* used mild conditions. For example, Przystalowska et al. (2015) employed low speed at 60 rpm for generating 1,3-PDO at 10.6 g/L. Whereas, Zhang et al., (2006) varied the concentrations of components in 1,3-PDO fermentation with fixed parameters at 180 rpm and 0.8 vvm. The 1,3-PDO concentration was at 43.1 g/L from 61.8 g/L of glycerol, 6.2 g/L of yeast

extract, and 49 mg/L of vitamin B12. On the other hand, for a native producer like L. reuteri CH53, the generation of 1,3-PDO was achieved efficiently under mild anaerobic conditions (no-aerated conditions) (Ju et al., 2020). 1,3-PDO from this strain under aerobic conditions was obviously such a low level at 1.51 g/L. Compared to anaerobic or unaerated conditions, 1,3-PDO were presented at a higher level of 11.88 or 11.57 g/L, respectively. In contrast these aforementioned to microorganisms, *Clostridium* spp. is different from other 1,3-PDO-producing bacteria because it requires anaerobic conditions for microbial growth (Szymanowska-Powałowska and Białas, 2014). As a result, aeration or speed is one of the most important parameters to optimize 1,3-PDO production. 1,3-PDO is thus varied considerably according to the different microorganisms and conditions.

2.10.2 pH

As well as other conditions most of the works focused only *K*. *pneumoniae*, by varying the pH in the short range. For example, Oh et al. (2018) determined the effect of pH in the range of 6.7 (6, 6.5, and 7) upon 1,3-PDO production in *K. pneumoniae*. The result showed that a pH of 6.5 is suitable for glycerol utilization. The 1,3-PDO concentration was around 53.5-57.7 g/L that was higher than those of pH 6 and 7 (47.4-49.9 and 44.6-49.8 g/L, respectively). On the other hand, Sattayasamitsathit et al. (2011) also studied the effect of pH at 6.5, 7.0, and 7.5, respectively. The lowest pH at 6.5 provided the highest 1,3-PDO concentration at 7.8 g/L compared to the pH at 7 and 7.5 (18.51 and 16.18 g/L). In contrast, the cell growth was high at pH 7.5 with a

specific growth rate of 0.31 h^{-1} . As a result, they combined two pHs in their fermentation by building biomass in the first 6 h at pH 7.5 and following with pH 6.5 until the end of fed-batch fermentation.

2.10.3 Co-substrate

Glycerol is the true substrate for 1,3-PDO production. To produce 1,3-PDO, the theoretical yield from glycerol is only 0.72 mole/mole. However, most of the research cannot make 1,3-PDO reaching that theoretical yield either natural or non-natural producers, this is due to naturally metabolic limitation. To make the redox balance, some by-products are thus generated, leading to reduce the maximum theoretical yield. However, if the system is supported by the source of reducing equivalent, the practical yield can be achieved close to theoretical yield at 1 mol/mol. By using co-substrate and several types of 1,3-PDO producing microorganisms, 1,3-PDO concentration can be improved evidently and always close to 1 mol/mol. Ju et al. (2020) used *L reuteri* CH53 for producing 1,3-PDO by using glycerol and glucose, yield of 1,3-PDO can be obtained at 0.80 g 1,3-PDO g glycerol (0.96 mol/mol). Likewise, Xin et al (2016) employed *C diolis* for producing 1,3-PDO from a combination of glucose and glycerol. They found that yield was improved by 22%, accounting for 0.86 mol/mol.

2.10.4 Biomass and waste

The resource of glycerol does not come from only the industry of purified glycerol. The utilization of crude glycerol derived from biodiesel is one of the interesting choices for adding the value of waste. The powerful final target of applying several native producers is always a demonstration of crude glycerol consumption. Furthermore, agricultural residue like corn steep liquor sometimes is combined with crude glycerol, serving as nitrogen source instead of yeast extract or beef extract in 1,3-PDO production (Oh et al., 2018). For example, Dolejĵš et al. (2019) determined 1,3-PDO production from crude glycerol using C. butyricum. 1,3-PDO was generated at 12.6 g/L from 40.2 g/L of crude glycerol, accounting for the yield of 0.35 g/g. Whereas, Laura et al., (2020) also selected *K. pneumoniae* as a host for producing 1,3-PDO from crude glycerol. These studies indicate that 1,3-PDO-producing microorganisms are powerful and possible for developing on an industrial scale. Importantly, by using crude glycerol, the purification process is not required. However, caution for using crude glycerol should take considered. The certain microorganism may not tolerate methanol or ash which is a residual component from the biodiesel process. For example, Laura et al., (2020) used crude glycerol as a substrate, 1,3-PDO was reduced ้⁷วักยาลัยเทคโนโลยีสุร^{นโ} to 51% compared to pure glycerol.

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Microorganism	Aeration	Main Carbon Source	Cultivation Condition	Concentration (g/L) Yield [mol/mol]	Productivity (g/L/h)	Sources
<i>E. coli</i> AG1 overexpressing <i>dha</i> operon (<i>K</i> .	Anaerobic (ND)	Gly	Modified ST medium + YE	0.7 [0.46 mol/mol glycerol]	0.02	Tong et al. (1991)
pneumoniae) on pTC1	Anaerobic (ND)	Gly/Glu	Modified ST medium + YE	1.2 [0.63 mol/ mol glycerol]	0.04	Tong and Cameron. (1992)
	Anaerobic (ND)	Gly/Glu	Minimal medium + YE	6.3 [0.82 mol/mol glycerol]	0.20	Skraly et al. (1998)
<i>E. coli</i> BL21 overexpressing <i>dha</i> operon (<i>K.</i> <i>pneumoniae</i>) on pACYCDuet- <i>dhaBT</i>	Aerobic conditions, Shake flask, 150 rpm	Gly	Minimal medium + YE	11.3 [0.34 mol/mol glycerol]	0.38	Ma et al. (2009)
<i>E. coli</i> K-12 overexpressing <i>dhaB12</i> derived from <i>C. butyricum</i> and <i>yqhD</i> derived from <i>E.</i> <i>coli</i> on pDY220	2 step fermentation. First: aerobic conditions at 2 vvm, 400 rpm	Gly/Glu	Minimal medium + YE	104.4 [1.09 mol/mol glycerol]	2.61	Tang et al. (2009)

 Table 2.1 Comparison of 1,3-propanediol fermentations by using the developed recombinant E. coli strains.

ND = Not determined, YE = Yeast extract, Gly = Glycerol, Glu = Glucose

Microorganism	Aeration	Main Carbon	Cultivation Condition	Concentration	Productivity	Sources
		Source	Condition	[mol/mol]	(g/L/II)	
	Second: anaerobic conditions		A			
Co-culture of <i>E. coli</i>	Anaerobic	Gly/Glu	Minimal	41.6	0.69	Yun et al. (2021)
Rosetta-dhaB12	conditions,		medium + YE	[0.67 mol/mol		
derived from C.	Shake			glycerol		
<i>butyricum</i> on pANY with <i>E. coli</i> BL21	(200 rpm)					
overexpressing <i>dhaT</i> from <i>C. butyricum</i> on						
pET-30a(+)						
E. coli DH5a	Micro-	Glu	Minimal	12.1	0.2	Liang et al.
overexpressing <i>dha</i> operon (<i>K. pneumoniae</i>)	aerobic conditions,		medium (M9) +	[0.47 mol/mol glucose]		(2011)
and <i>gpd1-gpp 2 (S</i> . <i>cerevisiae</i>) on pQKG	Shake flask (200 rpm)					
and the stress-induced	(200 1pm)					
region (SIR) on pSCD						

Table 2.1 Comparison of 1,3-propanediol fermentations by using the developed recombinant *E. coli* strains (Continued).

ND = Not determined, YE = Yeast extract, Gly = Glycerol, Glu = Glucose

Microorganism	Aeration	Main Carbon Source	Cultivation Condition	Concentration (g/L) Yield	Productivity (g/L/h)	Sources
<i>E. coli</i> BL21 overexpressing <i>dha</i> operon from <i>C. freundii</i>	Micro-aerobic conditions, Fermenter (0.4 vvm, 60	Gly	Minimal medium + YE	[mol/mol] 10.6 [0.4 mol/mol glycerol]	0.06	Przystalowska et al. (2015)
and <i>anal from</i> <i>K.pneumoniae</i> on pCF1 and pKP2	(pm)		A	A		
<i>E. coli</i> JM- 30BY15AB overexpressing <i>dha</i> operon (<i>K. pneumoniae</i>) and <i>yqhD</i> (<i>E.</i> <i>coli</i>) on dual vector pQE30 and pQE15A	Aerobic conditions, Fermenter (- vvm, 400 rpm)	Gly		13.1 [0.59 mol/mol glycerol]	ND	Hong et al. (2015)

 Table 2.1
 Comparison of 1,3-propanediol fermentations by using the developed recombinant *E. coli* strains (Continued).

ND = Not determined, YE = Yeast extract, Gly = Glycerol, Glu = Glucose

<i>E. coli</i> EB13-1	Micro-aerobic	Glu	Minimal	ND	ND	Lee et al. (2018)
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Microorganism	Aeration	Main Carbon Source	Cultivation Condition	Concentration (g/L) Yield [mol/mol]	Productivity (g/L/h)	Sources
($\Delta glpK$	conditions,		medium (<mark>M9</mark>) +	[0.47 mol/mol		
$\Delta glpABC$	Shake flask		YE	glucose]		
$\Delta glpD, \Delta glpR$	50 mL/100 rpm)					
∆edd-eda ∆pta-						
ackA)						
overexpressing						
dhaB, gdrAB,			A			
and <i>dhaT</i> (K.						
pneumoniae)						
and gpd1-gpp2						
(S. cerevisiae) on						
pABG and pDPT		Ch		10		
ND = Not de	termined. YE = Yeast ex	tract. Glv = Glvcer	ol. Glu = Glucose	20123		

 Table 2.1
 Comparison of 1,3-propanediol fermentations by using the developed recombinant *E. coli* strains (Continued).

ed, YE = Yeast extract, Gly = Glycerol, Glu = Glucose

Microorganism	Aeration	Main Carbon Source	Cultivation Condition	Concentration (g/L) Yield [mol/mol]	Productivity (g/L/h)	Sources
<i>E. coli</i> JA11 overexpressing <i>gdrAB-dhaB123-</i> <i>yqhD</i> on pTrc02 and employing optimizing UTR combined with overexpressing <i>gapN</i> and <i>galP-glK</i> on pACYC09	Micro- aerobic conditions, Shake flask 80 mL/220 rpm)	Gly/Glu	Minimal medium + YE	13.5 [0.64 mol/mol glycerol]	0.22	Yang et al. (2018)
E. coli NSK015 (ΔldhA::gdrAB- dhaB123 ΔackA::FRT ΔpflB::yqhD Δfrd::cat-sacB)	Aerobic conditions (1 vvm, 400 rpm)	Gly/Glu	Minimal medium (AM1) วิวักยาลัยเทคโนโ	36.8 [0.99 mol/mol glycerol]	0.34	This study
v ´	(1 vvm, 400 rpm)	Gly/Cassava starch	Minimal medium (AM1)	31.9 [0.84 mol/mol glycerol]	0.29	This study

 Table 2.1
 Comparison of 1,3-propanediol fermentations by using the developed recombinant *E. coli* strains (Continued).

CHAPTER III

FUNCTIONAL EXPRESSION OF ARTIFICIALLY REARRANGED GLYCEROL DEHYDRATASE OPERON AND 1,3-PROPANEDIOL OXIDOREDUCTASE GENE DEVOID OF PLASMID EXPRESSION SYSTEMS IN *Escherichia coli* FOR THE HIGH YIELD OF 1,3-PROPANEDIOL

3.1 Introduction

As stated in previous chapter, 1,3-Propanediol (1,3-PDO) has been known for a long time to have the diverse benefits and can be applied in different products such as cosmetics, laminates, coating, lubricant, antifreeze, etc. Importantly, 1,3-PDO is also essential component to produce polytrimethylene terephthalate (PTT) which further transformed to use in the fiber and apparel industries. 1,3-PDO is thus still attracted from several scientists to produce via biological process for replacing the conventionally chemical method, although DuPont is a forefront company to produce the bio-based 1,3-PDO with the breakthrough methods. Numerous studies are still focus to observe the alternative way. To present the new aspect of 1,3-PDO production via biological process in this study, our metabolically engineered *E. coli* NSK015 for producing 1,3-PDO was developed and disclosed in this chapter by introducing the *gdrAB-dhaB123* derived from *Klebsiella pnumoniae* and overexpressing *yqhD* under the control of *E. coli ldhA* and *pflB* promoters, respectively. The positive clones were then verified the expression of heterologous genes on the modified aldehyde indicator plates before further eliminating the major by-products genes. The different developed 1,3-PDO-producing *E. coli* was subsequently determined the efficiency of 1,3-PDO in the shake flask. The details can be found in this chapter.



2Glycerol + Glucose

2(1,3-Propanediol)+ 2Acetate + 2Formate

Figure. 3.1 Proposed metabolism for 1,3-propanediol production in a metabolically engineered *E. coli* NSK015. Genes involving in the pathway are summarized in the box. Red cross signs represent the gene deletion. Genes *gdrAB-dhaB123* and *yqhD* were expressed under the control of native *ldhA* and *pflB* promoters.

3.2 Materials and methods

3.2.1 Microbial strains, plasmid, media, primers, and growth conditions

Table 3.1 shows plasmids, primers and microbial strains used in this study. *E. coli* C was used as a parental strain for engineering its metabolic pathway. *E. coli* TOP 10 was used to maintain constructed plasmids. Microbial strains were cultivated and maintained in LB broth or agar, at 37° C Apramycin (Apra), Chloramphenicol (Cm), Ampicillin (Amp), and Kanamycin (Km) antibiotics were supplemented into media as appropriated. An LB aldehyde indicator plate containing 3% (w/v) glycerol, 0.5% (w/v) sodium bisulfite and 0.05% (w/v) pararosaniline in ethanol was modified from the report of Conway et al. (1987) and used to identify the expression of glycerol dehydratase operon under the native *ldhA* promoter. Coenzyme-B12 was also provided at the concentration of 15 μ M during 1,3-PDO production. The low salt AM1 production medium (Martinez et al., 2007) was used throughout this study. Yeast extract (5 g/L) was also provided in AM1 as appropriate to determine its effect over fermentation.

3.2.2 Construction of an artificially expressing operon of glycerol dehydratase with its reactivating factors

Genes encoding glycerol dehydratase operon (*dhaB1*, *dhaB2*, and *dhaB3*) and its re-activating proteins (*gdrA* and *gdrB*) were derived from *K. pneumoniae* subsp. *pneumoniae* TISTR1867. Since *gdrB* gene is naturally oriented in the opposite

direction to other genes in *dhaB* operon and *gdrA*, the expressing cassette for glycerol dehydratase operon and its reactivating factors was then artificially constructed by the re-arrangement of the gdrB orientation to be identical to other genes in the operon. This artificial cassette was then expressed under the control of an intact E. coli ldhA promoter. The detail for construction of the artificial cassette is shown in Fig. 3.2. In step 1, gdrB gene and its surrounding sequences were amplified using CompgdrA(down)-gdrB-R (forward) and DH-ldhAp-gdrAB-dhaB123-F (reverse) primers. The forward primer contained sequences corresponding to the downstream region of gdrA (dash underlined letters) with space sequences between gdrA and gdrB genes (bold letters), and the upstream region of gdrB gene (underlined letters) at 5[,] and 3[,] ends respectively. The reverse primer contained sequences corresponding to the downstream region of the *ldhA* gene of *E* coli host cell (bold) and to the termini of *gdrB* gene (underlined letters) at 5[,] and 3[,] ends respectively. The amplified PCR fragment (gdrB fragment) was cloned into pCR2.1-TOPO (Step 2), and the plasmid containing gdrB sequences lying in the same orientation of dhaB operon was selected and designated to be pKJ1010 (Step 3). On the other hand, dhaB123 and gdrA genes and their surrounding sequences were also amplified using Comp-gdrB(up)-gdrA-F (forward) and DH-ldhAp-gdrAB-dhaB123-R (reverse) primers. The forward primer contained sequences corresponding to the upstream region of gdrB (dash underlined letters) with space sequences between gdrAand *gdrB* genes (bold letters), and the downstream region of *gdrA* gene (underlined letters) at 5[,] and 3[,] ends respectively. The reverse primer contained sequences corresponding to the upstream region of the *ldhA* promoter of *E. coli* host cell (bold letters) and to the start codon of *dhaB1* gene (underlined letters) at 5[,] and 3[,] ends respectively. The resulting PCR fragment contained 5[,]-gdrA-dhaB123-3[,] sequences lying in the same orientation of gdrB gene on the plasmid pKJ1010 (Step 4).

The plasmid pKJ1010 was further used as the template for PCR amplification in an inside-out direction using dhaB1-M13-R/F (forward) and CompgdrA(down)-gdrB-R (reverse) primers. The forward primer contained sequences complementary to the upstream region of dhaB1 gene (bold letters) and corresponding to M13 region on the plasmid (dash underlined letters) at 5, and 3, ends, respectively. This resulting PCR fragment was then treated with DpnI restriction enzyme and purified. The PCR fragment was a linearized pKJ1010 plasmid missing surrounding sequences of gdrB gene (Step 5). To facilitate the ligation of the linearized plasmid with an insert cassette (gdrA-dhaB123) in the next step, the linearized DpnI-treated pKJ1010 plasmid was used as the template to generate sticky ends at both sides using either dhaB1-M13-R/F or Comp-gdrA(down)-gdrB-R primers in separated PCR reactions (Step 6). Both amplified single-stranded PCR fragments were combined. The re-combined double-stranded PCR product was therefore treated with calf intestinal phosphatase (CIP) and purified. The resulting double-stranded CIP-treated PCR product contained sequences derived from the linearized pKJ1010 plasmid with sticky ends at both sides (Step 7). Consequently, the amplified PCR product in Step 4 was also used as a template in separate PCR reactions using either Comp-gdrB(up)-gdrA-F and M13-dhaB1-R

primers (Step 8). The M13-dhaB1-R primer contained sequences corresponding to M13 region on the plasmid and the upstream neighbouring region of *gdrA* (underlined letters). Both amplified single-stranded PCR fragments were combined, followed by the annealing reaction between two single-stranded PCR fragments performed in the same manner as previously mentioned. The resulting double-stranded PCR fragment contained sequences derived from the *gdrA-dhaB123* with sticky ends at both sides. The double-stranded PCR fragment was further treated with T4 polynucleotide kinase (T4-PNK) in Step 9. Finally, the CIP-treated linearized pKJ1010 plasmid was ligated with the T4-PNK treated *gdrA-dhaB123* insert cassette facilitated by the complementary sequences at overhangs on their both ends. The ligated plasmid contained the artificially expressing cassette *gdrAB-dhaB123* and was designated to pKJ1011 (Step 10).

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Figure 3.2 Construction of artificially operon of *gdrAB-dhaB123* on pKJ1011.

3.2.3 Construction of an expressing cassette encoding the native *E. coli yqhD*

E. coli yqhD gene was also cloned, integrated, and overexpressed under the control of the *E. coli pflB* promoter. This strategy was performed by the replacement of the *pflB* gene with the expressing cassette, thus simultaneously preventing the accumulation of acetyl-CoA, a precursor for acetate, but facilitating a higher re-oxidation rate of NADPH (Fig. 3.1). The detail for construction of the expressing cassette is shown in Fig. 3.3. The upstream of *pflB* was first amplified by yqhD-pflB-F (forward) and upfocA-R (reverse) primers (Step 1). The forward primer contained sequences corresponding to the upstream regions of yqhD gene (bold letters) and *pflB* gene (underlined letters) at 5[,] and 3[,] ends, respectively. The resulting PCR fragment contained $yqhD^{,...}pflB-focA^{,...}$ sequences (Table 3.1).

The *yqhD* gene including its own promoter was further amplified by pflB-yqhD-F and pflB-yqhD-R (Step 2) while the downstream of pflB gene was amplified by pflA-F and yqhD-pflB-R primers (Step 3). The yqhD-pflB-R primer contained sequences corresponding to the downstream regions of yqhD gene (bold letters) and *pflB* gene (underlined letters) at 5[,] and 3[,] ends respectively (Table 3.1). After individual fragments were generated, the downstream of the *pflB* fragment was initially joined to yqhD fragment. Briefly, both PCR fragments were combined as a molar ratio of 1.1 and added in the PCR reaction containing Phusion polymerase (Invitrogen, USA) without primers using a slow annealing procedure at a decreasing rate of 0.1 °C/min for 15 cycles (Step 4). The gradual annealing step was facilitated by the overlap regions derived from yqhD-pflB-R and pflB-yqhD-F primers. The resulting fragment contained yqhD'-yqhD-pflB''-pflA'' sequences. The fragment was further amplified using pflA-F and pflB-yqhD-R primers (Step 5). The amplified fragment was then joined to the upstream fragment of *pflB* as aforementioned (*yqhD-pflB-focA*) from Step 1 in PCR reaction containing Phusion polymerase without primers using a slow annealing procedure at a decreasing rate of 0.1 °C/min for 15 cycles (Step 6). The annealed fragment was used as a template for the amplification of the shorter fragment containing *focA*-*yqhD*-*pflB*["] sequences using pflA-F and focA-R primers (Step 7). The *focA*-*yqhD*-*pflB*["] is thus cloned into pCR2.1-TOPO (Step 8), generating the plasmid pKJ1014.



Figure 3.3 Construction of pKJ1014 for the replacement of *pflB* with *yqhD*.

3.2.4 Deletions of *ldhA*, *ackA* and *frdABCD* genes, and chromosomal integrations of the artificial *gdrAB-dhaB123* operon and *yqhD* gene in *E. coli* C

Gene encoding lactate dehydrogenase (*ldhA*) was initially deleted in *E. coli* C wild type. Briefly, the *cat-sacB* (chloramphenicol acetyltransferase-levan sucrase) cassette was amplified by PCR using the genomic DNA of *K. oxytoca* KMS005 (Jantama et al., 2015) as a template. The forward primer (DH-ldhAp-cat-sacB-F) contained hybrid sequences homologous to the downstream regions of *E. coli ldhA* gene and the *cat-sacB* cassette (bold letters) at 5[,] and 3[,] ends respectively, while the reverse primer (DH-ldhAp-cat-sacB-R) contained hybrid sequences homologous to upstream regions of *E. coli ldhA* gene and the *cat-sacB* cassette (bold letters) at 5[,] and 3[,] ends respectively, while the reverse primer (DH-ldhAp-cat-sacB-R) contained hybrid sequences homologous to upstream regions of *E. coli ldhA* gene and the *cat-sacB* cassette (bold letters) at 5[,] and 3[,] ends respectively. The amplified PCR fragment (*ldhA cat-sacB-ldhA*^{,*}) was then transformed into *E. coli* C carrying the plasmid pLOI3420. The homologous recombination of the fragment on the chromosomal *ldhA* region of *E. coli* C wildtype was confirmed using the colony PCR technique. The verified clone was renamed as *E. coli* NSK001 (*AldhA:cat-sacB*).

Gene encoding acetate kinase (*ackA*) was also interrupted using techniques previously developed by Datsenko and Wanner (2000). Briefly, the PCR fragments containing kanamycin resistant gene flanked by parts of upstream and downstream sequences encoding *ackA* genes, were amplified by PCR reaction using the pKD4 plasmid as a template. The forward (ackA-pKD4-F) and reverse (ackA-pKD4-R) primers contained hybrid sequences homologous to upstream and downstream regions of *E. coli ackA* gene at both 5[,] ends, respectively, followed by upstream and downstream regions of aminoglycoside phosphotransferase (*km*) gene on pKD4 plasmid at both 3[,] ends (bold letters) respectively. The resulting PCR fragments (*ackA*, FRT-*km*-FRT-*ackA*,) were transformed into *E. coli* NSK001 carrying the plasmid pLOI3420. The integration of the fragment on the chromosomal *ackA* region of *E. coli* NSK001 was also confirmed using the colony PCR technique using ackA-F and JMackA-R primers. The verified clone was renamed as *E. coli* NSK002 ($\Delta ldhA$::*cat-sacB* $\Delta ackA$::FRT-*km*-FRT).

E. coli NSK002 strain further replaced the *cat.sacB* cassette on its chromosomal DNA by the artificially expressing cassette *gdrAB-dhaB123*. Briefly, *E. coli* NSK002 strain carrying pLOI3420 was transformed with the fragment of *gdrABdhaB123* cassette derived from pKJ1011 by electroporation. A specific integration of the artificially expressing *gdrAB-dhaB123* cassette at *E. coli ldhA* region occurred via a double homologous re-combination event facilitated by homologous sequences between upstream and downstream regions on chromosomal DNA and flanked regions on the cassette. The *gdrAB-dhaB123* cassette was designed to integrate in frame with the start codon at the *ldhA* promoter. The cassette therefore was expressed under the control of the host *ldhA* promoter. Colonies that exhibited loss of apramycin, and chloramphenicol resistances and viability on sucrose medium subsequently verified the integration of the cassette at the *ldhA* promoter by the PCR technique. The confirmed clone was designated *E. coli* NSK003 (*\LdhA::gdrAB-dhaB123\DackA::*FRT*-km*-FRT). To
remove the *km* gene, a plasmid pFT-A expressing a recombinase was transformed into *E. coli* NSK003. *E. coli* NSK003 strain carrying pFT-A was further cultured in LB medium containing chlortetracycline (20 μ g/mL) at 30°C for 8 h to allow self-recombining between FRT sites facilitated by the recombinase. Then single colonies were obtained on LB agar on incubating plates at 45°C for 24 h. Clones that did not confer antibiotic resistances were picked and confirmed for the loss of antibiotic resistant genes by the colony PCR technique using ackA-F and JMackA-R primers. The verified clone was renamed as *E. coli* NSK012 ($\Delta ldhA$::gdrAB-dhaB123 Δ ackA::FRT).

Gene encoding pyruvate formate lyase (*pflB*) was further deleted in *E.coli* NSK012 following the same strategy for deletion of *ldhA* gene as previously mentioned. The forward (WMpflBA-F) and reverse (WMpflBA-R) primers contained hybrid sequences homologous to downstream and upstream regions of *pflB* gene at both 5[,] ends (underlined) respectively, followed by downstream and upstream regions of *cat-sacB* cassette on genomic DNA of *K. oxytoca* KMS005 at both 3[,] ends (bold letters) respectively. The amplified PCR fragment (*pflB*[,]*cat-sacB-pflB*[,]) was then transformed into *E. coli* NSK012 carrying the plasmid pLOI3420. The integration of the *pflB*[,]*cat-sacB-pflB*[,] fragment on the chromosomal *pflB* region of *E. coli* NSK012 was further confirmed using the colony PCR technique. The verified clone was re-named as *E. coli* NSK013 ($\Delta ldhA$::*gdrAB-dhaB123* $\Delta ackA$::FRT $\Delta pflB$::*cat-sacB*). *E. coli* NSK013 strain carrying pLOI3420 was further transformed with the fragment of *focA*[,]*-yqhD*-*pflB*[,] derived from pKJ1014. Colonies were screened on replica plates including LB agar

supplemented with sucrose, apramycin, and chloramphenicol. Colonies that exhibited loss of apramycin, and chloramphenicol resistances and viability on sucrose medium subsequently verified the integration of the cassette at the *pflB* promoter by PCR technique. The confirmed clone was designated *E. coli* NSK014 ($\Delta ldhAB$::*gdrABdhaB123* $\Delta ackA$::FRT $\Delta pflB$::*yqhD*).

Gene encoding fumarate reductase (*frdABCD*) was further deleted in *E*. coli NSK014 following the same strategy of inside-out technique (Jantama et al., 2008), the *frdABCD* fragments were amplified by frdABCD-F and frdABCD-R primers and cloned into pCR2.1-TOPO. The cloned plasmid was used as the template in the inside-out PCR reaction by IOfrd-F and IOfrd-R primers. The inside-out resulting PCR product contained sequences derived from pCR2.1-TOPO with the flanking regions of *frdABCD* gene thus missing its central sequences. The *cat-sacB* cassette was then amplified from *K. oxytoca* KMS005 by catfrd-F and catfrd-R primers using the PCR reaction and ligated into the inside-out resulting plasmid, generating pKJ1012. The amplified PCR fragment (frd'-cat-sacB-frd") derived from pKJ1012 was then transformed into E. coli NSK014 carrying the plasmid pLOI3420. The integration of the frd'-cat-sacB-frd" fragment on the chromosomal frd region of E. coli NSK014 was also confirmed using the colony PCR technique. The verified clone was re-named *E. coli* NSK015 ($\Delta ldhA$::gdrAB-dhaB123 Δ ackA::FRT $\Delta pflB$::yqhD $\Delta frdABCD$::cat-sacB).



Figure 3.4 An overview of molecular engineering for constructing recombinant *E. coli* NSK strains to produce 1,3-propanediol. (A) Genome replacement between the artificial operon of *gdrAB-dhaB123* and *ldhA* genes by double homologous recombination in *E. coli* C (B) Schematic model of the integrated artificial operon of *gdrAB-dhaB123* under the control of *E. coli ldhA* promoter. (C) Sequential development of *E. coli* NSK strains for 1,3-PDO production.

3.2.5 Enzyme assay

Cells at the late-log phase cultures were harvested by centrifugation of 4000 g at 4°C for 10 min. The pellets were washed using 0.05 M PBS buffer (pH 8.0) twice then re-suspended with a final volume of 2 mL. The cell was further disrupted by an ultrasonic disintegrator at 40 kHz with alternate pulses of sonication for 10 min. The cell debris in the crude lysate were separated. The supernatant was used to determine the total protein concentration by the Bradford's method (Bradford, 1976) and its enzymatic activities.

To determine the activity of glycerol dehydratase, the method of Knietsch et al. (2003) was applied. The apparent molar extinction coefficient derived from 3-hydroxypropionaldehyde(3-HPA) used for the calculation of the specific enzyme activity of glycerol dehydratase was 5.23×10^3 M⁻¹cm⁻¹. The specific glycerol dehydratase activity was reported as amounts of aldehydes formed per mg total protein. To determine the 1,3-PDO oxidoreductase activity, the enzymatic compositions in the reaction were the same as in those of the glycerol dehydratase activity according to the method of Knietsch et al. (2003). However, 1,3-PDO was used as substrate instead of glycerol. The specific 1,3-PDO oxidoreductase activity was reported as amounts of aldehydes formed per mg total protein.

3.2.6 Fermentation experiment

In shake flask experiments, the seed cultures were inoculated in 100 mL AM1 medium containing 20 g/L glycerol with or without either glucose or yeast extract at an initial OD₅₅₀ of 0.1 in 500 mL in an Erlenmeyer flask at 37°C, 200 rpm for 72 h. Samples were intermittently taken every 12 h for analyses.

E. coli NSK015 strain was grown in 100 mL LB supplemented with 2% (w/v) glucose in a 250 mL shaking flask at 37°C, 200 rpm for 12 hours. The inoculum was transferred into 1 L AM1 medium containing substrates with the initial OD₅₅₀ of 0.1 in a 2L fermenter. The solution of 3M KOH was automatically added into the broth to maintain pH of cultures at 7.0. The aeration rate of 1 vvm and agitation of 200 rpm were provided. One mL of sample was taken every 12 hours for analyses for 108 h.

The mixed cassava starch (60 g/L) and glycerol (50 g/L) in AM1 medium containing coenzyme B12 were used as substrates for studying 1,3-PDO production in a 2L fermenter. After sterilization, the solution mixture was cooled down to 60°C. Amylase (Amy) and amylo-glucosidase (AMG) at concentrations of 100 U and 600 U per gram cassava starch respectively to hydrolyse the cassava starch. The cassava hydrolysis was allowed for 4 h at 50°C for 2 h. After hydrolysis, the seed of *E. coli* NSK015 strain was inoculated into the medium with the initial OD₅₅₀ of 0.1and temperature was maintained at 37°C during fermentation.

3.2.7 Analytical method

Fermentation broth was collected and separated by centrifugating at 15,000 rpm for 5 min. The supernatant was analysed with HPLC while cell pellet was mixed with 1 mL DI water and determined the cell biomass. The biomass concentration was determined (3 $OD_{550} = 1$ g/L cell dry weight; Khunnonkwao et al. (2018)). HPLC

analysis coupled with an anion exclusion column (Bio-RAD, Aminex, HPX-87H, USA) was used to analyse sugars and fermentative products. The column was controlled at the temperature of 45°C. The solution of 4 mM H₂SO₄ was used as a mobile phase with a flow rate of 0.4 mL/min. The experiments were performed in triplicate. Average values with the standard deviation were reported. Duncan's multiple-ranges test (DMRT) was used to determine differences among mean values at 95% significance level (P<0.05).

Strains	Relevant Characteristic	Source
Plasmid	<i>, </i>	
pLOI3420	acc $\gamma \beta exo$ (Red recombinase), sensitive	Wood et al. (2005)
	replicon of temperature	
pKD4	bla ;FRT-kan-FRT cassette	Datsenko and
		Wanner (2000)
pFT-A	bla flp low-copy vector containing	Datsenko and
	recombinase and temperature-conditional pSC101 replicon	Wanner (2000)
pCR2.1-TOPO	bla, kan TOPO TA cloning vector	Invitrogen
pKJ1010	pCR2.1-TOPO-gdrB	This study
pKJ1011	pCR2.1-TOPO- gdrAB-dhaB123	This study
pKJ1012	pCR2.1-TOPO-frd'-cat-sacB-frd''	This study
pKJ1013	pCR2.1-TOPO-yqhD	This study

Table 3.1 Plasmids, strains and primers used for this study.

Strains	Relevant Characteristic	Source
pKJ1014	pCR2.1-TOPO-focA'-yqhD-pflB'	This study
E. coli TOP10		Invitrogen
E. coli C	Wild type	ATCC
KMS005	K. oxytoca $\Delta adh E \Delta ack A-pta \Delta ldh A::cat-sac B$	Jantama et al.
		(2015)
NSK001	E. coli C ΔldhA::cat- <mark>s</mark> acB	This study
NSK002	E. coli C ΔldhA::cat-sacB ΔackA::FRT-km-FRT	This study
NSK003	E. coli C ΔldhA::gdrAB-dhaB123ΔackA::FRT-km-FRT	This study
NSK012	E. coli C Δ <mark>ldhA</mark> ::gdrAB-dhaB123ΔackA::FRT	This study
NSK013	E. coli C AldhA::gdrAB-dhaB123AackA::FRTApflB::cat-	This study
	sacB	
NSK014	E. coli C ΔldhA::gdrAB-	This study
	dhaB123 Δ ackA::FRT Δ pflB::yqhD	
NSK015	E. coli C ΔldhA::gdrAB-	This study
	$dhaB123\Delta ackA$::FRT $\Delta pflB$::yqhD $\Delta frdABCD$::cat-sacB	
Primers		
Comp-	5'CGGGCTGCTACTGGCCGGTCAGGCGAATTAA	This study
gdrA(down)-	AGTGTTAACGAGGGGGACCGTC <u>ATGTCGCTTTC</u>	
gdrB-R	<u>ACCGCCA</u> 3'	

 Table 3.1
 Plasmids, strains and primers used for this study (Continued).

Strains	Relevant Characteristic	Source	
Primers			
DH-ldhAp-	5 [,] ATTGGGGATTATCTGAATCAGCTCCCCT	This study	
gdrAB-	GGAATGCAGGGGAGCGGCAAGA		
dhaB123-F	TCAGTTTCTCTCACTTAACG3,		
Comp-	5 <u>GCGTACGCCTGGCGGTGAAAGCGACAT</u>	This study	
gdrB(up)-gdrA-	GACGGTCCCCTCGTTAACACTTTAATTCG		
F	<u>CCTGACCGGCCAGT</u> 3 [,]		
DH-ldhAp-	5 AAATATTTTTAGTAGCTTAAATGTGATT	This study	
gdrAB-	CAACATCACTGGAGAAAGTCTT <u>ATGAAA</u>		
dhaB123-R	AGATCAAAACGATT3,		
dhaB1-M13-	5.CCAGTACTGCAAATCGTTTTGATCTTTT	This study	
R/F	CAT <u>CACTGGCCGTCGTTTTAC</u> 3,		
M13-dhaB1-R	5 [,] GTAAAACGACGGCCAGTG <u>ATGAAAAGA</u>	This study	
	TCAAAACGATTTGCAGTACTGG3,		
DH-ldhAp-cat-	5 [,] ATTGGGGATTATCTGAATCAGCTCCCCT	This study	
sacB-F	GGAATGCAGGGGAGCGGCAAGAACACTG		
	CTTCCGGTAGTCAA3,		

 Table 3.1
 Plasmids, strains and primers used for this study (Continued).

Strains	Relevant Characteristic	Source
Primers		
DH-ldhAp-cat-	5 [,] AAATATTTTTAGTAGCTTAAATGTGATTC	This study
sacB-R	AACATCACTGGAGAAAGTCTTCGGCACGT	
	AAGAGGTTCCAA3 [,]	
ackA-pKD4-F	5 [,] CTATGGCTCCCTGACGTTTTTTTAGCCAC	This study
	GTATCAATTATAGGTACTTCCGTGTAGGCT	
	GGAGCTGCTTC3,	
ackA-pKD4-R	5 [,] TCAGGCAGTCAGGCGGCTCGCGTCTTGC	This study
	GCGATAACCAGTTCTTCCATATGAATATCC	
	TCCTTAG3	
ackA-F	5 [,] CACCTCAACTTCACATATAAAG3 [,]	This study
JMackA-R	5GCACGATAGTCGTAGTCTGA3,	This study
yqhD-pflB-F	5 [.] CTTTCGCACGTACACCGGTGGTAACACCT	This study
	ACCTTCTTAAG3	
yqhD-pflB-R	5 [,] GTATATACGAAGCCGCCCGCTAATAGAT	This study
	TTGACTGAAATCGT3,	
pflB-yqhD-F	5 [,] ACGATTTCAGTCAAATCTATTAGCGGGC	This study
	GGCTTCGTATATAC3,	

 Table 3.1
 Plasmids, strains and primers used for this study (Continued).

Strains	Relevant Characteristic	Source
Primers		
pflA-F	5 [,] AACCGTTGGTGTCCAGACAG3 [,]	Jantama et al.,
		2008
upfocA-R	5'AGATCGCCAGCCGCTGCAAT3'	Jantama et al.,
		2008
focA-R	5 [,] ACCTGGTCGGCGCACTGCTG3 [,]	This study
WMpflBA -F	5 [,] TTACATAGATTGAGTGAAGGTACGAGTA	Jantama et al.,
	ATAACGTCCTGCTGCTGTTCTACACTGCTT	2008
	CCGGTAGTCAA3·	
frdABCD-F	5 [,] TTAGATTGTAACGACACCAATCAG3 [,]	This study
frdABCD-R	5 [,] GTTCGCAGACCGTAACTTT3 [,]	This study
IOfrd-F	5 [,] TTGGAACCTCTTACGTGCCGTCGCCACCC GCTACTGTATC 3 [,]	This study
IOfrd-R	5 [,] TTGACTACCGGAAGCAGTGTCGCGCAGA	This study
	GCTTCATTGGTC 3 [,]	
catfrd-F	5 [,] GACCAATGAAGCTCTGCGCGACACTGCT	This study
	TCCGGTAGTCAA 3 [,]	

 Table 3.1
 Plasmids, strains and primers used for this study (Continued).

Strains	Relevant Characteristic	Source	
Primers			
catfrd-R	5 [,] GATACAGTAGCGGGGGGGGGGGGGGGGGCACG	This study	
	TAAGAGGTTCCAA 3 [,]		
sequence-ldhA-F	5 [,] CACATGCTGCCGGAAATCATC3 [,]	This study	
sequence-ldhA-R	5 [,] ACGCGTAATGCGTGGGCTTTC3 [,]	This study	

 Table 3.1
 Plasmids, strains and primers used for this study (Continued).

3.3. Results and discussion

3.3.1 Functional expression of the gdrAB-dhaB123 operon for 1,3-

propanediol pathway in E. coli

K. pneumoniae, naturally utilizes glycerol to produce 1,3-PDO, it is though considered as a potential pathogen and may not be safe for operators in an industrial scale of production. Over the two past decades, all previous studies made efforts in *E. coli* as an alternative host to produce 1,3-PDO by the over-expression of heterologous genes involved in the glycerol utilization (Table 2.1). However, a plasmid instability in developed *E. coli* strains may cause a low production rate and yield of 1,3-PDO during cultivation of them on an industrial scale (Friehs, 2004). Additionally, antibiotics and IPTG are required for the maintenance of plasmids and the induction of gene expressions for 1,3-PDO production, resulting in ever-increasing production cost. To resolve these obstacles, genes involved in glycerol metabolism through 1,3-PDO producing pathway should be designed to be integrated into the *E. coli* genome and expressed under the control of one of the host constitutive promoters.



Figure. 3.5 Comparison of developed dark pink color on a modified aldehyde indicator plates containing glycerol between developed *E. coli* NSK strains and *K. pneumoniae* (positive control) and *E. coli* C as wildtype (negative control).

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In this study, the glycerol dehydratase operon (*dhaB123* genes) from *K*. *pneumoniae* subsp. *pneumoniae* TISTR1867 was simultaneously expressed with its reactivating factors (*gdrAB*) in *E. coli* NSK002. Glycerol dehydratase requires reactivating factors for glycerol conversion to 3-HPA. Since *dhaB123* and *gdrA* genes in *K. pneumoniae* are arranged in the opposite direction to *gdrB* gene (Rathnasingh et al., 2009), the re-arranged *gdrAB-dhaB123* cassette was used as a one operon (pKJ1011; Fig. 3.2). The structural *ldhA* gene of *E. coli* NSK002 was simultaneously deleted and replaced by the artificially constructed *gdrAB-dhaB123* operon (Fig. 3.1). The operon was then stably integrated in frame with the *E. coli ldhA* promoter. The automated sequencing analysis confirmed that no mutations at the *ldhA* promoter region and *gdrAB-dhaB123* gene were found (Fig. 3.4a-c) on the genome of recombinant clones. All positive clones (*E. coli* NSK003 strains) showed the dark pink colonies on the aldehyde indicator plate, thus resemble to that of *K. pneumoniae* while *E. coli* C wild type did not (Fig. 3.4d; Fig. 3.5). This result indicated that the artificially constructed *gdrABdhaB123* operon was successfully expressed and functioned in the *E. coli* NSK003 clones under the control of *ldhA* promoter and the clones accumulated 3-HPA from glycerol. This also confirmed that the strategy for the expression of the *gdrAB-dhaB123* operon under the native *ldhA* promoter was sufficient to allow a glycerol conversion to 3-HPA.

Previous studies suggested that an imbalance expression and an instability of glycerol dehydratase enzyme caused the accumulation of 3-HPA thus exhibiting cell growths and the production of 1,3-PDO (Chen et al., 2011; Kim et al., 2014). Then glycerol dehydratase was expressed on low-copy number plasmids (Rathnasingh et al., 2009) or its codon optimization was performed (Lim et al., 2016) to ensure that the glycerol conversion to 3-HPA was not too rapid by the high activity of the enzyme, while 3-HPA did not accumulate such an inhibitory level of inside cells. Based on this study s result, it is likely that the *ldhA* promoter is suitable for the

employment of a constitutive expression of the gdrAB-dhaB123 operon without the growth retardation (Fig. 3.6A) in E. coli NSK012 (a derivative of E. coli NSK003) in the shake flask experiment. It is well known that lactate dehydrogenase (LDH) is substantially present under both aerobic and anaerobic conditions with a basal level but is activated when E. coli grew on a variety of sugars under at acidic pH (Bunch et al., 1997). Additionally, E. coli NSK012 possessed the significantly higher activity of glycerol dehydratase at the level of 0.38 ± 0.07 U/mg protein compared with that of its parental *E. coli* NSK002 strain (0.26±0.05 U/mg protein). However, the background of the enzyme activity may be a result from the presence of another iso-functional diol dehydratases usually found in *E. coli* and *Klebsiella* species. Surprisingly, the specific activity of 1,3-PDO oxidoreductase catalysing 3-HPA to 1,3-PDO was significantly increased in E. coli NSK012 (0.25±0.02 U/mg protein) compared to that of E. coli NSK002 (0.17±0.01 U/mg protein) (Table 3.2) even though it was not overexpressed in the strain. This result suggested that the over-expression of glycerol dehydratase may up-regulate an enzymatic activity of 1,3-PDO oxidoreductase to catalyse the accumulated 3-HPA and to prevent its negative effects in E. coli NSK012. E. coli NSK012 produced 1,3-PDO from glycerol at the level of 4.5 g/L within 72 h incubation in a shaking flask while no 1,3-PDO production was observed by E. coli NSK002 (Fig. 3.6B and C). Acetate and succinate were major by-products. This study's success in the demonstration of the functional expression of the gdrAB-dhaB123 operon by the ldhA promoter to produce 1,3-PDO in E. coli NSK012 resembles the work of Zhang et al. (2007) in which they successfully utilized the host ldhA promoter for driving the expression of alanine dehydrogenase (*alaD*) to produce L-alanine. The engineered *E*. *coli* strain produced L-alanine at its high production yield.

3.3.2 Improved 1,3-propanediol production caused by deletion of *pflB* but not the over-expression of *yqhD*

Acetate up to 2.0 g/L with a few formate was accumulated during 1,3-PDO production by *E. coli* NSK002 and *E. coli* NSK012 strains even though both strains possessed the deletion of *ackA* (encoding acetate kinase enzyme) (Fig. 3.6C). In et al. (2020) demonstrated that the elimination of *pflB* gene in *K. oxytoca* KIS004-91T led the strain lacking the ability to produce acetyl-CoA and formic acid from pyruvate in the central metabolism, thus significantly lowering the acetate production during productions of succinic acid and lactic acid, respectively, under anaerobic conditions. After the deletion of *pflB*, *E. coli* NSK013 drastically increased the level of 1,3-PDO up to 5.8 g/L (Fig. 3.6C), which was about a 26.1% increase, compared to that of *E. coli* NSK012 (4.6 g/L).

Additionally, acetate production significantly declined by three folds in *E. coli* NSK013 (0.8 g/L), compared to that of *E. coli* NSK012 (2.4 g/L). The 1,3-PDO production yield by *E. coli* NSK013 was significantly improved to be 0.38 mol/mol glycerol used in the shaking flask conditions (Fig. 3.6B). This suggested that the deletion of *pflB* conserved the carbon flux and diverted it through 1,3-PDO pathway. The result also demonstrated that the activation of the *pflB* transcription may be relieved from an

induction of FNR and ArcAB regulatory proteins under aerobic conditions (shaking flasks in our study), even though Hasona et al. (2004) have revealed that PflB catalyses the non-oxidative cleavage of pyruvate to acetyl-CoA and formic acid under anaerobic conditions. This study's results also showed that E. coli NSK012 accumulated acetyl-CoA, a precursor of acetate, at a higher level than that of E coli NSK013 regardless of oxygen deprivation. Wagner et al. (2001) and Zhu et al. (2007) revealed that YfiD protein, a pyruvate formate-lyase activase enzyme, activated PfIB by acting as the glycyl radical domains to reconstitute the catalytic center of oxygen-fragmented PfIB, and to rapidly recover PfIB activity, thus generating ATP in the presence of oxygen even or under oxidative stress in E. coli. These researchers also suggested that E. coli utilized both PfIB and pyruvate dehydrogenase to supplement the acetyl-CoA pool under aerobic or micro-aerobic conditions.

10 oxidoreductase for different developed E. coli strains Enzyme Enzymatic activity (U/mg protein) /Strain **NSK002 NSK012 NSK013 NSK014 NSK015** K. pnuemoniae Glycerol 0.64 ± 0.11^{a} 0.26 ± 0.05^{b} $0.38 \pm 0.07^{\circ}$ $0.44 \pm 0.04^{\circ}$ 0.42±0.04° $0.27{\pm}~0.06^{b}$

Fable3.2	Enzymatic	activity	of	glycerol	dehydratase	and	1,3-propanediol
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The superscript	letters represent	the significant	difference of	values betweer	columns in th	e same row
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1,3-PDO

0.25±0.02°

0.47±0.03^d

 0.50 ± 0.07^{d}

0.24±0.07^{b,c}

 0.17 ± 0.01^{b}

dehydratase

oxidoreductase

1.11±0.13^a



Figure 3.6 Comparison of microbial growth (A) and maximum 1,3-propanediol production yield (B) of developed *E. coli* NSK strains in AM1 medium containing 20 g/L glycerol and 100 mM KHCO₃ with or without glucose (Glu) and yeast extract (YE) in shaking flasks. Fermentation profiles of developed *E. coli* NSK strains in AM1 medium containing only 20 g/L glycerol and 100 mM KHCO₃ in 250 mL shaking flasks (C).



To further increase 1,3-PDO production, *pflB* gene was simultaneously replaced by the native E. coli yqhD gene and the 1,3-PDO oxidoreductase (YqhD or PDOR) was simultaneously overexpressed. Surprisingly, the over-expression of yqhD gene did not result in an increase in 1,3-PDO titre and yield in E. coli NSK014 compared to those of E. coli NSK013 (Fig. 3.6B and C). The results of the enzymatic activity analysis also revealed that both E. coli NSK013 and E. coli NSK014 strains showed comparable levels of glycerol dehydratase activity (0.44±0.04 and 0.42±0.04 U/mg protein, respectively) compared to that of E. coli NSK012 (0.38±0.07 U/mg protein). However, E. coli NSK013 (0.47±0.03 U/mg protein) possessed the YqhD activity at a significantly higher level (2 folds increase) than that of *E. coli* NSK012 (0.25±0.02 U/mg protein) but at an equivalent level compared to that of E. coli NSK014 (0.50±0.07 U/mg protein) (Table 3.2). These results indicated that the over-expression of glycerol dehydratase and the accumulation of 3-HPA in E. coli NSK013 were able to enhance its intact YqhD activity to detoxify 3-HPA and to sufficiently complete the 1,3-PDO pathway. This was confirmed by that the yqhD was overexpressed in E. coli NSK014 though the YqhD activity was still comparable to that of E. coli NSK013 (Table 3.2). No further increase in 1,3-PDO yield (Fig. 3.6B and C) was thus observed in E. coli NSK014. Our results were in an accordance with the study of Emptage et al. (2003) in which they also demonstrated that *E. coli* strain was more preferrable to utilize its intact YqhD to efficiently converted 3-HPA to 1,3-PDO. The further overexpression of a dual 1,3-PDO

dehydrogenase/oxidoreductase (encoded by *dhaT*) or similar dehydrogenases from *Klebsiella* species did not substantially enhance the 1,3-PDO yield.

The conversion of 3-HPA by the activity of PDOR may be considered as a rate-limiting step in the 1,3-PDO biosynthesis (Jiang et al., 2015). The availability of reducing powers, NADH or NADPH, during the reduction of 3-HPA by PDOR activity, may additionally govern the reaction. Yun et al. (2018) investigated the biotransformation of 3-HPA derived from the whole cell C. butyricum YJH-09 that cocultured with E. coli BL21 overexpressing dhaT. The 1,3-PDO concentration and yield at levels of 20.8 g/L and 0.51 mol/mol glycerol were obtained, respectively, while C. butyricum YJH-09 alone produced 1,3-PDO at the concentration and yield of only 11.7 g/L and 0.28 mol/mol, respectively. However, the supplement of exogeneous NADH to the co-culture enhanced the metabolic shift toward reduction of 1,3-PDO oxidoreductase and improved 1,3-PDO yield from the mixture of glycerol and glucose. The concentration and yield of 1,3-PDO were increased up to 25.9 and 0.65 mol/mol glycerol, respectively. Yang et al. (2018) also overexpressed the NADP⁺-dependent glyceroldehyse-3-phosphate dehydrogenase to increase NADPH pool in E. coli JA11. The strain synthesized 13.5 g/L 1,3-PDO and could enhance two folds of 1,3-PDO yield (0.64 mol/mol) from glycerol/glucose mixture compared to its parental strain. Not only enhancing the activity of PDOR but also the increasing NAD(P⁺)H/NAD ratio must be employed to detoxify 3-HPA through 1,3-PDO biosynthesis pathway.

3.3.3 1,3-propanediol production is associated with a bicarbonate availability and succinate production in *E. coli* NSK strains

Surprisingly, growth and glycerol assimilation of all engineered strains used in this study were impaired in AM1 medium without supplementation of 100 mM bicarbonate (Fig. 3.7). The glycerol was rarely consumed by all strains compared to those cultivated with AM1 medium containing bicarbonate (source of CO₂). This implied that bicarbonate was essential for the growth and utilization of glycerol during 1,3-PDO production for the strains. This speculation was supported by the better growth of *E. coli* NSK012 than those of *E. coli* NSK derivatives that possessed the *pf1B* deletion (Fig. 3.7). Dharmadi et al. (2006) revealed that glycerol fermentation was associated with an availability of CO₂ (produced by pyruvate by the activity of Pf1B) via the oxidation of formic acid by the formate-hydrogen lyase enzyme under acidic conditions in *E. coli* regardless of oxygen availability. The produced CO₂ derived from formate was usually utilized by the reductive branch of TCA cycle via carboxylation of PEP for succinate. The researchers also demonstrated that the consumed glycerol by *E.*

coli was generally converted to ethanol and succinate due to its highly reduced nature of glycerol. Though, succinate was rarely formed in aerobic conditions but was efficiently produced under anaerobic conditions (Jantama et al., 2008). They also demonstrated that *E. coli* KJ122, a derivative of *E. coli* C, efficiently produced succinate when KHCO₃ was supplemented into the fermentation medium under both aerobic and anaerobic conditions. This confirmed that CO_2 had a key role for PEP carboxylation to generate

OAA in the reductive branch of TCA cycle. McClosky et al. (2018) also indicated that the production of succinate was relatively high in *E. coli* C wild type compared to other *E. coli* wild type strains regardless of level of oxygen. This may also support that *E. coli* NSK014 as a *E. coli* C derivative diverted glycerol through the succinateproducing pathway. Based on this study s results in Fig. 3.6C, *E. coli* NSK014 produced up to 4.3 g/L succinate as a major by-product, which accounted for about 20.7% of the carbon recovery during fermentation. It was likely that 1,3-PDO biosynthesis by developed strains was also the CO₂ dependent process that concomitantly produced succinate via the reductive branch of TCA cycle to maintain the overall redox balance. This was reflected by a co-production of reduced compounds including 1,3-PDO and succinate during glycerol fermentation.

To further conserve more carbon flux and NADH through 1,3-PDO route, the *frdABCD* genes were therefore deleted from *E. coli* NSK014. The resulting strain, *E. coli* NSK015, did not exhibit an impaired growth. However, the 1,3-PDO yield of 0.43 mol/mol was improved compared to that of *E. coli* NSK014 (Fig. 3.6B). Additionally, succinate was also dramatically abolished in *E. coli* NSK015 (Fig. 3.6C). The results suggested that *E. coli* NSK015 possessed 1,3-PDO pathway as a major route for NADH or NADPH reoxidation. Even though *E. coli* NSK015 lacking FRD activity, the glycerol utilization by the strain in the medium required the supplementation of bicarbonate (Fig. 3.5). Dharmadi et al. (2006) suggested that bicarbonate or CO₂ was essentially required when glycerol was metabolized during the growth of *E. coli* for the

biosynthesis of macromolecules for anabolism. Lacoursiere et al. (1986) and Repaske and Clayton (1978) also found that *E. coli* grew without a lag period under anaerobic or microaerobic conditions if CO₂ was provided at the suitable level. The metabolic analysis in *E. coli* by Cintolesi et al. (2012) also disclosed that activities of glycerol dehydrogenase and dihydroxyacetone kinase increased under aerobic conditions when using glycerol as substrate with the CO₂ availability resulting in increasing rates of glycolytic flux and glycerol utilization.



Figure 3.7 Fermentation profiles of 1,3-propanediol production by developed E. coli NSK strains from 20 g/L glycerol in AM1 medium without 100 mM KHCO₃ in 500 mL shaking flasks.

3.3.4 Improved 1,3-propanediol yield by the supplementation of glucose but not yeast extract

The yield of 1,3-PDO production reached 0.43 mol/mol by E. coli NSK015. However, the yield did not yet approach the theoretical maximum. Tong and Cameron (1992) suggested that the maximum theoretical yield of 1,3-PDO varied in the range of 0.67-1.0 mol/mol depending on the substrate used. This also depended on the availability of external sources of reducing powers (derived from glucose, xylose, or H₂) and nutritional constraints during fermentation (Stephanopoulos et al. 1998; Tong et al, 1991). This study at first hypothesized that yeast extract serving as building blocks and additional reducing equivalents may be required by the developed strains during 1,3-PDO production. An addition of yeast extract (5 g/L) into AM1 medium resulted in a two folds increase of biomass up to 3 g/L (OD₅₅₀ \approx 9.0) in both E. coli NSK014 and NSK015 strains compared to those without yeast extract (Fig. 3.6A). However, the 1,3-PDO yields and concentrations by both strains were not improved (Fig. 3.8A and B) when compared to those without yeast extract (Fig. 3.6B and C). Further investigation was done by the addition of both yeast extract and glucose (20 g/L) into AM1 medium containing glycerol. The 1,3-PDO concentrations and yields (Fig. 4.8C and D) were dramatically improved up to 0.94 mol/mol for both strains, accounting for two folds increases (Fig. 3.6B). This clearly indicated that glucose had much more affected the 1,3-PDO production over yeast extract. Further, yeast extract was removed from the medium to observe the sole effect of glucose on 1,3-PDO biosynthesis. The yields from both strains (0.90-0.92 mol/mol) in the medium containing both glucose and glycerol as co-substrate were also comparable to those with the supplementation of yeast extract (Fig. 3.6B and Fig. 3.8E and F) while no production of 1,3-PDO was observed in the medium containing only glucose (data not shown). An enhanced 1,3-PDO yield by glucose may be explained by two possibilities. First, glucose may serve as a source of carbon intermediates during glycolysis thus generating ATP and NADH for biosynthesis. This may result in a reduced carbon partitioning from glycerol through the central metabolism for biosynthesis and may elevate the direct conversion of glycerol through 1,3-PDO pathway. Second, glucose may be partially utilized by hexose monophosphate (HMP) shunt to generate NADPH as a preferred reducing equivalent coupled in the reduction of 3-HPA to 1,3-PDO by the activity of YqhD (Fig. 3.1). The additional reducing equivalents derived from glucose in both glycolysis and HMP pathways contributed to the high yield of 1,3-PDO closed to the theoretical maximum, 1.0 mol/mol glycerol used. Lee et al. (2010) discovered that YqhD had an alternative activity of an NADPH-dependent aldehyde reductase that involves in glyoxal and aldehyde detoxification and was a part of a glutathione-independent response to a lipid peroxidation. Additionally, Alkim et al. (2015) suggested that YqhD appeared to be the major glycolaldehyde reductase in E coli that is required NADPH for an optimal activity. Both activities may also attribute to the reduction reaction of 3-HPA to 1,3-PDO.



Figure 3.8 Fermentation profiles of developed *E*.*coli* NSK strains in AM1 medium containing only 20 g/L glycerol and 100 mM KHCO₃ in 500 mL shaking flasks)A-F (and in AM1 containing 50 g/L co-substrate of glycerol and glucose)G (or cassava starch)H (in 2 L fermenter .Gly; glycerol, Gluc; glucose and YE; yeast extract.



Figure 3.8 (Continued).

3.3.5 1,3-propanediol production from co-substrate of glycerol and glucose or cassava starch in a 2 L fermenter by E. coli NSK015

The production of 1,3-PDO was evaluated in AM1 medium containing 5% (w/v) co-substrate of pure glucose and glycerol at the equimolar ratio by E. coli NSK015 in a 2L scale. Glucose was exhausted after 96 h incubation while glycerol remained (Fig. 3.8G). The level and yield of 1,3-PDO at 36.8 g/L and 0.99 mol/mol glycerol utilized respectively, were produced with acetate as a sole by-product of 5.0 g/L. This 1,3-PDO yield was reached the theoretical maximum (1.0 mol/mol) defined by the proposed equation of 2Glycerol + glucose \rightarrow 2(1,3-PDO) + 2 acetate + formate by Tong and Cameron (1992). Comparison to previous works that utilized the mixture of

G.

glycerol and glucose for 1,3-PDO, Skraly et al. (1998) reported that 1,3-PDO was produced at 6.3 g/L with the yield of 0.82 mol/mol by E. coli over-expressing K. pneumoniae dha operon harbouring on pTC53 plasmid. Yang et al. (2018) produced 1,3-PDO at the level of 13.5 g/L with the yield of 0.64 mol/mol by *E. coli* carrying two inducible plasmids harbouring *dhaB-gdrAB-yqhD* and *gapN-galP-glk* operons. Yun et al. (2021) demonstrated the co-culture of E. coli Rosetta-dhaB12 and E. coli BL21 overexpressing *dhaT* in the medium containing glucose/glycerol mixture. The yield of 1,3-PDO at 0.67 mol/mol was only achieved. Tang et al. (2009) reported the yield of 1,3-PDO yield at 1.09 mol/mol when overexpressing C. butyricum dhaB12 operon and *E. coli yqhD* harbouring on an inducible plasmid. However, they reported the yield was over the theoretical maximum due to the use of yeast extract that may provide an additional source of reducing powers during fermentation. Additionally, on the contrary to our work in this study, all previously developed strains required plasmid-dependent systems to overexpress heterologous genes required for 1,3-PDO production pathway. Antibiotics and IPTG, and even yeast extract were mandatory for enhancing growth, preventing the plasmid instability, and maintaining enzymatic activities essentially for 1,3-PDO production. To this end, the production cost may be increased.

Cassava starch is a cheap agricultural product containing glucose subunits. It has been used as a carbon substrate for sustainable productions of biochemicals including succinate, lactate and 2,3-butanediol (Khor et al., 2016; In et al., 2020; Khunnonkwao et al., 2020). Therefore, *E. coli* NSK015 was elucidated an ability to cometabolize glycerol and cassava starch as a source of glucose for the of biosynthesis 1,3-PDO to reduce production cost related to the use of a pure glucose. The result revealed that the use of hydrolysed cassava starch with glycerol did not affect the microbial growth. The cell growth was similar to that observed when a pure glucose was used in which the maximum biomass was reached 2 g/L after 36 h. Glucose was exhausted within 84 h while about 10 g/L of glycerol remained. However, 1,3-PDO concentration was reduced to 31.9 g/L with the yield of 0.84 mol/mol. Acetate was produced with a two-fold increase as a main by-product (11.4 g/L) compared to that of a pure glucose (5.5 g/L) (Fig. 3.8G and H). The glucose consumption rate by the strain in the condition of cassava starch was significantly higher than that of glycerol. This may be the effect of nitrogen sources in the cassava starch thus accelerating glucose consumption. The rapid consumption of glucose derived from cassava starch caused a high glycolytic flux from glycolysis through the acetate-producing pathway to prevent the redox imbalance (high NADH/NAD ratio) and resulted in the carbon wasting to an acetate excretion rather than the biosynthesis of 1,3-PDO. It also affected the generation of NAD(P)H via HMP pathway thus depleting 1,3-PDO production. With our results, E. coli NSK015 possessed heterologous enzymes over-expressed on its genome essentially for the biosynthesis of 1,3-PDO and offered the highest yield and concentration in the mineral salts medium containing glucose or cassava starch and glycerol without rich nutrients (yeast extract and peptone) compared to those of strains published in previous literature (Table 2.1).

3.4. Conclusions

This is a pioneer work reported the 1,3-PDO yield at 99% of the theoretical maximum by a metabolically *E. coli* NSK015 over-expressing heterologous genes devoid of strong and inducible plasmids-expression systems. Antibiotics and IPTG were not required for plasmid maintenance and gene induction. A mineral salts AM1 medium without rich nutrients was used to produce 1,3-PDO thus reducing costs related to medium preparation, separation and purification, and waste disposal. The substitution of pure glucose by cassava starch was also demonstrated for efficiently biosynthesizing 1,3-PDO. *E. coli* NSK015 may be one of the potential microbial factories for the industrial production of 1,3-PDO.



CHAPTER IV

OPTIMIZATION OF METABOLICALLY ENGINEERED Escherichia coli NSK015 FOR PROGRESSIVE IMPROVEMENT OF 1,3-PROPANEDIOL PRODUCTION

4.1 Introduction

The productivity of 1,3-Propanediol or 1,3-PDO from previous study was still not impressive (0.34 gLh). To improve 1,3-PDO productivity and concentration, optimization process was then applied in this study. Various parameters including cosubstrate utilization, yeast extract, aerations, inoculum size, pH and fed-batch mode were adjusted for improving 1,3-PDO production by *E. coli* NK015 strain. Different forms of the cofactor, including vitaminB12 and coenzymB12 were also evaluated thus affecting 1,3-PDO production. The difference of space between two Rushton impellers was observed upon the effect of productivity. Additionally, different agitation speeds with increasing kLa values were employed to find the best conditions. Mode of fed-batch fermentation was finally designed to avoid the substrate inhibition and achieved the high 1,3-PDO production. These optimized conditions would be further applied and being the start point in the production of 1,3-PDO by metabolically engineering *E. coli* in larger scale.

4.2 Methods and materials

4.2.1 Microorganism

E. coli NSK015 was mainly used for 1,3-PDO production in this study. Briefly, *E. coli* C was metabolically engineered by replacing *gdrAB-dhaB123* operon in the region of *ldhA*. The major by-product genes, including *ackA* (acetate kinase), *frdABCD* (fumarate reductase), *pflB* (pyruvate-formate lyase) were removed for eliminating by-products formation.

4.2.2 Medium and culture conditions

A few colonies of *E. coli* NSK015 was re-streaked on LB agar containing 20 µg/mL chloramphenicol and incubated at 37°C for 16 h. To prepare inoculum for fermentation in the fermenter, one full loop of *E. coli* NSK015 was then cultured in 250 mL Erlenmeyer flask containing 100 mL LB broth and 20 µg/mL chloramphenicol. The inoculum was agitated at 200 rpm, 37°C. After 16 h, the inoculum was following provided in the fermenter with total final working volume 1100 mL of AM1 medium by setting the initial OD₅₅₀ of 0.1 in fermenter. Aeration was provided by air flowing through sterilized filters and constantly provided at 1.0 vvm. pH was monitored at 7.0 by adjusting with 3M KOH automatically. The percentage of dissolved oxygen (DO) was determined by DO probe detector connecting to monitor of fermenter and must be calibrated before performing fermentation. The initial DO before adding microorganism was therefore 100%. Luria-Bertani (LB) agar and broth containing 5 g/L yeast extract, 5 g/L NaCl, and 10 g/L tryptone was routinely used for culturing *E. coli* NSK015. AM1 medium containing 19.92 mM (NH₄)₂HPO₄, 7.56 mM NH₄H₂PO₄, 2.0 mM KCl, 1.50 mM MgSO4·7H₂O, 1.0 mM Betaine·KCl, 1X Trace metal (Martinez et al., 2007), and CoenzymeB12 (7.5-30 μ M depending on experiment) was routinely used as a medium for fermentation and optimization. To reduce foam forming in the fermentation, 10%(v/v) antifoam (Antifoam Y-30 Emulsion, A5758, Sigma) was prepared. The 1-2 mL of the diluted antifoam was added directly to the fermenter when the presence of foam.

4.2.3 Batch fermentation

To find the suitable conditions for enhancement of 1,3-PDO concentration and productivity, different parameters were verified by performing batch fermentation. The optimization was divided into 2 phases according to the number of Rushton turbine used for agitation. The first phase depended on using only one Rushton turbine for agitation (Fig. 4.1A). Vitamin B12 was initially replaced to coenzyme B12 for reducing the cost and examining the possibility to use vitamin B12 instead of coenzyme B12. In the second phase of optimization, two Rushton turbines were adjusted and used for agitation. The effect of speed levels and k_La were varied at 200, 300, and 400 rpm. To further reduce the cost of coenzyme B12, the varied amounts of coenzyme B12 at 7.5 and 15 μ M were then observed the effect upon 1,3-PDO production at the optimized speed. All batch fermentations were performed in duplicate. Fermentation was carried out in 2 L fermenter (BIOFLO 110, New Brunswick Scientific, USA)

containing 1100 mL of working volume. In optimization step, pH, temperature, and aeration were constantly set at 7.0, 37°C, and 1.0 vvm, respectively, throughout experiments.

4.2.4 Fed-batch fermentation

Fed-batch fermentation was performed in 2 L fermenter (BIOFLO 110, New Brunswick Scientific, USA) with 1 L of working volume. The inoculum was fed into the fermenter with the final OD_{550} of 0.1. The agitation speed of 300 rpm and the use of 2 impellers for agitation (Fig. 4.1B) were used as the controlled parameters in the fed-batch fermentation. Different modes of co-substrate feeding were designed as shown in Table 4.3 and investigated the suitable condition. Substrate stock was prepared with two different ratios. The first substrate stock was prepared by using the same ratio of glycerol and glucose at 50 g: 50 g in a total volume of 180 mL (mild continuous mode). The second substrate stock used for mode 2 to mode 4 (low continuous, high continuous, and 2 continuous-pulsed modes) was prepared by using the ratio of glycerol/glucose at 1.2:1.5 according to the substrate uptake rates in the optimized batch fermentation. All component of AM1 were also combined in the substrate stock with a half of concentrations that was usually used in the batch fermentation. Coenzyme B12 was provided additionally in the substrate stock by the ratio of 50 g/L glycerol:7.5 µM coenzyme B12. To provide substrates during fermentation, time at 24 h was selected as the initial feeding.

4.2.5 Determination of k_La

In order to optimize the production of 1,3-PDO, k_La or volumetric mass transfer coefficient was evaluated as an important parameter to improve concentration and productivity of 1,3-PDO by using the equation expression as following;

$$k_La = ABS(ln(1-(C/C*)))$$

where C* is the saturation concentration of the oxygen and C is the actual concentration of the oxygen. The k_La at 1.0 L working volume of AM1 medium with 50% (w/v) glycerol and glucose was determined under different agitation speed of 200, 300 and 400 rpm.

4.2.6 Analytical method

Sample was taken twice per day to determine metabolites and cell growth. One mL of sample was divided and centrifuged at 12,500 rpm for 4 min. Cell pellet was then resuspended with 1 mL of DI water and measured OD₅₅₀ by spectrophotometer (Spekol®1500). Whereas, the supernatant used for further analysis of metabolites by HPLC. Biomass was finally estimated from an equation derived from the graph of OD₅₅₀ plotting against dry mass. Approximately, 2.68 OD₅₅₀ equals to one gram of biomass.

To measure the metabolites, 200 microliters of sample supernatant derived from centrifugation were diluted with 800 μ L of 20 mM H₂SO₄ and passed through the 0.2 μ m filter membrane before analysis. The Rezex ROA organic acid H+ (8%), 250×4.6 mm phase-reverse column (Phenomenex, France), connecting to a

Refractive Index detector was used with HPLC for separation of metabolites. The temperature of column was kept constantly at 30°C and used with 10 mM H_2SO_4 as a mobile phase at a flow rate of 170 µL per min.

4.2.7 Statistic analysis

The batch fermentations were carried out in duplicate. The data values were reported with standard deviations and their significant differences were determined by SPSS. Duncan's algorithm was selected to compare the significant differences between mean values of individual groups under different conditions.



Figure 4.1 Schematics of different impeller localizations in 2 L fermenters. A: Type 1 model using one impeller for agitation and B: Type 2 model using two impellers for agitation (performed at INP, France).
4.3 **Results and discussion**

4.3.1 Effect of vitamin B12 and coenzymeB12 on 1,3-propanediol production

In our previous study, coenzyme B12-dependent glycerol dehydratase genes were overexpressed under ldhA promoter to produce 1,3-PDO in the recombinant *E. coli* NSK015. This strain requires coenzyme B12 serving as a cofactor of the enzyme for converting glycerol to 3-HPA. However, E. coli cannot synthesize vitamin B12 or coenzyme B12 by itself. Out of around 30 genes in native coenzyme B12 producers like *Samonella typhimurium*, only two-third of certain genes involving in the cobalamin synthesis are found in E. coli (Roth et al., 1993). As a result, cobalamin or coenzymeB12 is generated only the presence of the complex intermediate, cobinamide (Lawrence and Roth, 1995). Both vitamin B12 and coenzyme B12 have the same core structure with corrin ring. Only the upper part of the structure, especially, adenosyl group, is different and plays a key role for catalysis of glycerol to 3-HPA. Though, the use of coenzyme B12 may thus not be worthy to an economical production of 1,3-PDO as the price for coenzyme B12 was more expensive than vitamin B12 about 3 folds. DuPont used vitamin B12 for supplementing the medium for 1,3-PDO production instead.

To observe the possibility to use vitamin B12 as a cofactor, this study was conducted to prove whether btuR (adenosyltransferase gene), one of cobalaminutilizing genes in *E. coli* functions in converting vitamin B12 to be an active adenosylated form of vitamin B12 (Rosnow et al., 2018). It thus could further activate the glycerol dehydratase activity. The same fermentative condition of 200 rpm agitation, 1 vvm aeration with 15 μ M vitamin B12 was carried out to evaluate the efficiency of vitamin B12 as a cofactor for coenzyme B12-dependent glycerol dehydratase genes (Fig. 4.2A). The result showed that 1,3-PDO was produced up to 19.3 g/L (yield=0.85 mol/mol, Table 4.1). At the end of fermentation (168 h), 1,3-PDO concentration was not increased and lower than that of the coenzyme B12 (around 30 g/L) (Fig.4.2B). This implied that *btuR* was capable of catalyzing the adenosylation of vitamin B12. However, its activity may still not strongly enough to efficiently generate the adenosylated vitamin B12 reaching the suitable level for supporting the full activity of glycerol dehydratase compared to those of directly utilizing coenzyme B12 thus delaying 1,3-PDO production. To further improve 1,3-PDO production in the future study, *btuR* may be overexpressed to enhance its enzymatic level and activities.





Figure 4.2 Effect of vitaminB12 and coenzymeB12 as a cofactor in the fermenter with type 1 of impeller localization for 1,3-propanediol production with 200 rpm agitation and 1.0 vvm aeration. A. vitaminB12. B. coenzymeB12.



Conditions	$k_L a \ (h^{-1})$	Max Biomass (OD550)	Time for observation	1,3- PDO (g/L)	Malate (g/L)	Acetate (g/L)	Glycerol residue (g/L)	Glucose residue (g/L)	Yield (mol/mol)	Productivity (g/L/h)
Coenzyme B12ª	ND	6.01±0.03	108	36.8±0.8	ND	5.5±0.3	5.7±0.8	-	0.99±0.04	0.34±0.01
Type 1 of impeller, Vitamin B12	19.61±3.17	5.91±0.72	102	19.3±1.3	10.3±3.1	3.0±0.9	23.9±10.4	5.2±3.4	0.85±0.25	0.19±0.01
Type 1 of impeller, Coenzyme B12	22.01±6.61	5.92±2.0	168	29.5±1.3	8.00±1.3	6.4±2.7	5.5±5.3	2.7±1.1	0.84±0.09	0.18±0.01

Table 4.1Comparison of fermentation profile in type 1 of impeller localization with 200 rpm agitation and 1.0 vvm aeration.

^a Data obtained in the previous chapter (SUT, Thailand) with different fermenter geometry. ND=Not determined

Condition	$k_L a \ (h^{-1})$	Maximum Biomass (OD550)	Time	1,3-PDO (g/L)	Malate (g/L)	Acetate (g/L)	Glycerol residue (g/L)	Glucose residue (g/L)	Yield	Productivity (g/L/h)
200 rpm, 15 μM	14.48±1.72	4.20±0.13ª	102	25.8±3.8 ^a	13.6±2.7ª	2.4±0.8 ^a	6.9±7.1ª	5.3±5.5 ^{a,b}	0.70±0.12 ^a	0.25±0.04 ^a
300 rpm, 7.5 μM	33.62±4.53	11.64 ± 0.05^{b}	48	38.1±4.4 ^b	5.2±3.7 ^b	5.78±0.1 ^{a,b}	3.1±1.1 ^a	2.2±2.0 ^{a,b}	0.93±0.08 ^a	0.79±0.09°
300 rpm, 15 μM	33.62±4.53	9.65±0.14 ^b	48	35.9±0.8 ^b	11.1±0.8 ^c	4.4±0.5 ^{a,b}	7.7±0.1 ^a	1.1±1.6 ^a	0.88 ± 0.18^{a}	0.75±0.02 ^c
400 rpm, 15 μM	59.33±3.51	21.65±3.13 ^c	48	22.7±0.8ª	0.70 ± 1.0^{d}	8.3±4.0 ^b	23.0±0.8 ^b	11.3±1.1 ^b	0.80 ± 0.09^{a}	0.47 ± 0.02^{b}

Table 4.2 Comparison of fermentation profile in type 2 of impeller localization with different coenzyme B12

^(a-c)The superscript represents the significant difference of individual average (p < 0.05) in different conditions.



4.3.2 Effect of k_La and localization of impeller

k_La, one of important parameters, was used for an optimization in this study. It represents how much oxygen is being transmitted into the medium and diffusing into the microbial cell in that specific system. The presence of oxygen is known as the importance for cell metabolism and viability (Rao et al., 2009). Based on a previous study, 1,3-PDO concentration of 36.8 g/L was obtained within 108 h with productivity of 0.34 g/L/h and yield of 0.99 mol/mol. To improve productivity and concentration, E. coli NSK015 was carried out to observe 1,3-PDO production at the same speed and aeration (200 rpm, 1 vvm, 37°C, k_La of 22.01 h⁻¹) in Type 1 fermenter (INP, France) as previously performed in the previous chapter in the different geometry fermenter performed at SUT, Thailand (unknow kLa). The results confirmed that both concentration and productivity did not seem satisfactory. The 1,3-PDO at the concentration of 29.5 g/L (Fig.4.2B) were lower than that (36.8 g/L) of the previous study (Fig.3.8A). Furthermore, the total fermentation time was longer (168 h), leading the decrease in productivity about 50% (0.18 g/L/h) than that of previous experiments (Chapter 3).

To make the controlled condition similar to the previous study, number of impellers were adjusted in which two impellers of Rushton turbines (Type 1) were used instead of one impeller as a Type 2 fermenter (Fig.4.1). The concentration of 1,3-PDO reached 25.8 g/L within 102 h, (Fig. 4.3). The productivity was improved about 39 % from 0.18 g/L/h to 0.25 g/L/h. (Table 4.1 and 4.2). Though, k_La at 14.48 h⁻¹ in type 2 of impeller fermenter seemed lower than type 1 fermenter. However, productivity was not the same to previous study at 0.34 g/L/h. The possible reasons may be the geometry of fermenters and baffles were not the same, leading to the difference of vortex flow and influencing 1,3-PDO production (Rao et al., 2009; Echaroj et al., 2020)

Further, k_La optimization by increasing agitation speeds to 300 and 400 rpm, by maintaining aeration of 1 vvm and temperature of 37°C as shown in Fig.4.4 was performed. By increasing speed from 200 to 300 rpm (33.62 h⁻¹), 1,3-PDO concentration (35.9 g/L) was produced within the short time at 48 h, resulting in the improvement of almost 4 folds of productivity (0.75 g/L/h) and a 50% increase of biomass compared to those of 200 rpm (0.25 g/L/h) (Table 4.2). An increasing k_La of 59.33 h⁻¹ at 400 rpm was further studied. In contrast, 1,3-PDO concentration (22.7 g/L) and productivity were not high as expected, compared to those of 200 and 300 rpm, respectively. Only cell density was risen dramatically to 21.65 of OD₅₅₀ within 16 h and then dropped to 13.65. Acetate was produced around 10 g/L which was higher than that of 300 rpm. At the same time, the DO value of 400 rpm (Fig.4.5E) was increased up to 100% rapidly from 16 to 86 h. This may reflect that carbon substrates were overflown to glycolysis with the high metabolic rate and built up only biomass, making the rapid reduction of DO (dissolve oxygen) within 16 h. After that, the metabolic rate was then slow down reflected by the lower requirement of oxygen than that supplemented to the fermenter. As a reduced metabolic rate, the production of 1,3-PDO was low. These results demonstrated that products are not always obtained with high concentration along with the increasing agitating speed. Either the higher speed or the increasing level of DO may result in the shear force, leading to the degradation of microbial cell and less formation of product thus impacting upon yield and productivity (Audet et al., 1996; Rao et al., 2009)

At 300 rpm, the condition seemed to be suitable. Even this result was different to several previous reports in which they indicated that anaerobic or microaerobic conditions were the most suitable conditions for 1,3-PDO in either *K. pneumoniae* or *L. reuteri* (Chen et al., 2003; Huang et al., 2002; Ju et al., 2020). This was likely due to that glycerol dehydratase has ever understood that it functions effectively under anaerobic conditions (Forage and Foster, 1982). In contrast, Lama et al. (2020) recently reported that *K. pneumoniae* could efficiently produce 1,3-PDO under aerobic conditions from glucose. This study also used a similar condition at 300 rpm and confirmed that 1,3-PDO can be produced under aerobic conditions. As a result, the condition of 300 rpm, was selected and determined the glycerol and glucose uptake rates for further conducting the fed-batch fermentation.

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Figure 4.3 Comparison of 1,3-propanediol production at different kLa in type 1 and type
2 of impeller localization at 200 rpm, 1 vvm. A: 1,3-propanediol production,
B: cell growth, C: glycerol consumption, and D: glucose consumption.



Figure. 4.4 1,3-PDO profiles under different k_La in type 2 of impeller localization (A-C); A. 14.48 h⁻¹ or 200 rpm, B. 33.62 h⁻¹ or 300 rpm, and C. 59.93 h⁻¹ or 400 rpm.



Figure 4.5 Comparison of 1,3-PDO production under different k_La. (A) cell growth (B) 1,3-PDO production (C) glucose consumption (D) glycerol consumption (E) %DO (F) yield, concentration, and productivity.

4.3.3 Effect of concentration of coenzyme B12 with optimal kLa

Figure. 4.6 Comparison of different coenzyme B12 concentrations with optimal k_La. A. 15 μM coenzyme B12 in 300 rpm, B. 7.5 μM coenzyme B12 in 300 rpm.

As mentioned, coenzyme B12 is required to produce 1,3-PDO when non-native producer like *E. coli* was used as a host. It is necessarily provided coenzyme B12 into the fermentation broth. This eventually results in the increment of cost in the industry production. Several publications reported about 1,3-PDO or 3hydroxypropionic acid (3-HP) production using the different coenzyme B12 concentration in range of 1.5-50 μ M (Chu et al., 2015; Huang et al., 2017; Kim et al., 2014; Kwak et al., 2013; Wang et al., 2007; Yang et al., 2018). However, these works supplied yeast extract into the medium that may contain vitamin B12 thus partially supporting the function of glycerol dehydratase. Based on the determination of glycerol dehydratase *in situ* studied by Honda et al. (1980), the concentration of 15 μ M coenzyme B12 was brought maximum to verify the activity in this study. To reduce the cost in the industrial scale, concentrations of coenzymeB12 at 7.5 and 15 μ M were verified during the production of 1,3-PDO by the strain in the optimized agitating speed of 300 rpm and the aeration of 1.0 vvm. The results showed that the reduction of coenzyme B12 amount from 15 to 7.5 μ M did not affect the production of 1,3-PDO, even though yeast extract did not provide in the fermentation. The concentration of 1,3-PDO and productivity derived from 7.5 μ M coenzyme B12 (38 gL) was not different significantly from 15 μ M (36 g/L) (Fig. 4.6). This result seemed that coenzyme B12 was used with smaller amount in comparison to the work of Zhang et al. (2006) that 1,3-PDO could be achieved up to 43.9 g/L in the presence of 61.8 mg/L glycerol, 6.2 g/L yeast extract, and 49 mg/L vitamin B12. As a result, our study demonstrated that it can probably reduce the cost of 1,3-PDO production by reducing the concentration of coenzyme B12 in addition the absence of yeast extract.

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Mode	Feeding time)	Feeding type	Stock of glycerol (g)/glucose(g)	Fed volume (mL)	Glycerol feeding rate (g/h)	Glucose feeding rate (g/h)
1	24-88 = 64 h	Mild continuous	50/50	180	0.78	0.78
2	24-144 = 120 h	Low continuous (LC)	60/70	150	0.5	0.58
3	24-144 = 120 h	High continuous (HC)	130/160	315	1.08	1.33
4	24-56 = 32 h/ 80-112 = 32 h	2 continuous - pulsed (2CF)	70/96	190	1.10	1.50

Table 4.3Detail of different modes of co-substrate feeding.

4.3.4 Improvement of 1,3-PDO concentration by fed-batch fermentation

To improve the concentration of 1,3-PDO, a high concentration of substrate should be provided into the fermentation. However, *E. coli* cannot typically tolerate the high concentrations of substrates. Fed-batch fermentation, one of effective approach to reduce the osmotic pressure from high concentrations of substrates was then adopted by gradually feeding substrates. Different modes of co-substrate feeding as shown in Table 4.3 were designed to observe the effect upon the productivity, the concentration of 1,3-PDO, and the remained glycerol in the fermenter.

At the time of 24 h, the beginning time point of stationary phase in the optimized batch fermentation (Fig. 4.7) was used as the initial time for feeding. Glycerol and glucose feeding rates were determined based on glycerol and glucose uptake rates in batch fermentation between 1.2 g/h and 1.5 g/h, respectively. Co-substrate feeding was then divided into two main modes including a constantly continuous feeding and the two continuously pulsed feeding (Fig. 4.7). The mode of constantly continuous feeding was further divided into 3 different modes including mild, low, and high feeding rates. For the pulsed feeding, substrates were fed at a continuously high rate but paused in the certain periods.



Figure 4.7 Fed-batch fermentation with the different co-substrate feedings. (A) mild
(B) low and (C) high rates of co-substrate feedings. (D) 2 continuous-pulsed feeding (1) glycerol and glucose concentrations during cultivation. (2) metabolites and cell growth.

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In the mild feeding as shown in Fig.4.7A., glycerol and glucose were designed and limited to the same ratio of 50 g and 50 g in 180 mL of stock solution, identical to the initial concentration of co-substrates in the fermenter. Then, the stock solution of glycerol and glucose was fed at the rate of 0.78 g/h, respectively from 24 h till to 88 h. At the end of fermentation of 160 h, 1,3-PDO concentration was reached to 64.8 g/L (Fig.4.7A), accounting for about 70% increment, while the productivity (0.41 g/L/h) was not improved but was lower than that of the batch conditions at 0.75 g/L/h.

Glycerol had remined at 18.6 g/L that was higher than the glucose concentration (Table 4.4). The possible reason may be explained that the glucose consumption rate in aerobic conditions was slightly higher than glycerol. This led the reducing power not sufficient and not efficiently re-oxidized, causing no further glycerol consumption.

Feeding rates were re-assessed by increasing the glucose concentration higher than the glycerol concentration and dividing it into different modes (LC, HC, and 2CF). The low feeding mode (LC) was tested with the feeding rate at an approximate half of co-substrate uptake rate in batch fermentation (Fig.4.7B). Not as expected, 1,3-PDO was obtained at 64.2 g/L within 144 h and not significantly different in comparison to the mild feeding mode (Table 4.4). Whereas glycerol (42.8 g/L) and glucose (41.4 g/L) were still presented at very high concentrations at 144 h (Table 4.4). Further, the high feeding mode (HC) was investigated at the same rate of co-substrate uptake rates (Fig. 4.7C) by feeding from 24 h to 144 h. Certainly, glycerol and glucose were accumulated at the rate of 0.53 and 0.73 g/h, respectively. This indicated that substrates were excessively supplied than those of the consumption rates. However, these results implied that continuous feeding throughout the process may not be appropriated to apply in the fed-batch fermentation.

Additionally, the substrate feeding was divided into 2 periods (2CF); the first period was started from 24 to 56 h and the second period was started from 80 to 112 h with the high continuous feeding rate closed to the glycerol and glucose uptake rates in the optimized batch mode. By using the alternate continuous feeding, the maximum biomass was obtained at the lowest cell dry weight (3.7 g/L) compared to all other modes. But 1,3-PDO was increased with the impressive concentration of 88.6 g/L with productivity of 0.62 g/L/h (Fig.4.7D, Table 4.4).



Figure. 4.8 Fermentation profiles in the informal condition. A. represents substrates

profile and feeding approach. B. represent the metabolites profile and cell growth.

With another attempt, glycerol and glucose were allowed to almost exhausting and then the co-substrate was fed at a single time (Fig.4.8). This may eliminate the problem of the insufficient essential mineral, nitrogen source and glycerol accumulation during fermentation. By feeding with a single time, the microbial cell might utilize substrates and nutrition sufficiently. 1,3-PDO was obtained in the relatively short time with the concentration of 77.6 g/L within 120 h (0.65 g/L/h). However, this mode should be further investigated in depth since compositions of the nutrients and substrates could not be easily determined.

4.4 Conclusions

This study demonstrated the optimization process for the metabolically engineered *E. coli* NSK015 to improve 1,3-PDO production. The aerobic conditions at 300 rpm, 1 vvm with the reduced coenzymeB12 of 7.5 μ M were identified as the best condition and applied in fed-batch fermentation. Under the fed-batch fermentation, *E. coli* NSK015 could produce 1,3-PDO up to 88.6 g/L with the productivity of 0.62 g/L/h.



Table 4.4Comparison of fermentation profiles at different feeding modes in type 2 fermenter.

Conditions	Feeding type	Maximum CDW	1,3- PDO (g/L)	Total added glycerol/glucose (g)	Residual glycerol (g/L)	Residual glucose (g/L)	Consumed glycerol and glucose (g/L)	Yield	Productivity (g/L/h)
1	MC	4.35	64.8	107.9/104.7	18.6	3.8	89.3/102.3	0.88	0.41
2	LC	5.72	64.2	108.3/127.9	42.8	41.4	65.4/86.5	1.19	0.45
3	НС	4.62	55.0	180.3/214.4	98.3	117.3	82.0/97.7	0.82	0.39
4	2CF	3.66	88.6	123.9/144.3	22.7 ยาลยเทค	27.6 S	101.2/116.8	1.0	0.62

CHAPTER V

GENERAL CONCLUSION AND RECOMMENDATIONS



Figure 5.1 Summary of optimized process for 1,3-PDO production by *E. coli* NSK015.

In this dissertation, we reported the production of 1,3-PDO by using the novel metabolically engineered *Escherichia coli* NSK015 in mineral AM1 medium. The efficient production of 1,3-PDO w determined and divided into two main parts including the development of the metabolically engineered *E. coli* NSK015 and the optimization of 1,3-PDO in the fermenter. This study can solve problems on 1,3-PDO production when *E. coli* is employed as a microbial host while overexpression of heterologous genes has been done on genomic DNA using the native host promoter.

In the first part of this study, E. coli C was used as parental strain for engineering to efficiently produce 1,3-PDO. 1,3-PDO pathway containing 2 simple reactions by conversion of glycerol to 3-hydroxypropionaldehyde (3-HPA) and further catalyzation of 3-HPA to 1,3-PDO, respectively, were adopted and designed in E. coli. Genes involving glycerol dehydratase with reactivating factors derived from K. pneumoniae were rearranged to further transcribed in the one direction. The gdrAB*dhaB123* operon was exchanged with the structural *ldhA* gene for its overexpression under *ldhA* promoter of *E. coli*. By this means, *ldhA* was simultaneously removed and substituted with the new heterologous genes. To evaluate the expression of heterologous gene, the clones capable of producing 1,3-PDO were initially verified on the modified aldehyde indicator plates. The clones could change the pale pink color to the dark red color on the modified aldehyde indicator plate. This ability indicated that the heterologous gene could function and the clones were further selected to evaluate 1,3-PDO production in the shaking flask. In addition, the removal of ackA and pflB genes from the clone were also deleted to minimize acetate and formate formation.

To reduce the possibility of the toxic 3-HPA intermediate accumulation, *yqhD* gene encoding 1,3-PDO oxidoreductase was integrated in frame with the *pflB* promoter in the same way of *ldhA*. The *frdABCD* was finally interrupted for removing succinate production. The final strain, *E. coli* NSK015, was constructed and used as a biocatalyst for 1,3-PDO production throughout this study. The yield of 1,3-PDO was achieved at 0.43 mol/mol glycerol when glycerol was used as a sole substrate. In the presence of

glucose, the yield was improved almost two folds to 0.90 mol/mol in the shaking flask. In 2 L fermenter, the metabolically engineered *E. coli* NSK015 was tested for 1,3-PDO production in AM1 medium without supplementing yeast extract at 200 rpm agitation and 1 vvm aeration using 50 g/L of each substrate of glycerol and glucose. 1,3-PDO concentration was achieved at 36.8 g/L in 108 h with the yield of 99% theoretical yield (0.99 mol/mol). When cassava starch was used instead of glucose, 1,3-PDO concentration was reduced to 31.9 g/L with yield of 0.84 mol/mol. All of results indicated that the heterologous genes were capable of overexpressing under native promoters regardless of the use of IPTG for inducing gene expression in which the plasmid instability was neglectable.

The improvement of 1,3-PDO productivity was further accomplished by the conventional optimization process. Vitamin B12, kLa, impeller positions, coenzyme B12 concentration, and agitation speed were verified in batch fermentation to select the best condition, leading to increase the productivity of 1,3-PDO production. As a result, the most suitable condition for applying the fed-batch fermentation was identified using optimized parameters from batch fermentation in which two impellers for agitation were used for agitation at 300 rpm. The reduced concentration of coenzyme B12 of 7.5 μ M was done without affecting 1,3-PDO production. By discovery of this condition, the productivity was improved from 0.34 to 0.75 g/L/h. The glycerol and glucose uptake rates were taken to calculate the feeding rate to employ in fed-batch fermentation.

To observe the possible maximum concentration of 1,3-PDO from *E coli* NSK015, the fed-batch fermentation was carried out with the different modes of cosubstrate feeding. By using the continuously pulsed feeding mode, glycerol and glucose were prepared in the ratio of 1.2:1.5 with the final concentration in the stock of 370 g/L and 505 g/L. Feeding was implemented intermittently for two periods from 24-56 h and 80-112 h, respectively, with the rate of feeding glycerol and glucose at 1.1 and 1.5 g/h, respectively. Under this condition, *E coli* NSK015 could produce 1,3-PDO with impressive concentration up to 88.6 g/L within 144 h. The yield of 1 mol/mol was achieved. However, glycerol and glucose at about 22-28 g/L were still accumulated at the end of fermentation (Fig. 5.1).

This work can solve problems and support other works in research field of 1,3-PDO and even 3-hydroxypropionic acid (3-HP) productions. Because both products are linked together with the first same reaction of the pathway. This work demonstrated the overexpression of heterologous genes that efficiently function under native host promoters while some previously published works showed that expression levels of enzymes involving in 1,3-PDO pathway on inducible plasmids did not efficient enough to drive 1,3-PDO production at high concentration. Consequently, our success can reduce the concern of instability of plasmid and reduce the cost of 1,3-PDO production in the industrial scale. For making 1,3-PDO production better and more interested, the future work can be further done as followings. 1. Since aldehyde-dependent pathway is one of the routes to produce both 1,3-PDO and 3-HP. Therefore, 3-HP can be further produced by removing *yqhD* from this strain and other aldehyde dehydrogenases should be introduced and overexpressed. Based on this study, the fermentation conditions could be applied directly to produce 3-HP.

2. Since glycerol dehydratase genes derived from *K. pneumoniae* depend on coenzyme B12, this seemed to kind of expensive and uneconomical for the industry. In contrast, the glycerol dehydratase derived from *Clostridium* is independent on coenzyme B12. Then coenzyme B12 independent-glycerol dehydratase of *Clostridium dhaB1* and *dhaB2* genes can be substituted *dhaB123* derived from *K. pneumoniae* to test the possibility to produce 1,3-PDO without coenzymB12.

3. Conversion of glucose to glycerol and then finally to 1,3-PDO might be better alternative choices in *E. coli*. Genes involving in conversion of glucose to glycerol may be further introduced into *E. coli* NSK015 and should be overexpressed under one of some native host promoters.

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APPENDIX

1. Determination of k_La

To optimize the production of 1,3-PDO, k_{La} is used as an important parameter to verify how to improve concentration and productivity of 1,3-PDO. k_{La} is thus determined from equation 1. The equation 1 is then integrated, becoming to Eq. 2. A slope of Eq.2 is then used as k_{La} .

 $dC/dt = k_L a (C^* - C_L)$ Eq. 1

To measure kLa, water or AM1 medium without our microorganism was added into fermenter with the same working volume used in this study. DO probe was connected to controller for at least 6 h before taking the data for calculating kLa. After 6 h, DO probe was initially calibrated to prepare for measurement of oxygen dissolved in the medium. Fermenter was flushed with nitrogen and waited until value became zero or stable. DO was set zero. Oxygen or air was then flushed in the fermenter until DO value reached 100 and set span as 100. Nitrogen was finally flushed and set to be zero again after DO value decreased to zero. After DO probe was calibrated. Oxygen was flushed into fermenter. DO value was recorded every second until the value was around 100 or stable. The final DO at the stable point was taken account to C* or saturation concentration of oxygen in Eq 2. For every DO value, they were then taken account to C in Equation 2. Finally, graph was plotted for ABS(ln(1-(C/Cs))) against time in the unit of minute. The slope was calculated as k_{La} . The unit of k_{La} derived from Eq.2 is expressed as the unit per minute.

$$k_{La} = ABS(ln(1-(C/C^*)))$$
 Eq. 2

2. Determination of cell dry weight

In order to evaluate cell dry weight directly from optical density, OD₅₅₀ was plotted against cell dry mass for obtaining the slope value from equation. Twelve mL of individual sample was taken from cultivation to measure OD₅₅₀ and cell dry mass. One mL of sample was divided for centrifugation at 12,500 rpm for 4 min and resuspended with 1 mL DI water before determining optical density by spectrophotometer at wavelength of 550 nm.

For measurement of cell dry mass, two mL of Eppendrof tube was dried at 105°C for 2 days before pre-weighing. Ten mL of sample was then centrifuged for collecting 1 mL cell suspension and aliquoted to the 2 mL pre-weight Eppendrof tube. The cell suspension was centrifuged again to remove water before drying at 105°C for 3 days and weighed. Biomass was finally calculated by determination of the difference between the pre-dried weight and post-dried weight of Eppendrof tube. The different value was

divided by total volume of sample (10 mL). Finally, the value was multiplied by working volume. All data from individual batch of fermentation were rearranged from low to high values and plot between OD and biomass. Equation was derived from this graph and determined biomass directly from OD with $2.68OD_{550} = 1g/L CDW$ (Fig. A.1).



Figure A.1 Graph plotting between OD₅₀₀ against cell dry weight.



BIOGRAPHY

Miss Nonthaporn Wong was born on March 25, 1991, Nauru. She graduated with a Bachelor's degree in Biology and Biotechnology from Nakhon Sawan Rajabhat University in 2014. In 2014, she was selected from Her Royal Highness Princess Maha Chakri Sirindhorn to be Thai Representative to participate in DESY summer student camp, Hamburg, Germany. She received the Royal Golden Jubilee Ph.D. (RGJ-PHD) Program (Grant no. PHD/0125/2556) under supervision of Assoc. Prof. Dr. Kaemwich Jantama. RGJ-PHD program that supports all expenses to study and performs research at School of Biotechnology, Institute of Agricultural Technology, Suranaree University of Technology and Institute National Polytechnique de Toulouse, France.

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