# การโคลนยีน และการวิเคราะห์ลักษณะของโปรตีน MreB และ FtsZ จากเชื้อแบคทีเรียบาซิลลัส ซับทิลิส 



วิทยานิพนธ์่นี้เป็นส่วนหนึ่งของการศึกษาตามหลักสูตรปริญญาวิทยาศาสตรมหาบัณฑิต สาขาวิชาชีวเคมี มหาวิทยาลัยเทคโนโลยีสุรนารี

# GENE CLONING AND CHARACTERIZATION OF MREB AND FTSZ FROM BACILLUS SUBTILIS 



A Thesis Submitted in Partial Fulfillment of the Requirements for the Degree of Master of Science in Chemistry

สุนารี โชคนัด : การโคลนยีน และการวิเคราะห์ลักษณะของโปรตีน MreB และ FtsZ จากเชื้อ แบคทีเรียบาซิลลัส ซับทิลิส (GENE CLONING AND CHARACTERIZATION OF MREB AND FTSZ FROM BACILLUS SUBTILIS). อาจารย์ที่ปรึกษา : อาจารย์ ตร.เศกสิทธิ์ ชำนาญศิลป์, 105 หน้า.

MreB และ FtsZ เป็นโปรตีน โครงสร้างของเซลล์แบคทีเรีย MreB มีความสำคัญสำหรับการ ควบคุมทิศทางในการสร้างผนังเซลล์ ในขณะที่ FtsZ มีความจำเป็นในการแบ่งเซลล์ ไม่นานนี้มี รายงานว่าทั้งสองโปรตีนได้ทำอันตรกิริยากันโดยตรง บริเวณที่มีการสร้างผนังกั้นเซลล์ แนะให้เห็น ว่าโปรตีนทั้งสองทำงานร่วมกันในการแบ่งเซลล์และการสร้างผนังเซลล์ การเกิดเป็นเส้นใยของ MreB และ FtsZ เป็นขั้นตอนแบบพลวัต โดยทั่วไปถูกควบคุมโดยการสลาย ATP และ GTP ตามลำดับ และการเกิดเป็นเส้นใยของ MreB ที่บริเวณผิวภายในเซลล์นั้น ยังต้องการ $\mathrm{Mg}^{2+} \mathrm{MreB}$ ทำ หน้าที่เป็นโครงสร้างค้ำยันเพื่อเชื่อมโยงเอนไซม์ peptidoglycan synthases โดยการทำอันตรกิริยา ผ่านกลุ่มโปรตีนชื่อ penicillin binding proteins FtsZ ไม่ได้มีหน้าที่ในการสร้างผนังเซลล์โดยตรง เช่นเดียวกับ MreB แต่ FtsZ ทำหน้าที่ขับเคลื่อนการสร้างผนังเซลล์ในระหว่างการแบ่งเซลล์ ณ เวลา และสถานที่ที่จำเพาะ ซึ่งอาจสันนิษฐานได้ว่า เป็นผลที่เกิดจากการทำอันตรกิริยากับโปรตีนในกลุ่ม Min ในการกำหนดตำแหน่งของการแบ่งเซลล์ และทำอันตรกิริยากับ MreB ในการกำหนดทิศทาง ในการสร้างแผ่นกั้นเซลล์ (Septum)

การทดลองในวิทยานิพนธ์นี้ ประกอบด้วย การ โคลนยีน mreB-Bs และ ftsZ-Bs การแสดง ออกของโปรตีน การทำโปรตีน MreB-BS และ FtsZ-Bs ให้บริสุทธิ์ และการพิสูจน์เอกลักษณ์ของ คุณสมบัติการสลายนิวคลีโอไทด์ของโปรตีนทั้ง 2 ชนิด และตรวจสอบผลของ apigenin baicalein luteolin $\alpha$-mangostin และ naringenin ซึ่งเป็นสารสกัดจากธรรมชาติ ที่สามารถยังยั้งการแบ่งเซลล์ และรบกวนความสมบูรณ์ของผนังเซลล์ แต่ยังไม่ทราบเป้าหมายระดับโมเลกุล โดยวิธี malachite green assay

จากการทดลองพบว่า MreB สามารถสลาย ATP และ GTP ได้ที่ช่วง $\mathrm{pH} 5.5-8.0$ MreB สลาย ATP ได้ดีที่สุดที่ pH 7.0 และสลาย GTP ได้ดีที่สุดที่ pH 6.5 ส่วน FtsZ สามารถสลายได้ เฉพาะ GTP ในช่วง $\mathrm{pH} 5.5-8.0$ ย่อยสลายดีที่สุดที่ pH 6.5 ผลการศึกษาสารสกัดจากธรรมชาติ

พบว่า apigenin สามารถลดการสลาย GTP ของ FtsZ ได้ร้อยละ 33.3 และ baicalein ร้อยละ 42.5 เมื่อ เปรียบเทียบกับผลการศึกษาของ Mayer และ Amann ในปี ค.ศ. 2009 ผลการทดลองนี้ บ่งชี้ว่าการ สลายนิวคลิโอไทด์และการเกิดเป็นสายของ $\mathrm{MreB}-\mathrm{Bs}$ เป็นกระบวนการที่ไม่ขึ้นตรงต่อกัน และยัง สนับสนุนรายงานข้างต้นที่ว่าการเกิดเป็นสายของ $\mathrm{MreB}-\mathrm{Bs}$ ไม่ต้องการการสลาย นิวคลิโอไทด์ ยิ่ง ไปกว่านั้น ผลการศึกษาครั้งนี้ถือว่าเป็นการรายงานผลครั้งแรกที่แนะว่า FtsZ -Bs เป็นเป้าหมาย ระดับโมเลกุลของ apigenin และ baicalein อย่างไรก็ตาม ยังต้องมีการทำการทดลองเพื่อยืนยัน และ ศึกษาเพิ่มเติมเกี่ยวกับสมบัติทั้งทางด้านชีวเคมี และโครงสร้าง เพื่อให้เข้าใจมากขึ้นถึงกระบวนการ และขั้นตอนการทำงานของโปรตีนทั้งสองชนิด เพื่อจะนำมาซึ่งข้อมูลที่เป็นประโยชน์ในการ ออกแบบยาต่อไป

สาขาวิชาเคมี
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ลายมือชื่อนักศึกษา
ลายมือชื่ออาจารย์ที่ปร็กษา $\qquad$

# SUNAREE CHOKNUD : GENE CLONING AND CHARACTERIZATION OF MREB AND FTSZ FROM BACILLUS SUBTILIS. THESIS ADVISOR : SAKESIT CHUMNARNSILPA, Ph.D. 105 PP. 

## BACTERAL CYTOSKELATON/ MREB/ FTSZ/ ATPASE/ GTPASE/ NATURAL PRODUCT

MreB and FtsZ are key cytoskeletal proteins of bacteria. MreB plays important roles in bacterial cell wall synthesis while FtsZ is crucial for septum formation and cell division. Interestingly, these two proteins make direct interaction and colocalize at the septum suggesting cooperative functions of these proteins in cell wall synthesis during cell division. Polymerization of MreB and FtsZ is a dynamic process regulated by ATP and GTP hydrolysis, respectively. Polymerization of MreB into filaments at the cell periphery beneath the cell membrane responds to the presence of $\mathrm{Mg}^{2+}$, and ATP hydrolysis. MreB acts as a scaffold for tethering of peptidoglycan synthases to the cell membrane by a mechanism that relies on penicillin binding proteins. Similar to MreB, FtsZ has no peptidoglycan synthases activity. However, FtsZ drives peptidoglycan synthesis during cell division at the particular time and place, presumably via interaction with Min-family of proteins and MreB.

This thesis included gene cloning, protein expression, protein purification, and nucleotides hydrolysis characterization of the Bacillus subtilis versions of these proteins, MreB-Bs and FtsZ-Bs. The work found that MreB-Bs has optimum pH for nucleotide hydrolysis at 7.0 , which is different from the optimum pH for the protein polymerization, indicating that these two processes occur independently. This result
supports nucleotide hydrolysis independent polymerization by MreB-Bs, reported by Mayer and Amann, 2009. The work also investigated the effect of apigenin, baicalein, luteolin, $\alpha$-mangostin, and naringenin on ATP and GTP-hydrolysis of MreB and FtsZ by malachite green assay. The results show that MreB was able to hydrolyze both ATP and GTP over a broad pH range ( $5.5-8$ ), with the optimum pH for ATP hydrolysis and GTP hydrolysis of 7.0 and 6.5 , respectively. On the other hand, FtsZ was able to hydrolyze only GTP over a broad pH range (5.5-8), with optimum pH at 6.5. Importantly, this work also found that apigenin and baicalein were able to inhibit the GTPase activity of FtsZ-Bs by $33.3 \%$ and $42.5 \%$, respectively. These results suggest that FtsZ-Bs is a molecular target of apigenin and baicalein in cell wall deformation. In order to get clearer ideas of these two proteins for drugs design, further biochemical and structural studies of these two proteins need to be done.
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## GENE CLONING AND CHARACTERIZATION OF MREB

## AND FTSZ FROM BACILLUS SUBTILIS

Suranaree University of Technology has approved this thesis submitted in partial Fulfillment of the requirements for the Master's Degree

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## AKNOWLEDEMENTS

First and foremost, I would like to thank my thesis advisor, Dr. Sakesit Chumnarnsilpa for the opportunity to conduct my thesis work under his guidance. He has been teaching me the theories and techniques that are involved in all biochemistry, always available for scientific discussion and has emphasized the importance of thinking independently. He provided a pathway for me to follow which allowed me to successfully complete my thesis program.

I expressed my sincere gratitude to Prof. Dr. James R. Ketudat-Cairns for providing me the opportunity to study towards my master degree in Biochemistry. He has away been supported during my time as a graduate student at SUT.

I sincerely thank Assoc. Prof. Dr. Jatuporn Wittayakun, and Asst. Prof. Dr. Panida Khunkaewla for patiently reading this dissertation and providing helpful comments.

Special thanks are extended to all my friends in the School of Biochemistry, Suranaree University of Technology for their help.

I would like to express my deepest gratitude to my family. They were always supporting me, encouraging me and cheering me up with their best wishes.

Finally, I would like to thank School of Biochemistry, Center for Protein Structure, Function and Application, Institute of Science, and Institute of Research and development, Suranaree University of Technology.

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## LIST OF ABBREVIATIONS AND SYMBOLS

ADP ATP

ATPase

## bp

BSA

DNA
DNase
dNTP
DTT
EDTA
EtBr

F

G
GDP
GTP
GTPase
HEPES
Hsp70
IF
IMAC

S-(3,4-Dichlorobenzyl)isothiourea hydrochloride
Adenosine diphosphate
Adenosine triphosphate
Adenosine triphosphate hydrolase
Base pair
Bovine serum albumin

Deoxyribonucleic acid
Deoxyribonuclease
Deoxy nucleotide triphosphates
Dithiothreitol
Ethylenediaminetetraacetate
Ethidium bromide

Filament
Globular
Guanosine diphosphate
Guanosine triphosphate
Guanosine diphosphate hydrolase
Hydroxyethyl piperazineethanesulfonic acid
70 kilodalton heat shock protein
Intermediate filament
Immobilized metal affinity chromatography

## LIST OF ABBREVIATIONS AND SYMBOLS (Continued)

| IPTG | Isopropyl thio- $\beta$-D-galactoside |
| :---: | :---: |
| kDa | Kilo Dalton |
| MES | 2-(N morpholino)-ethanesulfonic acid |
| MF | Microfilament |
| MT | Microtubule |
| MRSA | Methicillin-resistant Staphylococcus aureus |
| MWCO | Molecular weight cut off |
| OD | Optical density |
| PCR | Polymerase chain reaction |
| PBP | Penicillin-binding protein |
| rpm | Revolutions per minute |
| SDS | Sodium dodecyl sulfate |
| SDS-PAGE | Polyacrylamide gel electrophoresis with SDS |
| TEMED | Tetramethylenediamine |
| Tris | Tris-(hydroxymethyl)-aminoethane |
| VRE | Vancomycin resistant Enterococci |
| VRSA | vancomycin-resistant Staphylococcus aureus |
| X-gal | 5-bromo-4-chloro-3-indolyl- $\beta$-D-galactopyranoside |

## CHAPTER I

## INTRODUCTION

### 1.1 General introduction

Cytoskeletal proteins play crucial roles in cellular organization. It was believed those cytoskeletal proteins do not exist in bacteria. Bacterial cell morphology was traditionally assumed to be determined by the external cell wall (Carballido-López, 2006a). It is now known that bacterial cytoskeletons, MreB, FtsZ, are crucial for bacterial survival.

Similar to actin, MreB (an actin orthologe) can assemble into filaments. Assembly of MreB into filaments or polymerization is a dynamics process, which relates to ATP hydrolysis (Mayer and Amann, 2009; Bean and Amann, 2008). The polymerization of MreB is involve in many cellular processes especially in cell wall synthesis (Carballido-López and Errington, 2003), which is crucial for cell elongation and cell division (Fenton and Gerdes, 2013; van den Ent, Amos, and Löwe, 2010).

FtsZ is a tubulin orthologe, which forms a ring-like structure at the mid cell, called the Z-ring. Formation of the Z-ring through polymerization requires GTP hydrolysis. GDP from GTP hydrolysis is used to support the polymer (Löwe, van den Ent, and Amos, 2004). The polymerization of FtsZ is the rate limiting step in septum formation under the membrane at the site of cell division (Erickson, Anderson, and Osawa, 2010). Z-ring is necessary for the localization of a variety of other proteins (Aarsman, Piette, Fraipont et al., 2005).

Both FtsZ and MreB tether peptidoglycan synthases by a mechanism that relies on penicillin binding protein 2, PBP2 (Varma and Yong, 2009). A recent report has shown that direct interactions between FtsZ and MreB are necessary for appropriate cell division in Escherichia coli, which suggests a potential mechanism for the coordination of cell elongation and cell division (Fenton and Gerdes, 2013). In the absence of MreB, FtsZ can direct peptidoglycan incorporation into the lateral walls of E. coli, which may indicate a more general role in coordination of the peptidoglycan synthases. FtsZ may drive a general form of peptidoglycan synthesis during cell division at a particular time and place. The genesis of cell shape is probably based on a specific geometry of cell wall growth that is directed by MreB (Varma and Yong, 2009).

Bacteria are unicellular microbes. Cell morphology influences bacterial existence. A small cell size increase surface area relative to high surface-to-volume ratio (Capaldo-Kimball and Barbour, 1971). This leads to obvious benefits such as selective pressures-access to nutrients, cell division, predation, and motility (Yang, Blair, and Salama, 2016; Jiang and Sun, 2010). Bacteria usually attach to surfaces through specific cell morphology, which is essential for the overall survival of the bacteria, especially pathogenic ones (Okagaki, Strain, Nielsen et al., 2010).

Together, the preceding information shows that polymerization and direct interactions of MreB and FtsZ are very important for survival of bacteria. The molecular detail and mechanism remain unclear, and understanding of these processes is necessary and will provide fundamental information for controlling bacterial infection and drug design. This study focuses on an investigation of effects of natural products on nucleotide hydrolysis by MreB and FtsZ.

### 1.2 Research objectives

1.2.1 To clone and express mreB and ftsZ of Bacillus subtilis in E. coli.
1.2.2 To purify MreB and FtsZ of B. subtilis for biochemical studies.
1.2.3 To study the effects of some natural products on nucleotides hydrolysis by MreB and FtsZ of B. subtilis.

## CHAPTER II

## BACKGROUND

### 2.1 Eukaryotic cytoskeletons

In eukaryotes, the cytoskeletal proteins play crucial roles in the organization of the cells. The three classes of eukaryotic cytoskeletal fibers are microfilaments, microtubules and intermediate filaments (as show in Figure 2.1). Microfilament, actin filament (F-actin) is a polymer of globular actin or G-actin (Moriyama and Yahara, 2002). Microtubule is a polymer of $\alpha, \beta$-tubulin heterodimers, while $\gamma$-tubulin is not part of the tubulin subunit (Desai and Mitchison, 1997). Lastly, intermediate filaments are homopolymers formed by several classes of cell-specific subunit proteins, including keratins, lamins, and vimentin (Herrmann and Aebi, 2004). A complex filamentous assemblage of microfilaments, microtubules and intermediate filaments form a highly dynamic network that controls a multitude of cell processes, such as cell motility, cell morphology, cell division, cell adhesion.

### 2.1.1 Microfilament (MF)

Actin, the monomeric form of MF, is one of the most abundant and highly conserved proteins among eukaryotic cells (Pantaloni, Hill, Carlier et al., 1985). It is a member of a larger superfamily of proteins, which include a group of ATPases such as Hsp70 (Bork, Sander, Valencia et al., 1992).
(A) Microfilaments

(B) Microtubules
(D) $A+B+C$ merge
(C) Intermediate filaments


Figure 2.1 Fluorescence images representation of eukaryotic cytoskeleton. The three major components of the cytoskeleton of Huh7 cultured human hepatocytes were triple-stained with phalloidin to visualize microfilaments (blue, A), anti-tubulin antibody to visualize microtubules (green, B), and anti-K8/K18 antibody to visualize intermediate filaments (red, C). A superimposed image is shown in (D). Scale bar represent $10 \mu \mathrm{~m}$ (Omary, Ku, Tao et al., 2006).

In vivo, switching between G-actin and F-actin is a dynamic process, actin dynamics. This process is essential for various cellular processes, including cell motility, cell migration, phagocytosis, vesicular movement, cytokinesis, and
molecular transport (Hoglund, Karlsson, Arro et al., 1980). Actin dynamics is regulated by innate properties of actin itself and by actin-binding proteins, ABS (Schüler, 2001).


Figure 2.2 Representation of the conformational change of G-actin upon polymerization. Superimposition the structure of ADP-actin shown in yellow (PDB: 1J6Z; Otterbein, Graceffa, and Dominguez, 2001) onto the structure of ATP-actin excised from a fiber diffraction derived model shown in cyan (Oda, Iwasa, Aihara et al., 2009) demonstrates that subdomains III and IV are rotated with respect to subdomain I and II about the rotation axis (red line) in the direction indicate by the red arrow. The two conformations are related by a $20^{\circ}$ rotation of the major domains around an axis passing along the front of subdomain I and the side of subdomain III, and the DNase I-binding, which loop extends to make contacts to an adjacent actin molecule in the filament, is in different position. The rotation is associated with bends of the peptide chain, as indicated in blue.

Superimposition of the crystal structure of G-actin with ADP in nucleotide binding pocket (Otterbein et al., 2001) onto actin monomer (ATP-actin) excised from fiber diffraction of actin filament (Oda et al., 2009) has demonstrated that the conformation of actin in monomeric state is different from the polymeric state (Figure 2.2). It suggests that the conformation of actin affects the polymerization process of actin.

The process of actin polymerization is shown in Figure 2.3. Polymerization of G-actin to form F-actin involves uptake of a divalent salt cation $\left(\mathrm{Ca}^{2+}\right.$ or $\left.\mathrm{Mg}^{2+}\right)$ that ATP presented (Blanchoin and Pollard, 2002). G-actins carrying ATP will form a stable nucleus, composing of 2-3 G-actin in the process. During the process of elongation, actin depolymerizing factor (cofilin family) binds to the side of ADP-actin filaments and induces pointed end depolymerization to increase the concentration of G-actin at steady state (Moriyama and Yahara, 2002). In the next step, profilin enhances the exchange of ADP for ATP to recycle G-actins (Pantaloni et al., 1985; Nürnberg, Kitzing, Grosse et al., 2011). The profilin-actin complex assembles exclusively at the barbed end by blocking the barbed ends of major F-actin to increase G-actin at steady state and funnel the flux of G-actins to the non-capped filaments, which take another G-actin in order to start a new cycle of polymerization (Pollard, Blanchoin, Mullins et al., 2000).


Figure 2.3 A schematic representation actin polymerization and roles of actin binding proteins: (Adapted from Nürnberg et al., 2011).

### 2.1.2 Microtubules (MT)

Microtubules are long filamentous, tube-shaped protein polymers involve in many essential cellular processes, including cell division, ciliary and flagella motility, intracellular transport, and development and maintenance of cell shape (Jordan and Wilson, 2004).

Microtubules are non-covalent polymers of the two related protein monomers $\alpha$ - and $\beta$-tubulins, in the presence of GTP hydrolysis (Hyman, Salser, Drechsel et al., 1992; Mandelkow and Mandelkow, 1990). Figure 2.4 shows that polymerization of tubulin occurs by a nucleation-elongation mechanism in which the relatively slow formation of a short microtubule or nucleus is followed by rapid elongation of the microtubule at its ends by the reversible addition of tubulin dimers (Ottaviani, Pregnolato, Cangiotti et al., 2012).


Figure 2.4 Polymerization of microtubules. Heterdimers of $\alpha$ - and $\beta$-tubulin assemble to form a short microtubule nucleus. Nucleation is followed by elongation at both ends to form a cylinder (plus (+) end with $\beta$-tubulin facing, and $\alpha$-tubulin facing the minus end (-)) (Adapted from Jordan and Wilson, 2004).

The tubulins show complex polymerization dynamics that use energy provided by GTP, which is in a direct contact with loops T1 to T6 of the GTPase domain (Desai and Mitchison, 1997). Figure 2.5 shows the structure of tubulin in complex with Taxol (Nogales, Whittaker, Milligan et al., 1999). The GTPase domains are showing in red and the activation domains in blue. The core helix that connects the two globular domains in each monomer is yellow and the C -terminal domain on the external surface is green.


Figure 2.5 The model of the tubulin dimer show $\alpha$-tubulin with GTP bound and $\beta$ tubulin containing GDP and Taxol bound. The arrow indicates the direction of the protofilament and microtubule axis (Amos and Schlieper, 2005).

The GTP is sandwiched between the $\alpha$ - and $\beta$-tubulin subunits of each heterodimer, being bound to $\alpha$-tubulin by loops T1-T6 and also makes contact with loop T7 of $\beta$-tubulin. The nucleotide bound to $\beta$-tubulin has been hydrolyzed to GDP through contact with helix H 8 and loop T 7 of the activation domain of another $\alpha$ tubulin subunit. Taxol sits in the pocket of $\beta$-tubulin on the inside face of microtubules. In $\alpha$-tubulin, this pocket is occupied by the extended L loop (Amos and Schlieper, 2005; Amos, 2004).

### 2.1.1 Intermediate filament (IF)

IFs typically form a network throughout the cytoplasm, surrounding the nucleus and extending out to the cell periphery (Cooper, 2000). They are often anchored to the plasma membrane at cell-cell junctions such as desmosomes, where the external face of the membrane is connected to that of another cell (Osborn and Weber, 1983).

The family of IF proteins include two common traits that define members. Firstly, the (IF) proteins exhibit a characteristic tripartite domain organization. It consists of a highly conserved $\alpha$-helical central rod domain (subdomains 1A, 1B, 2A and 2B) flanked by variable N -terminal head and C-terminal tail domains (Eriksson, Dechat, Grin et al., 2009), a generic version of which is shown in Figure 2.6. Secondly, the proteins can self-assemble into cytoskeletal filaments, which usually appear as homogeneous, apolar fibers that have a $10-12 \mathrm{~nm}$ diameter (Coulombe, Ma, Yamada et al., 2001).


Figure 2.6 Genetic secondary structures of intermediate filament proteins (Coulombe et al., 2001).

IFs are composed of a variety of proteins that are expressed in different types of cells (Cooper, 2000). The more than 65 different IF proteins are divided into six chemically distinct classes, which are described show in Table 2.1 (Cooper, 2000). At the protein level, polymerization properties define 3 assembly groups (A, B, and C), shown in Figure 2.7. Assembly group A, Keratins assemble from heterodimeric tetramers by lateral and nearly concomitant longitudinal assembly into heterogenous full-width filaments. Assembly group B, Vimentin-type assembly starts from antiparallel, half-staggered double dimers (or tetramers) to form full-width, unitlength filaments. Assembly group C, Lamin dimers associate first into head-to-tail filaments that later laterally associate. The orientation of the two associating filaments is arbitrary (Herrmann and Aebi, 2004; Kim and Coulombe, 2007).


Figure 2.7 Schematic models of the prime association reactions occurring between dimers and double dimers, respectively, of the three major IF assembly groups (Herrmann and Aebi, 2004).

Table 2.1 Classification of intermediate filament proteins.

| Type | Protein | $\begin{aligned} & \text { Size } \\ & (k D) \end{aligned}$ | Assembly group | Site of expression |
| :---: | :---: | :---: | :---: | :---: |
| I | Acidic keratins <br> (~15 protein) | 40-60 | A | Epithelial cells |
| II | Neutral or basic keratins ( $\sim 15$ protein) | 50-70 | A | Epithelial cells |
| III | Vimentin | $54$ | B | Fibroblasts, white blood cells, and other |
|  | Desmin | 53 | B | cell types |
|  | Glial fibrillary acidic protein |  | B | Muscle cells |
|  | Peripherin |  | B | Glial cells <br> Peripheral neurons |
| IV | Neurofilament protein-L <br> Neurofilament protein-M | $67$ $1150$ | B | Neurons <br> Neurons |
|  | Neurofilament protein-H | 200 | B | Neurons |
|  | $\alpha$-Internexin | 66 | B | Neurons |
| V | Nuclear lamins | 60-75 | C | Nuclear lamina of all cell types |
| VI | Nestin | 200 | B | Stem cells of central nervous system |

### 2.2 Prokaryotic cytoskeleton proteins

In recent years, scientists have found that bacteria present a number of cytoskeletal structures. The most understood bacterial cytoskeleton proteins, FtsZ, MreB and crescentin are orthologe of the three major types of eukaryotic cytoskeletal proteins actin, tubulin, and intermediate filament, respectively. These proteins play essential roles in dictating cell shape, motility, chromosome separation and cell division.

### 2.2.1 MreB

In rod shape bacteria, the cytoskeletal protein MreB is an actin orthologue that plays important roles in several cellular functions in bacteria, especially regulation of cell shape (Bean, Flickinger, Westler et al., 2009; Fenton and Gerdes, 2013). The mre $B$ gene is located in the gene cluster mre (murein cluster e). Bioinformatics analysis indicated that MreB has a sequence pattern in common with the actin superfamily (van den Ent et al., 2001). The crystal structure of Thermotoga maritima MreB1 revealed that actin is the most related protein to MreB, in overall structure, as shown in Figure 2.8 (van den Ent et al., 2001).

MreB plays an important role in regulation of cell shape (van den Ent et al., 2010). Mutation of $m r e B$ causes $E$. coli to lose in normal rod-shape to become a spherical shape (Doi et al., 1988; Yamachika et al., 2012). In E. coli, MreB polymerizes into filament bundles in a reversible process, which responds to ions $\left(\mathrm{Mg}^{2+}\right)$ and nucleotides (ATP) (van den Ent et al., 2001). This suggests that polymerization proceeds with a nucleation step in much the same way as that of eukaryotic actin (Nurse and Marians, 2012).


Figure 2.8 Superimposition of the crystal structure of MreB (PDB: 1JCE) in blue onto actin (PDB: 3HBT) in grey. Both structures include 4 subdomains and ATP binds in a cleft between the domains.

The early study by Bean and colleagues (Bean et al., 2009) has shown that S-(3,4-Dichlorobenzyl) isothiourea (A22) increases the critical concentration for ATP-bound MreB assembly from 500 nM to approximately 2000 nM . Suggesting that A22 acts as a competitive inhibitor of ATP binding to MreB, and MreB is unable polymerize when bound to A22. The structure of MreB of Caulobacter crescentusin complex with A22 shows that A22 binds closely to the nucleotide in MreB, presumably preventing nucleotide hydrolysis and destabilizing double protofilaments (van den Ent et al., 2014). In vivo study has shown that A22 inhibits growth and induces a morphological change of $P$. aeruginosa, as shown in Figure 2.9 (Cowles and Gitai, 2010; Yamachika et al., 2012).


Figure 2.9 Effects of A22 on the morphology of P. aeruginosa. A22 absent (A) A22 present (B). Scale bars represent $2 \mu \mathrm{~m}$ (Cowles and Gitai, 2010).

### 2.2.2 FtsZ

The tubulin orthologue, Filamenting temperature-sensitive mutant $\underline{Z}$ (FtsZ) is one of the major cytoskeletal protein present in eubacteria, and is also found in archae and chloroplasts (Erickson, Anderson, Osawa et al., 2010; Löwe, van den Ent, Amos et al., 2004). FtsZ forms a dynamic ring-like structure, called the Z-ring at mid cell, under the membrane, at the site of cell division. It is necessary for the localization of various proteins that are required for cell division (Erickson et al., 2010).

FtsZ forms polymers, with the GTPase active-site split across two monomers. One monomer provides the GTP-binding site and the other, through its T7 loop, nucleotide hydrolysis, as illustrated in Figure 2.10 (Erickson and Osawa, 2010). The polymers contain a substantial amount of GTP, which suggests that hydrolysis occurs with some lag following assembly. After the polymerization, the polymer condenses to form a Z-ring. Some models suggest that rearrangement of FtsZ drives
condensation of the ring and generates adequate force to pinch the cell at the site of the division plane (Lan et al., 2009).


Figure 2.10 The structure of FtsZ protofilament. FtsZ forms a chain of molecules with the GDP bound between two subunits. Residues in loop T7 contact the nucleotide directly and are required for hydrolysis (Löwe et al., 2004).

The divisome of $E$. coli is nucleated initially by the assembly of FtsZ, and then the rest of the ring proteins are incorporated sequentially: FtsA, ZipA, FtsK, FtsQBL, FtsW, FtsEX, and finally FtsN. ZipA and FtsA interact with and stabilize the Z ring at the inner membrane, as showed in Figure 2.11 (Aarsman, Piette, Fraipont et al., 2005). The divisome contains essential cell division proteins, such as the peptidoglycan synthases PBP1B and PBP3, and amidase enzymes (Ami) with their activators (EnvC), as well as proteins of the Tol-Pal complex for constriction of the outer membrane. Activity of the PBPs is regulated in part by outer membrane-
anchored lipoproteins (LpoB) and lytic transglycosylase (LT). Cells lacking functional FtsZ are unable to divide and instead grow as filaments (Typas, Banzhaf, Gross et al., 2011).


Figure 2.11 The schematic representation of a transverse section of a divisome of $E$. coli cell. The complex is composed of FtsA, ZipA, FtsK, FtsQBL, FtsW, FtsEX, FtsN, PBP1B, PBP3, Ami, EnvC, Tol-Pal, LpoB, and LT (Typas et al., 2011).

### 2.2.3 Crescentin

Crescentin (CreS) is the only IF orthologue identified in prokaryotic cells at the moment (Shih and Rothfield, 2006), based on structural prediction and in vitro polymerization properties (Charbon, Cabeen, Jacobs-Wagner et al., 2009). The amino acid sequence of CreS has a distinct seven-residue repeat that is predicted to
form coiled-coil structures (Herrmann and Aebi, 2004). Because of the dominating coiled-coil repeat, sequence comparisons are unreliable, but CreS shares some important overall features with eukaryotic IF (Herrmann and Aebi, 2004; Margolin, 2004). Analysis has revealed that the domain organization of CreS is similar to eukaryotic IF, suggesting that CreS probably is a prokaryotic homologue of IF (Michie and Löwe, 2006).

CreS forms a polymer along the cell-length direction, localizes at the inner curvature of the cell, and attaches to the cytoplasmic side of the cell membrane (Gitai, 2005). CreS filament, when detached from the cell membrane through antibiotic treatment, collapses into a helix (Cabeen et al., 2009). This suggests CreS affixed to the cell membrane in a stretched configuration form in untreated cells, and may generate a constricting force on the cell wall, as shown in the model action of CreS shown in Figure 2.12.

CreS forms filaments to generate the curved cell morphology in commashaped Caulobacter crescentus that were identified in a screen for C. crescentus transposon insertion mutations that affected cell shape (Briegel, Dias Jensen et al., 2006). Loss of the structural gene for CreS, leads to a change in cell shape from comma to rod (Ausmees and Wagner, 2003). Previous report indicated bacterial cells lacking CreS lose their curved morphology and adopt a straight rod-like shape, the mechanism of CreS induced curvature is likely to be simply mechanical. The difference of cell morphology is shown in Figure 2.13 (Charbon et al., 2009).


Figure 2.12 Model for CreS action. A rod-shaped cell lacking CreS, turgor pressure strains peptide bridges (A). A cell with a CreS structure, which is affixed to the cell membrane in a stretched configuration (B). This in turn produces a compressive force on the cell wall. Cell elongation under these conditions produces a gradient of cell lengths, from line a to line c (Cabeen et al., 2009).


Figure 2.13 Composite phase-contrasts image of E. coli cell showed that CreS can produce cell curvature; with CreS indicate elongated and helical cells while without CreS indicate rod shape of $E$. coli (Cabeen et al., 2009).

### 2.3 MreB and FtsZ interactions

The process of cell elongation is controlled by MreB, which localize components of the peptidoglycan synthesis along the lateral cell wall, thereby governing the geometry during cell wall growth (Carballido-López, 2006b). During cell division on the inner membrane at the middle of the cell, it triggers invigilation by attracting a set of proteins to form a septal Z ring was formed with the tubulin homologue FtsZ (Gaballah, Kloeckner, Otten et al., 2011).

Recent studies by Fenton and Gerdes suggest that direct interaction of MreB and FtsZ is crucial for septum synthesis during cell division. In vivo study by bacterial two hybrid analysis suggests MreB-FtsZ crosslink. Mutagenesis showed that D258 of MreB interacts with the C-terminus of FtsZ. The MreB/FtsZ interaction localizes the Z-ring and supports septum formation (Fenton and Gerdes, 2013).

### 2.4 Natural products

A natural product is a chemical compound produced by a living organism and originating in nature. Natural products can also be produced by total synthesis or semi-synthesis processes. The products as long as a key role can be played in traditional medicine or other complementary and integrative health practices. The ideas are effect of small molecule that mediated by specific interactions of the drug molecule with biological macromolecules.

### 2.4.1 Apigenin

Apigenin (4',5,7-trihydroxyflavone), is a dietary flavonoid which is found in a wide variety of plants and herbs, including parsley, celery, rosemary, oregano, thyme, basil, and coriander (Figure 2.14). It is a natural product belonging to the flavone class. The chemical structure of apigenin is shown in Figure 2.15. Apigenin shows promising biological effects, such as prevention and therapy of prostate cancer, suppression of tumorigenesis and angiogenesis in melanoma (Caltagirone, Rossi, Poggi et al., 2000) and breast, skin, and colon carcinomas (Wang, Heideman, Chung et al., 2000).

Apigenin also has antimicrobial activity against oral pathogen agents (Cha, Kim, Cha et al., 2016). Ceftazidime is bactericidal in action, exerting its effect on target cell wall proteins and causing inhibition of cell wall synthesis. Recent report by Eumkeb (2012) has shown that the cells were treated with ceftazidime presenting elongated of shape and damaged cell wall especially at the cells polar. On the other hand, apigenin also causes cell elongation without disrupting cell wall integrity at the
poles of the cells as ceftazidime, as show in Figure 2.16 (Eumkeb and Chukrathok, 2013).


Coriander
(Coriandrum sativum)


Oregano (Origanum vulgare)


Basil (Ocimum basilicum)


Parsley (Petroselinum crispum)

Thyme (Thymus vulgaris)

Rosemary
(Rosmarinus officinalis)

Celery (Apium graveolens)

Figure 2.14 Examples of plants and herbs containing apigenin and luteolin.



Figure 2.15 Chemical structure of apigenin.


Figure 2.16 Ultrathin sections of log phase CREC grown in cation-adjusted Mueller Hinton broth containing: drug-free (A), $20 \mu \mathrm{~g} / \mathrm{ml}$ ceftazidime (B), $10 \mu \mathrm{~g} / \mathrm{ml}$ apigenin (C), $3 \mu \mathrm{~g} / \mathrm{ml}$ ceftazidime and $3 \mu \mathrm{~g} / \mathrm{ml}$ apigenin (D). Scale bars represent $0.5 \mu \mathrm{~m}$ (Eumkeb and Chukrathok, 2013).

### 2.4.2 Baicalein

Baicalein (5,6,7-trihydroxyflavone) is a type of flavonoid, a major flavone of baikal skullcap (Scutellariae baicalensis, Figure 2.17). The chemical structure of baicalein show as Figure 2.18. The ability of baicalein showed significant cytotoxicity against the hepatocellular carcinoma cells and moderate cytotoxicity against immortalized human hepatocytes (Zheng, Yin, Grahn et al., 2014), and act as an anti-inflammatory agent (Hsieh, Hall, Ha et al., 2007). The baicalein has a potential adjuvant role in clinical bactericidal therapy for severe enterococcal
infection that demonstrated baicalein and gentamicin can act synergistically in inhibiting vancomycin-resistant Enterococcus (Chang, Li, Tang et al., 2007). It also demonstrated strong antibacterial activity against clinically isolated methicillin and vancomycin-resistant Staphylococcus aureus that baicalein could be employed as a natural antibacterial agent against multidrug-resistant pathogens infection (Lee, Jung, Cha et al., 2015).


Figure 2.17 Appearance of baikal skullcap. It is member of the mint family (Lamiaceae) that is a perennial found in sandy mountain soils in northeast China and adjacent Russia, Korea, Mongolia, and Japan (Foster, 2004).


Figure 2.18 Chemical structure of baicalein.

Baicalein is a potential synergistic adjunct to ceftazidime for the treatment of S. pyogenes infections. It acts by exerting inhibition activity against $\beta$ lactamase. Compared with the controls (drug-free), combining ceftazidime with baicalein caused peptidoglycan and morphological damage (Figure 2.19) (Siriwong, Pimchan, Naknarong et al., 2015).


Figure 2.19 Ultrathin sections of $\log$ phase S. pyogenes DMST 30653 grown in cation-adjusted Mueller-Hinton broth with lysed horse blood ( $2.5 \% \mathrm{v} / \mathrm{v}$ ) containing: Drug-free (A), $0.25 \mu \mathrm{~g} / \mathrm{ml}$ ceftazidime (B), $128 \mu \mathrm{~g} / \mathrm{ml}$ baicalein (C), $0.09 \mu \mathrm{~g} / \mathrm{ml}$ ceftazidime and $24 \mu \mathrm{~g} / \mathrm{ml}$ baicalein. Scale bars represent $0.5 \mu \mathrm{~m}$ (Siriwong et al., 2015).

### 2.4.3 Luteolin

Luteolin, $3^{\prime}, 4$ ',5,7-tetrahydroxyflavone (Figure 2.20), is a common flavonoid that exists in many types of plants, including fruits, vegetables, and
medicinal herbs, such as celery, parsley, thyme, oregano, and rosemary (Figure 2.14), broccoli, green pepper, navel oranges, and olive oil. The hydroxylated flavone derivative is a strong antioxidant and radical scavenger. (Evans, Miller, Paganga et al., 1996). Luteolin activates both the extrinsic and intrinsic apoptosis pathways (Horinaka, Yoshida, Shiraishi et al., 2005). Luteolin is potent to inhibit angiogenesis (Bagli, Stefaniotou, Morbidelli et al., 2004), to prevent carcinogenesis, to reduce tumor growth (Fang, Zhou, Shi et al., 2007) and to sensitize tumor cells to the cytotoxic effects of some anticancer drugs, which suggests that this flavonoid has cancer chemopreventive and chemotherapeutic potential (López-Lázaro, 2009). It also plays a role in inhibition of fatty acid synthase activity (Coleman, Bigelow, Cardelli et al., 2009).


Figure 2.20 Chemical structure of luteolin.

### 2.4.4 $\alpha$-Mangostin

Alpha-mangostin is a natural xanthonoid that belongs to the family of xanthones (Figure 2.21). It is a pigment from mangosteen (Figure 2.22). The xanthone derivative has been shown to induce apoptosis via inhibiting fatty acid synthase (Quan, Wang, Ma et al., 2012). $\alpha$-Mangostin was found to be active against
vancomycin-resistant enterococci (VRE) and methicillin-resistant Staphylococcus aureus (MRSA) with synergism between alpha-mangostin and gentamicin (GM) against VRE, and $\alpha$-mangostin and vancomycin hydrochloride (VCM) against MRSA (Sakagami, Iinuma, Piyasena et al., 2005). The direct interactions of $\alpha$-mangostin with the $S$. aureus membrane are responsible for the rapid concentration-dependent membrane disruption and bactericidal action (Koh, Qiu, Zhu et al., 2013).


Figure 2.21 Chemical structure of $\alpha$-mangostin.

Figure 2.22 Appearance of mangosteen fruit (Garcinia mangostana). It is found particularly in the South-East Asian regions.

### 2.4.5 Naringenin

Naringenin is a flavanone, a type of flavonoid (Figure 2.23). It can be found in grapefruits, oranges and tomato skin. The pharmacological properties of naringenin, which has many potential applications, include an antioxidant effect (Andrade, Carvalho, Cunico et al., 2010), hepatoprotective (Lee, Yoon and Moon., 2004), anti-inflammatory (Bodet, La, Epifano et al., 2008), antiviral (Nahmias, Goldwasser, Casali et al., 2008), antihypertensive (Saponara, Testai, Lozzi et al., 2006) and antimutagenic effects (Renugadevi and Prabu, 2009). In addition naringenin displayed additive effects when combined with the different antibiotics at subinhibitory concentrations against multidrug resistant Staphylococcus aureus (Ng'uni, Mothlalamme, Daniels et al., 2015) and inhibited the growth of gram-positive and gram-negative bacteria (Andrade et al., 2010).


Figure 2.23 Chemical structure of naringenin.

## CHAPTER III

## MATERIALS AND METHODS

### 3.1 Materials

### 3.1.1 Chemicals

Chemical reagents and sources used in this work are listed in Table 3.1

Table 3.1 Chemical reagents and sources.

| Reagent | Source |
| :--- | :--- | :--- |
| - Coomassie brilliant blue R250 | Acros Organics |
| - Dithiothreitol $\left(\mathrm{C}_{4} \mathrm{H}_{10} \mathrm{O}_{2} \mathrm{~S}_{2}\right)$ |  |
| - Ethidium bromide $\left(\mathrm{C}_{21} \mathrm{H}_{20} \mathrm{BrN}_{3}\right)$ |  |
| - Malachite green $\left(\mathrm{C}_{23} \mathrm{H}_{25} \mathrm{ClN}_{2}\right)$ |  |
| - Methanol $\left(\mathrm{CH}_{3} \mathrm{HO}\right)$ |  |
| - Perchloric acid $\left(\mathrm{HClO}_{4}\right)$ |  |

- Triton X-100
- Ammonium persulfate $\left(\mathrm{NH}_{4}\right)_{2} \mathrm{~S}_{2} \mathrm{O}_{4}$
- Glycerol $\left(\mathrm{C}_{3} \mathrm{H}_{8} \mathrm{O}_{3}\right)$
- Sodium chloride ( NaCl )
- Magnesium chloride $\left(\mathrm{MgCl}_{2}\right)$
- Sodium dodecyl sulfate $\left(\mathrm{NaC}_{12} \mathrm{H}_{25} \mathrm{SO}_{4}\right)$

Table 3.1 Chemical reagents and sources (Continued).

| Reagent | Source |
| :--- | :--- |
| - Potassium chloride(KCl) | CARLO ERBA |
| - Tris(hydroxymethyl)aminomethane $\left(\mathrm{NH}_{2} \mathrm{C}\left(\mathrm{CH}_{2} \mathrm{OH}\right)_{3}\right)$ |  |
| - Hydrochloric acid $(\mathrm{HCl})$ |  |
| - Ethylenediaminetetraacetic acid $\left(\mathrm{C}_{10} \mathrm{H}_{16} \mathrm{~N}_{2} \mathrm{O}_{8}\right)$ |  |
| - Sodium molybdate $\left(\mathrm{Na}_{2} \mathrm{MoO}_{4}\right)$ |  |
| - Sodium acetate $\left(\mathrm{CH}_{3} \mathrm{COONa}^{2}\right)$ | HiMedia |
| - Bacto-agar |  |
| - Peptone |  |
| - Yeast extract |  |
| - Adenosine triphosphate |  |
| - Guanosine triphosphate |  |
| - Ampicillin |  |

- Bovine serum albumin
- Isopropyl- $\beta$-D thiogalactopyranosid
- Lysozyme
- N , N-Methylenebisacrylamide
- Acrylamide
- 4-(2 Hydroxyethyl)-1-piperazineethanesulfonic acid
- 2-(N morpholino)-ethanesulfonic acid
- Apigenin ( $\geq 97 \%$ purity)
- Baicalein ( $98 \%$ purity)

Table 3.1 Chemical reagents and sources (Continued).

| Reagent | Source |
| :--- | :--- |
| • Luteolin ( $\geq 98 \%$ purity $)$ | Sigma-Aldrich |
| - $\alpha$-Mangostin ( $\geq 98 \%$ purity $)$ |  |
| - Naringenin ( $\geq 95 \%$ purity $)$ |  |
| - Restriction enzymes |  |
| - EcoRI |  |
| - SfiI |  |
| - XhoI |  |
| - SalI |  |

- Agarose

Vivantis

- Deoxyribonucleotide triphosphate
- DNA ladder VC 100 bp plus
- Taq DNA polymerase
- T4-DNA ligase
- Protein maker
- Tetramethylethylenediamine


### 3.1.2 Instruments and equipment

3.1.2.1 TC-PLUS thermal cycler (Techne)
3.1.2.2 DNA gel electrophoresis apparatus (Amersham)
3.1.2.3 Vertical gel electrophoresis apparatus (Bio-RAD)
3.1.2.4 Innova 4300 shaker incubator (Brunswick Scientific)
3.1.2.5 Sorvall legend XFR centrifuge (ThermoScience)

### 3.1.2.6 Nano drop 2000 spectrophotometer (ThermoScience)

3.1.2.7 Labsystems iEMS Microplate Reader MF (MTX LabSystems)

### 3.1.3 Bacterial stains and plasmids

### 3.1.3.1 Bacterial template

Bacilus subtilis (TISTR No. 001, Thailand Institute of Scientific and Technological Research, Thailand).

### 3.1.3.2 Escherichia coli strains

1.) E. coli DH5 $\alpha$
E. coli strains $\mathrm{DH} 5 \alpha$ is suitable for cloning of genes. The mutations that the $\mathrm{DH} 5 \alpha$ has are: $\mathrm{F} \Phi 80 \mathrm{lacZ} \Delta \mathrm{M} 15 \Delta$ (lacZYA-argF) U 169 recA1 endA1 hsdR17(rK-mK+) supE44 - thi-1 gyrA96 relA1 (Taylor, Walker, McInnes et al., 1993). These mutations allow for blue-white screening, lower endonuclease degradation (which ensures higher plasmid transfer rates), increase insert stability and improve the quality of plasmid DNA preparation.
2.) E. coli BL21(DE3)
E. coli strains BL21(DE3) is used for performing protein expression that utilizes the T7 RNA polymerase promoter to direct high-level expression (Studier and Moffatt, 1986). The mutations that the BL21 has are: $\mathrm{F}^{-}$ompT gal dcm lon $\mathrm{hsdS}_{\mathrm{B}}\left(\mathrm{r}_{\mathrm{B}}{ }^{-} \mathrm{m}_{\mathrm{B}}{ }^{-}\right) \lambda(\mathrm{DE} 3)$. These expression strains naturally lack the Lon protease, which can degrade recombinant proteins. In addition, these strains are engineered to be deficient for a second protease, the OmpT protein (Grodberg and Dunn, 1988).

### 3.1.3.3 Plasmid

1.) pTG19-T cloning plasmid pTG19-T is a plasmid designed for rapid and efficient cloning of PCR products with $3^{\prime}-\mathrm{dA}$ overhangs (Figure 3.1). The linearized plasmid was engineered with a multiple cloning in the gene for $\beta$-galactosidase (also known as lacZ) (Langley, Villarejo, Fowler et al., 1975). Successful cloning of foreign DNA into the multiple cloning sites interrupts lacZ the genes producing a non-functional $\beta$ galactosidase enzyme. An unsuccessful cloning will produce a functional enzyme (Langley et al., 1975). The ligated plasmid without a foreign gene produces the functional enzyme that is able to hydrolyze X-gal, which creates a blue colony on LB agar plate supplemented by X-gal and IPTG.
2.) pSY5 expression plasmid

The pSY5 plasmid was modified from pET-21d(+), designed by the (Robert C Robinson group), Institute of Molecular and Cell Biology (IMCB), A*Star, Singapore. This plasmid enables expression the protein under the control of the T7 promoter. The pSY5 is encodes an N-terminal, His8-tag, followed by a human rhinovirus 3C protease cleavage site ahead of the N -terminus of the protein.


AGT ACC AGT TAG CTG TTT CCT GTG TGA AAT TGT TAT CCG CTC ACA ATT CCA CAC AAC ATA TCA TGG TCA ATC GAC AAA GGA CAC ACT TTA ACA ATA GGC GAG TGT TAA GGT GTG TTG TAT M13 Reverse primer


Figure 3.1 Map and multiple cloning site sequence of pTG19-T vector.

8x His - PreScission - MCS

ATG CGT CCG GCG tAg AgG atc gag atc tcg atc ccg cGA AAT TAA TAC GAC TCA CTA TAG
$\overrightarrow{\text { GGG }} \frac{l a c o p e r a t o r}{} \quad$ TGT GAG CGG ATA ACA ATT $C C C$ CTC TAG AAA TAA TTT TGT TTA ACT TTA AGA AGG

- AGA TAT ACC ATG GCA GAA GAA CAC CAC CAC CAC CAC CAC CAC CAC CTG GAA GTT CTG TTC

$\overline{T A A} C T A ~ G C A ~ T A A ~ C C C ~ C T T ~ G G G ~ G C C ~ T G A ~ T C T ~ A A A ~ C G G ~ G T C ~ T T G ~ A G G ~ G G ~ N o t e s: ~$
Big triangle denotes protease cleavage site Number refers to the nucleotide sequence

Expression vector based on pET-21d(+) (Novagen, cat no 69743-3).


Figure 3.2 Map and multiple cloning site sequence of pSY5.

### 3.1.4 Oligonucleotides

The oligonucleotides (primers) used for genes amplification were purchased from Sigma-Aldrich (Singapore) Co., Ltd. and are shown in Table 3.1.

Table 3.2 Details of primers.

| Construction | Primer | Sequence | Cloning |
| :--- | :--- | :--- | :--- |
| site |  |  |  |

### 3.2 General Methods

### 3.2.1 $\mathbf{C a C l}_{\mathbf{2}}$ Competent cell preparation

The E. coli DH5 $\alpha$ and BL21(DE3) were grown in 5 ml LB medium (10 $\mathrm{g} / \mathrm{l}$ peptone, $5 \mathrm{~g} / \mathrm{l}$ yeast extract, $5 \mathrm{~g} / \mathrm{l} \mathrm{NaCl})$ at $37^{\circ} \mathrm{C}$, 200 rpm for $16-18$ hours as a starter, then 1 ml starter was inoculated to 100 ml LB and cultured at $37^{\circ} \mathrm{C}, 200 \mathrm{rpm}$ until the optical density at 600 nm reached $0.4-0.5$. The cell pellet was collected by centrifugation at $3,000 \mathrm{xg}$, for 10 minutes, at $4^{\circ} \mathrm{C}$. The pellet was suspended in 10 ml of cold $0.1 \mathrm{M} \mathrm{CaCl}_{2}$, gently mixed and stored on ice for 10 minutes. The cell
suspension was centrifuged at $3,000 \mathrm{xg}$, for 10 minutes, at $4^{\circ} \mathrm{C}$, the supernatant discarded, and 2 ml of cold $15 \%$ glycerol in $0.1 \mathrm{M} \mathrm{CaCl}_{2}$ was added, and the cell suspension was mixed very well and aliquot into the micro centrifuge tubes with 100 $\mu \mathrm{l}$ per tube. The tubes of competent cells were used immediately or kept at $-80^{\circ} \mathrm{C}$.

### 3.2.2 Heat shock transformation of $\mathbf{C a C l}_{2}$ competent cell

One hundred microliters of competent cell in 1.5 ml micro centrifuge tube was thawed on ice. One microliter of plasmid (or $\sim 100 \mathrm{ng}$ ) was added to the cell and mixed by gently stirring the pipette tip. Incubation of the cells continued on ice for 30 minutes. The cells were heat pulsed at $42^{\circ} \mathrm{C}$ for 45 seconds followed by 2 minutes incubation on ice. The LB medium ( 0.9 ml ) was added to each tube of cells and the tubes were incubated at $37^{\circ} \mathrm{C}$ for 1 hour. A sample of $50 \mu 1$ was plated out on agar plates with appropriate antibiotics for selection. Plates were incubated at $37{ }^{\circ} \mathrm{C}$ overnight.

### 3.2.3 Cloning

DNA encoding the genes of $m r e B$ (Gene ID: 936759) and ftsZ (Gene ID: 935971) of B. subtilis were amplified by polymerase chain reaction (PCR). All forward primers contained a SfiI recognition site and reverse primers containing an EcoRI recognition site (Table 3.2). Chemical compositions of the PCR and thermocycler program are shown in Tables 3.3 and 3.4. The genes were amplified directly from B. subtilis cell by using Taq DNA polymerase. PCR products were cleaned by PCR Clean-up Kit. The concentration of cleaned PCR products was measured by Nano drop spectrophotometer. The cleaned PCR products were ligated into pTG19-T plasmid by using T4 ligase. The details of the ligation reaction are shown in Table 3.5. The reaction was incubated at $22{ }^{\circ} \mathrm{C}$ for 1 hour. The ligation product was
transformed into E. coli DH5a. The transformation reaction was spread onto an LB agar plate containing $100 \mu \mathrm{~g} / \mathrm{ml}$ ampicillin, 1 mM IPTG, and $50 \mu \mathrm{~g} / \mathrm{ml}$ X-gal.

Table 3.3 Chemical compositions of the PCR.

| Composition | Volume ( $\boldsymbol{\mu l}$ ) | Final concentration |
| :--- | :--- | :--- |
| $10 \times \mathrm{S}$ buffer | 2.5 | 1 x |
| dNTPs ( 0.2 mM each dNTP) | 2.5 | 0.2 mM |
| $2 \mathrm{mM} \mathrm{MgCl}_{2}$ | 2.0 | 2.0 mM |
| $5 \mu \mathrm{M}$ Primer forward | 2.5 | 0.5 mM |
| $5 \mu \mathrm{M}$ Primer reverse | 2.5 | 0.5 mM |
| DNA template | 2.0 | - |
| Taq DNA polymerase | 0.25 | 10. |
| Distilled water | $\mathbf{1 0 . 7 5}$ |  |
| Total reaction volume | $\mathbf{2 5}$ |  |
| The constructs were confirmed by DNA sequencing (Macrogen, Korea) |  |  | with the M13 forward primer and M13 reverse primer. The plasmid containing mreBBs was named pTG19-mreB-Bs and the plasmid containing ftsZ-Bs was named pTG19-ftsZ-Bs. The verified pTG19-mreB-Bs plasmid was digested and the mreB-Bs insert cloned into pSY5 at SfiI and EcoRI cloning sites (Table 3.6). The plasmid containing mreB-Bs was named pSY5-mreB-Bs. The verified pTG19-ftsZ-Bs was digested with SfiI and SalI and the insert cloned into pSY5 at SfiI and XhoI cloning sites (Table 3.6). The plasmid containing ftsZ-Bs was named $\mathrm{pSY} 5-f t s Z-B s$.

Table 3.4 Cycling parameters of the PCR.

| Step | Temperature $\left({ }^{\circ} \mathbf{C}\right.$ ) | Time (min) |
| :--- | :--- | :--- |
| Initial Denaturation | 95 | 5 |
| Denaturation | 95 | $0.30 \quad$ |
| Annealing | 68 | 0.30 |
| Extension | 72 | $1.30 \quad 30$ cycles |
| Polish extension | 72 | 5 |

Table 3.5 Composition of ligation reaction of the PCR product into the cloning plasmid.

| Composition | Volume ( $\mu \mathrm{l})$ |
| :--- | :---: |
| pTG19 $(25 \mathrm{ng} / \mu \mathrm{l})$ | 2 |
| Fresh PCR product $(55 \mathrm{ng} / \mu \mathrm{l})$ | 1 |
| 10X Buffer Ligase | 1 |
| T4 DNA Ligase (200u/ $\mu \mathrm{l})$ | 0.2 |
| Distilled water | $\mathbf{1 0} \boldsymbol{\mu l}$ |
| Total Volume |  |

Table 3.6 Composition of ligation reaction putting inserts into the expression plasmid.

| Composition | Volume $(\mu \mathrm{l})$ |
| :--- | :--- |
| pSY5 $(25 \mathrm{ng} / \mu \mathrm{l})$ | 4 |
| Gene insert $(20 \mathrm{ng} / \mu \mathrm{l})$ | 3 |
| 10X Buffer Ligase | 1 |
| T4 DNA Ligase $(200 \mathrm{U} / \mu \mathrm{l})$ | 0.2 |
| Distilled water | 1.8 |
| Total Volume | $\mathbf{1 0} \boldsymbol{\mu \mathrm { l }}$ |

### 3.3 Screening and optimization of protein expression

### 3.3.1 Screening of protein expression

The pSY5-mreB-Bs and pSY5-ftsZ-Bs plasmids were transformed into E. coli BL21(DE3) cells (expression cells). Ten single-colonies were selected from each transformation plate (pSY5-mreB-Bs or pSY5-ftsZ-Bs). Each single-colony was inoculated in 10 ml of LB media containing $100 \mu \mathrm{~g} / \mathrm{ml}$ ampicillin. The bacterial cultures were incubated at $37^{\circ} \mathrm{C}$, 200 rpm in a shaker incubator for 16 hours. Then, $2.5 \mathrm{ml}(5 \%)$ of each starter culture was inoculated in 47.5 ml of LB media containing $100 \mu \mathrm{~g} / \mathrm{ml}$ ampicillin. The cells were grown at $37^{\circ} \mathrm{C}, 200 \mathrm{rpm}$ in a shaker incubator until optical density at 600 nm reached $0.6-0.8$. The cultures were induced with 1 mM IPTG and incubated at $20^{\circ} \mathrm{C}, 200 \mathrm{rpm}$ in shaker incubator for 16 hours.

The induction cells were harvested by centrifugation at $3,000 \mathrm{xg}$ for 15 minutes. The cell pellets were suspended in 1 ml of 50 mM Tris- $\mathrm{HCl}, \mathrm{pH} 8,150 \mathrm{mM}$ NaCl , and $0.1 \mathrm{mg} / \mathrm{ml}$ lysozyme incubated at $37{ }^{\circ} \mathrm{C}$ for 30 minutes. The lysate were
clarified by centrifugation at $12,000 \mathrm{xg}$ for 10 minutes. The supernatant was applied to $20 \mu \mathrm{l}$ of $\mathrm{Ni}^{2+}-\mathrm{NTA}$ resin and gently shaken for 10 minutes. The resin were washed with 50 mM Tris- $\mathrm{HCl}, \mathrm{pH} 8.0,20 \mathrm{mM}$ imidazole, and 500 mM NaCl . The soluble protein was verified by SDS-PAGE.

### 3.3.2 Optimization of protein expression

The best expression clones were inoculated in 10 ml of LB media containing $100 \mu \mathrm{~g} / \mathrm{ml}$ ampicillin. The starter cultures were incubated at $37^{\circ} \mathrm{C}, 200$ rpm in shaker incubator for 16 hours. Then, 2.5 ml of starter culture was inoculated in 47.5 ml of LB media containing $100 \mu \mathrm{~g} / \mathrm{ml}$ ampicillin. The cells were cultured in LB media containing $100 \mu \mathrm{~g} / \mathrm{ml}$ ampicillin until the optical density at 600 nm reached $0.6-0.8$. The optimum concentration of IPTG and temperature for the protein expression were investigated by varying the concentration of IPTG from $0.25-1.00$ mM , and temperature from $20-30{ }^{\circ} \mathrm{C}$. The cell induced were collected by centrifugation at $3,000 \mathrm{xg}$ and suspended in 50 mM Tris- $\mathrm{HCl}, \mathrm{pH} 8.0,150 \mathrm{mM} \mathrm{NaCl}$, and $0.1 \mathrm{mg} / \mathrm{ml}$ lysozyme. The cell suspensions were incubated at $37^{\circ} \mathrm{C}$ for 30 minutes and the lysate was clarified by centrifugation at $12,000 \mathrm{xg}$ for 10 minutes. The supernatant was applied to $20 \mu \mathrm{l}$ of $\mathrm{Ni}^{2+}$-NTA resin. The resin was washed with 50 mM Tris-HCl, $\mathrm{pH} 8.0,20 \mathrm{mM}$ imidazole, and 500 mM NaCl . The soluble protein was verified by SDS-PAGE.

### 3.4 Protein purification

The expression cell pellets were suspended in 50 mM Tris-HCl, pH 8.0 , and 150 mM NaCl and lysed by sonication. The lysate was clarified by centrifugation at $12,000 \mathrm{xg}$ for 30 minutes. The supernatant was applied to a $\mathrm{Ni}^{2+}$ - NTA column. The
proteins were purified by $\mathrm{Ni}^{2+}$-NTA with stepwise elution using imidazole concentrations of $0,20,40,60,80,100,120,140,160,180,200,220$, and 240 mM in 500 mM NaCl and 50 mM Tris- $\mathrm{HCl}, \mathrm{pH} 8$. The pure proteins were dialyzed against buffer containing 50 mM Tris- $\mathrm{HCl}, \mathrm{pH} 8.0$, and 150 mM NaCl . The proteins were concentrated by a 30 kDa molecular mass cut off (MWCO) Centricon centrifugal filter.

### 3.5 Nucleotide removal

To examine the effects of concentration of nucleotide on the polymerization of MreB-Bs and FtsZ-Bs, nucleotides were removed from the proteins by treating the proteins in Dowex resin following Bean and Amann (2008). The proteins were mixed with $1 / 4$ its volume of $50 \%$ Dowex resin ( $1 \mathrm{X} 8-400 \mathrm{Cl}$ ) slurry in 10 mM Tris- $\mathrm{HCl}, \mathrm{pH}$ 8.0. The reaction was incubated on ice for 15 minutes with gentle mixing every $2-3$ minutes, the mixture was centrifuged at $4^{\circ} \mathrm{C}$ at $10,000 \mathrm{xg}$ for 2 minutes and the supernatant transferred to a separate tube. This procedure was carried out three times to ensure complete nucleotide removal.

### 3.6 Phosphate release assays

### 3.6.1 Effect of $\mathbf{p H}$ on nucleotides hydrolysis by the proteins

The ATP hydrolysis and GTP hydrolysis activity of MreB-Bs at different pH values was investigated by the malachite green assay of released phosphate (Carter and Karl, 1982) by buffering the reaction with sodium acetate, pH 5.5, MES pH 6.0, MES, pH 6.5 , HEPES, pH 7.0 , Tris-HCl, pH 7.5 and, Tris-HCl, pH 8.0.

The effect of pH on nucleotides hydrolysis by MreB-Bs was tested by mixing $5 \mu \mathrm{M}$ MreB-Bs in 50 mM buffer, $100 \mathrm{mM} \mathrm{NaCl}, 4 \mathrm{mM} \mathrm{MgCl} 2$, and 0.2 mM ATP or GTP (van den Ent et al., 2001).

The effect of pH on nucleotides hydrolysis by FtsZ-Bs was tested by mixing $10 \mu \mathrm{M}$ FtsZ-Bs in 50 mM Buffer, $200 \mathrm{mM} \mathrm{KCl}, 5 \mathrm{mM} \mathrm{MgCl} 2$, and 0.2 mM ATP or GTP (Matsui, Han, Yu et al., 2013).

The reactions were stopped by adding an equal volume of cold 0.6 M perchloric acid. Two volumes of filtered malachite green solution ( 0.15 g malachite green, 1 g sodium molybdate, 0.25 g Triton $\mathrm{X}-100$ in 500 ml 0.7 M HCl$)$ was added to the supernatants and the mixtures incubated at room temperature for 15 minutes. The absorbance at 620 nm was read by spectrophotometer. The phosphate concentrations were calculated from the standard curve plot (Appendix B), (Geladopoulos et al., 1991).

### 3.6.2 Screening of natural products effect on nucleotides hydrolysis by the

 proteins3.6.2.1 Screening of natural products effect on nucleotides hydrolysis of MreB-Bs

MreB-Bs $(5 \mu \mathrm{M})$ was mixed in 50 mM Tris-HCl, $\mathrm{pH} 7.5,100$ $\mathrm{mM} \mathrm{NaCl}, 4 \mathrm{mM} \mathrm{MgCl} 2$ and $200 \mu \mathrm{M}$ of one natural products (apigenin, baicalein, luteolin, $\alpha$-mangostin, or naringenin), and $200 \mu \mathrm{M}$ ATP or GTP was added and incubated for 1 hour.
3.6.2.2 Screening of natural products effects on nucleotides hydrolysis by FtsZ-Bs

FtsZ-Bs ( $10 \mu \mathrm{M}$ ) was mixed in 50 mM MES, $\mathrm{pH} 6.5,200 \mathrm{mM}$ $\mathrm{KCl}, 5 \mathrm{mM} \mathrm{MgCl} 2$ and $200 \mu \mathrm{M}$ of one natural products (apigenin, baicalein, luteolin, $\alpha$-mangostin, or naringenin), and $200 \mu \mathrm{M}$ GTP was added and incubated for 1 hour.

The reaction was stoped by adding an equal volume of cold 0.6 M perchloric acid. Two volumes of filtered malachite green solution ( 0.15 g malachite green, 1 g sodium molybdate, 0.25 g Triton X -100 in 500 ml 0.7 M HCl$)$ were added to the supernatants and the mixtures incubated at room temperature for 15 minutes. The absorbance at 620 nm was read by spectrophotometer. The phosphate concentrations were calculated from the standard curve plot (Appendix B), (Geladopoulos et al., 1991).

## CHAPTER IV

## RESULTS AND DISCUSSION

### 4.1 Cloning

### 4.1.1 Amplification of $m r e B-B s$ and $f t s Z-B s$ by PCR

DNA encoding the genes of mreB and $f t s Z$ of $B$. subtilis were amplified from the B. subtilis cells by Taq DNA polymerase with the pair of gene specific primers (Table 3.2). The best annealing temperature for the two genes was $62^{\circ} \mathrm{C}$. The PCR products with the expected sizes of $m r e B-B s(1,041 \mathrm{bp})$ and $f t s Z-B s(1149 \mathrm{bp})$ were obtained, as show in Figure 4.1.


Figure 4.1 Agarose gel electrophoresis analysis of $m r e B-B s$ and $f t s Z-B s$ amplification: Lane M; DNA ladder (VC 100 bp plus, Vivantis), (A) lanes 1-3; PCR product of $m r e B-B s,(B)$ lanes $1-3$; PCR product of $f t s Z-B s$.

### 4.1.2 Screening of pTG19-T cloning vector containing the mreB-Bs and

## ftsZ-Bs by PCR

The purified PCR products were ligated into pTG19-T cloning vector by using T4 ligase. The ligation products were transformed into E. coli DH5a. The transformation reaction was spread on LB agar plate supplemented by $2 \%$ (w/v) X-gal and 1 mM IPTG. The ligated plasmid without a foreign gene produces the functional enzyme that is able to hydrolyze X-gal, which creates a blue colony on the agar plate. Thus, the blue colony is a clone without insert and then the white colony is a positive clone (the clone with non-functional $\beta$-galactosidase). However, false white colonies can occur but doesn't have insert is possible, because single 3 '- dT overhangs on the vector may be degraded.

Therefore, colony PCR was performed by using the sets of cloning primers to screen and verify the plasmid containing gene insert. Figure 4.2 shows that clones numbers 2-8 yielded the PCR product with the size of $\operatorname{mreB-Bs}(1,041 \mathrm{bp})$, suggesting the presence of the insert of mreB-Bs in the pTG19. In the same way, Figure 4.3 shows that clones numbers $1,3-6$, and 10 produced the PCR products with the expected size of ftsZ-Bs (1149 bp), suggesting the insert of $f t s Z-B s$ in the pTG19.


Figure 4.2 Agarose gel electrophoresis analysis of mreB-Bs in pTG19 amplification: Lane M, DNA ladder; lanes $1-11$, PCR products that show the presence or absence of $m r e B-B s$ in $\mathrm{pTG} 19-\mathrm{T}$ clones numbers $1-11$.


Figure 4.3 Agarose gel electrophoresis analysis of ftsZ-Bs in pTG19 amplification: Lane M, DNA ladder; lane $1-11$, PCR products that show the presence or absence of $f t s Z-B s$ in pTG19-T clones numbers $1-11$.

### 4.1.4 Verification of the genes in pTG19-T DNA sequencing

The sequencing was carried out by Macrogen Inc. (Korea), (the details are shown in Appendix A). The pTG19 containing genes ftsZ-Bs was sequenced by using the M13 universal primers in both directions. The result showed that the pSY5-
mre $B-B s$ contains the mre $B-B s$ without mutation. However, the pTG19-ftsZ-Bs contained the mutation on the reverse primer (FtsZ-Bs Rev) at the late base of the EcoRI recognition site (from C to A, show in Figure 4.4). However, the SalI restriction site on the pTG19-ftsZ-Bs after the mutated EcoRI site remains intacted. We deiced to used SalI instead of EcoRI, because of the compatibility of the cohesive end of SalI and XhoI recognition sites (on the cloning site of pSY5). As a result, the construct was actually cloned into pSY5 at the SfiI and XhoI recognition sites.


Figure 4.4 The mutation of pTG19-ftsZ-Bs on the late base of EcoRI recognition site of the revers primer.

### 4.1.3 Cloning and screening of mreB-Bs and ftsZ-Bs into expression

 plasmid (pSY5)The verified genes were excised from the cloning plasmids (pTG19mreB and pTG19-ftsZ-Bs) and ligated into the expression plasmid. The ligation products were transformed into E. coli $\mathrm{DH} 5 \alpha$. PCR using the sets of cloning primers were performed to screen the E. coli DH5 $\alpha$ clones containing mreB-Bs and $f t s Z-B s$ in pSY5. Figure 4.5 shows that clone numbers $1-10$ gave the PCR product with the expected size of $m r e B-B s(1,041 \mathrm{bp})$, suggested the insert of $m r e B-B s$ in the pSY5.

The PCR product with the expected size of ftsZ-Bs (1149 bp) of clones $1,2,4,6-10$ suggested the ftsZ-Bs in these pSY5 (Figure 4.6).


Figure 4.5 Agarose gel electrophoresis analysis of mreB-Bs in pSY 5 amplification: Lane M, DNA ladder; lanes $1-10$ : PCR products that show the presence of $m r e B-B s$ in pSY5 clones numbers $1-10$.


Figure 4.6 Agarose gel electrophoresis analysis of $f t s Z-B s$ in pSY 5 amplification: Lane M, DNA ladder; lanes $1-11$, PCR products that show the presence or absence of $f t s Z-B s$ in pSY 5 clones numbers 1 - 11 .

### 4.2 Expressions and Purifications of MreB-Bs and FtsZ-Bs

### 4.2.1 Optimization of protein expression

Expressed His8-tagged proteins can be purified and detected easily because the string of 8 his8-tidine residues binds to IMAC $\left(\mathrm{Ni}^{2+}-\mathrm{NTA}\right)$, under specific buffer conditions. The expression plasmids containing the target genes, pSY5-mreB$B s$ and $\mathrm{pSY} 5-\mathrm{ftsZ} Z-B s$, were transformed into the expression strain, E. coli BL21(DE3). The expression cells were screened for the best clones in terms of expression by a small scale purification using $\mathrm{Ni}^{2+}-\mathrm{NTA}$.

The expressions of selected clones were optimized by varying the IPTG concentration and induction temperature. The protein expressions levels were analyzed by a small scale purification using $\mathrm{Ni}^{2+}-\mathrm{NTA}$ followed by SDS-PAGE. The patterns of proteins expressed from cell lysates from expression of MreB-Bs are show in Figures 4.7 A and the protein bound with $\mathrm{Ni}^{2+}$-NTA resins shown in Figure 4.7B. The black arrow indicates the protein with the 36 kDa .

The protein expression patterns of FtsZ-Bs were analyzed by SDSPAGE. The cell lysates are shown in Figures 4.8 A and the protein bound with $\mathrm{Ni}^{2+}$ NTA resin is shown in Figures 4.8B. The black arrow indicates the protein with the 38 kDa . The over expressions of MreB-Bs and FtsZ-Bs are not significantly different in every test condition. According to the screening tests, the optimal condition of the expression was selected at lower IPTG concentration ( 0.25 mM ) and lower temperature $\left(20^{\circ} \mathrm{C}\right)$.


Figure 4.7 SDS-PAGE of MreB-Bs expression pattern after induction with different concentrations of IPTG $(0.25,0.50$, and 1.00 mM$)$ at difference temperatures $(20,25$, and $30{ }^{\circ} \mathrm{C}$ ). (A) Cell lysated, and (B) proteins bound to $\mathrm{Ni}^{2+}-\mathrm{NTA}$ resin. The black arrow indicates MreB-Bs.


Figure 4.8 SDS-PAGE of FtsZ-Bs expression pattern after induction with different concentrations of IPTG $(0.25,0.50$, and 1.00 mM$)$ at difference temperatures $(20,25$, and $30{ }^{\circ} \mathrm{C}$ ). (A) Cell lysated, and (B) proteins bound to $\mathrm{Ni}^{2+}-\mathrm{NTA}$ resin. The black arrow indicates FtsZ-Bs.

### 4.2.2 Protein purification

The selected clones were expressed in the optimal condition ( 0.25 mM IPTG at $20^{\circ} \mathrm{C}$ ). The proteins were purified by $\mathrm{Ni}^{2+}$-NTA with stepwise elution using imidazole concentrations of $0,20,40,60,80,100,120,140,160,180,200,220$, and 240 mM in 500 mM NaCl and 50 mM Tris- $\mathrm{HCl}, \mathrm{pH}$ 8. The MreB-Bs was released out from the resin at the concentration of imidazole from $100-240 \mathrm{mM}$ (Figure 4.9). At fractions of $100-120 \mathrm{mM}$ imidazole were discarded due to their containing a significant amount of contaminating proteins. The FtsZ-Bs was released out from the resin at the concentration of imidazole from $80-180 \mathrm{mM}$ (Figure 4.10). However, the fraction at 80 mM imidazole was discarded, sine it contaminated a significant amount of contaminated protein.

The fractions of $140-240 \mathrm{mM}$ imidazole of MreB-Bs and the fractions with $100-180 \mathrm{mM}$ imidazole of FtsZ-Bs appeared to be more than $95 \%$ pure bases on SDS-PAGE profiles and were saved for the further experiments. Totally, approximately 1 mg of $\mathrm{MreB}-\mathrm{Bs}$, and approximately 2 mg of the FtsZ-Bs were obtained from 1 L of $E$. coli BL21(DE3) culture. Each protein was pooled and concentrated and removed nucleotides were removed by adsorption to Dowex resin. The proteins then were analyzed by SDS-PAGE, as show in Figure 4.11.


Figure 4.9 SDS-PAGE of MreB-Bs purification faction, including washes and elution with different imidazole concentrations. The black arrow indicates the MreBBs with an approximate size of 36 kDa .


Figure 4.10 SDS-PAGE of FtsZ-Bs purification faction, including washes and elution with different imidazole concentrations. The black arrow indicates the MreBBs with an approximate size of 38 kDa .


Figure 4.11 SDS-PAGE of MreB-Bs and FtsZ-Bs purified after removed nucleotide by 1X8-400Cl Dowex resin. The black arrows indicate MreB-Bs (approximately size 36 kDa and FtsZ-Bs approximate size 38 kDa ).

### 4.3 Phosphate release assay

The phosphate release assay is a simplified method based on free inorganic phosphate determination. The liberation of orthophosphate from ATP and GTP was generated by ATPases and GTPases. The technique is simple and cost effective functional assay for this class of enzymes.

Malachite green assay is a colorimetric method for measuring free inorganic phosphate. The assay is based on the formation of malachite green molybdophosphoric acid complex that absorbs light at 620-640 nm. It is directly related to the free inorganic phosphate concentration (Figure 4.12) (D'Angelo, Crutchfield, and Vandiviere et al., 2001). This application is a reliable and suitable means of quantifying minimal amounts of $1-100 \mu \mathrm{M}$ of inorganic phosphate and amenable to high-throughput screening applications (Attin, Becker, Hannig et al.,
2005). This includes quantification of phosphorylation and phosphate release from protein phosphatase substrates (Maehama, Taylor, Slama et al., 2000).

Releasing orthophosphate is a key factor in the control of MreB and FtsZ function. ATPase activity is a consequence of MreB polymerization. After the ATP hydrolysis, free inorganic phosphate is release from the protein filament. $₫ D$ During the course of MreB polymerization, there is a lag between polymerization and phosphate release (Esue, Cordero, Wirtz et al., 2005). This indicates that ATP hydrolysis occurs after MreB monomers are assembled into filaments (Esue et al., 2005). Likewise, the GTP-dependent assembly of FtsZ into protofilaments is followed by hydrolyzes GTP to GDP by an active site formed between two associated FtsZ monomers (PachecoGo'mez, Roper, Dafforn et al., 2011).


Figure 4.12 Representative of free inorganic phosphate in the malachite green assay. The samples containing malachite green molybdophosphoric acid complex present green color, the absence present yellow color, (yellow-green gradient bar shows increasing of inorganic phosphate ion concentration).

The ATP and GTP hydrolysis activities of MreB-Bs and FtsZ-Bs at different pH were investigated by the malachite green assay. The reactions under the condition were described in the method section 3.4.

### 4.3.1 Effect of $\mathbf{p H}$ on nucleotides hydrolysis by MreB-Bs

### 4.3.1.1 Effect of pH on ATP hydrolysis by MreB-Bs

MreB-Bs hydrolyzed ATP over a wide pH range. The inorganic phosphate $6.80,16.50,14.90,31.11,33.70$, and $33.24 \mu \mathrm{M}$ were released from ATP in the experiment at $\mathrm{pH} 5.5,6.0,6.5,7.0,7.5$, and 8.0, respectively (Figure 4.13).


Figure 4.13 pH effect on phosphate release from ATP by MreB-Bs.

### 4.3.1.2 Effect of pH on GTP hydrolysis by MreB-Bs

MreB-Bs hydrolyzed GTP over a wide pH range. The inorganic phosphate concentrations of 20.64, 28.64, 44.43, 39.81, 25.77, and 25.26 $\mu \mathrm{M}$ were released from ATP in the experimant at $\mathrm{pH} 5.5,6.0,6.5,7.0,7.5$, and 8.0 , respectively (Figure 4.14).


Figure 4.14 pH effect on phosphate release from GTP by MreB-Bs.

MreB is a member of a larger superfamily of proteins carrying ATPase activity (van den Ent et al., 2001) that can bind and hydrolyze both ATP and GTP. Our study found that the ATPase activity of MreB-Bs was favored at higher pH 7.0 8.0, but strongly favored at pH 7.5 , while the GTPase activity was most favored at pH $6.5-7.0$, but and highest at pH 6.5 . The nucleotide hydrolysis characteristics of MreB-Bs was similar to that from other species that have been reported, however nucleotide hydrolysis by MreB-Bs is not necessary for polymerization, as in other study species, as shown in Table 4.1.

Table 4.1 Characters of polymerization of MreB.

| MreB | pH <br> range | Optimal pH | Ion |  | NTP |  | Critical concentration ( $\mu \mathrm{M}$ ) | NTPase <br> dependent <br> polymerization | Gram/ <br> Morphology | Reference |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
|  |  |  | $\mathbf{M g}{ }^{2+}$ | $\mathrm{Ca}^{2+}$ | ATP | GTP |  |  |  |  |
| T. maritime | 4-9.5 | 6-7 | Yes | Yes | Yes | Yes | 5 | Yes | Negative/ | van den Ent et al., 2001 |
| (MreB-Tm) |  |  |  |  |  |  |  |  | rod-shape | Bean and Amann, 2009 |
|  |  |  |  |  |  |  |  |  | enveloped |  |
| B. subtilis | 5.5-8 | 5.5 | Yes | Yes | Yes |  | 0.9 | No | Positive/ | Mayer and Amann, |
| (MreB-Bs) |  |  |  |  |  |  |  |  | rod-shaped | 2009 |
| C. pneumonia | 5.5-7 | 6.5 | Yes | N/A | Yes | Yes | N/A | No | Negative/ | Gaballah et al., 2011 |
| (MreB-Cp) |  |  |  |  |  | $75$ |  | $a{ }^{5}$ | rod-shaped |  |
| E. coli | N/A | 7 | Yes | Yes | Yes | Yes | 1.5 | Yes | Negative/ | Nurse and Marians, |
| (MreB-Ec) |  |  |  |  |  |  |  |  | coccobacilli | 2012 |
| C. crescentus | N/A | 8 | Yes | N/A | Yes | N/A | N/A | Yes | Negative/ | van den Ent et al., 2014 |
| (MreB-Cc) |  |  |  |  |  |  |  |  | curved rod |  |



Figure 4.15 Phylogenetic tree visualizing the relationship of MreB proteins from various bacteria to each other and yeast actin. The tree was produced from the protein sequence alignment by using MEGA6. Yeast actin (GI: 170986), MreB-Bs (GI: 760457758), MreB-Cp (GI: 15618759), MreB-Ec (GI: 751252054), MreB-Cc (GI: 221234549), MreB-Tm (GI: 15988309).

The effect of pH on nucleotide hydrolysis by MreB has never been studied before. This study found that the optimum pH of the nucleotides hydrolysis by MreB-Bs was different from the optimum pH of the previously reported polymerization of the protein (Mayer and Amann, 2009). The result suggests that the two processes may be independent, which supports the earlier report that the polymerization of MreB-Bs does not require nucleotide hydrolysis (Mayer and Amann, 2009).

MreB shares $\sim 57 \%$ sequence identity among bacteria which, is less than its orthologue, actin. Actin is one of the most conserved proteins in eukaryotes, sharing $\sim 90 \%$ sequence identity. The conformational changes of actin regard to ADP- and ATP-bounding are innate properties of actin, which affect the polymerization process. The biochemical characters (Table 4.1) and the primary
structure analysis (Figure 4.15) classified MreB into two groups. The first group, which requires the nucleotides hydrolysis for the polymerization, comprises MreBTm, MreB-Ec, and MreB-Cc. The structural study (van den Ent et al., 2014) has demonstrated that the conformational changes of MreB-Cc result from the nucleotides binding.


Figure 4.16 Crystal structures of MreB in different nucleotide states reveal a propeller twist. (A) Superposition of ADP-MreB-Cc (orange) and AMPPNP-MreB-Cc (blue), a small movement of domain IB initiates the propeller twist observed upon polymerization. (B) Schematic drawing showing the propeller twists in MreB. The interdomain cleft narrows due to the movement of domain I towards domain II that is accompanied by a rotation of domain I resulting in flattening of the interfilament interface, M is indicated the membrane binding site (van den Ent, Izoré, Bharat et al., 2014).

This study proposed that the structure of MreB-Cc is an innate property that regulates the polymerization process, similar to actin. The second group consisting of MreB-Bs and MreB-Cp does not require the nucleotides hydrolysis for the polymerization. Biochemical studies (Mayer and Amann, 2009) have shown that MreB-Bs has very low critical concentration for the polymerization (Table 4.1) and monovalent cation $\left(\mathrm{K}^{+}\right)$inhibits the polymerization (Bean and Amann, 2009; Mayer and Amann, 2009). Together, this information suggests that MreB-Bs adopts a unique structure that has high affinity for polymerization. This unique structure is not affected by nucleotide binding.

### 4.3.2 Effect of $\mathbf{p H}$ on nucleotides hydrolysis by FtsZ-Bs

4.3.2.1 Effect of pH on ATP hydrolysis by FtsZ-Bs

ATP hydrolysis by FtsZ-Bs was not detected, since there was no release of free phosphate from ATP. The data are shown in Appendix B.

### 4.3.2.2 Effect of pH on GTP hydrolysis of FtsZ-Bs

FtsZ-Bs hydrolyzed GTP over a wide pH range. The inorganic phosphate concentrations of $9.00,13.71,14.81,11.15,9.76$, and $11.42 \mu \mathrm{M}$ were released from GTP in the experiment at $\mathrm{pH} 5.5,6.0,6.5,7.0,7.5$, and 8.0 , respectively (Figure 4.17).


Figure 4.17 pH effect on phosphate release from GTP by FtsZ-Bs.

FtsZ exhibits the GTP-dependent assembly, in which GTP is hydrolyzed to GDP by the active-site GTPase split across two monomers (PachecoGómez et al., 2011). Our experiment found that ATP hydrolysis of FtsZ-Bs was not detectable, indicating that was no release of free phosphate from ATP. The GTPase activity occurred over a wide pH range $(5.5-8.0)$, but was highest at pH 6.5 . The GTP hydrolysis character of FtsZ-Bs was similar to that reported earlier (Król and Scheffers, 2013).

### 4.4 Screening of natural products effects on nucleotides hydrolysis by the proteins

The effects of natural products on nucleotides hydrolysis by the proteins were screened against apigenin, baicalein, luteolin, $\alpha$-mangostin, and naringenin. Lysozyme and BSA were used as negative controls that there are has no nucleotide hydrolysis activity. The reactions were performed under optimal pH and condition as described in the methods section 3.4.

### 4.4.1 Screening of natural products effect on ATP hydrolysis by MreB-Bs

The effects of the natural products on ATP hydrolysis by MreB-Bs were screened by the malachite green assay in the presence and absence of the natural products (as described in section 3.4). Free phosphates at the final concentrations of $32.22,31.51,32.76,30.57,32.04$, and $29.34 \mu \mathrm{M}$, were released from ATP in the assay containing apigenin, baicalein, luteolin, $\alpha$-mangostin, and naringenin, respectively, compare to in the control reaction (Figure 4.18). Statistical analysis was used to analyze $p$-value (Appendix B). The results showed that the liberation of free phosphates from ATP in the presence of the natural products was not significantly different from that without the natural products shows no statistically significant (Appendix B). It indicated that the natural products have no effect on ATP hydrolysis by MreB-Bs under the conditions tested.


Figure 4.18 Screening of natural products effect on ATP hydrolysis by MreB-Bs.

### 4.4.2 Screening of natural products effects on GTP hydrolysis by MreB-Bs

The effects of the natural products on GTP hydrolysis by MreB-Bs were screened by the malachite green assay in the presence and absence of the natural products (as described in section 3.4). The final free phosphate concentrations were 38.37, $36.38,35.51,37.12,35.74$, and $38.93 \mu \mathrm{M}$, when phosphate was released from GTP in the reactions with apigenin, baicalein, luteolin, $\alpha$-mangostin, and naringenin, respectively compared to $38.37 \mu \mathrm{M}$ (Figure 4.19). The results were showed that the liberation of free phosphates from the reaction in the presence of the natural products was not significantly different compare with the reaction without the natural products (Appendix B). This indicated that the natural products have no effect on GTP hydrolysis by MreB-Bs.


Figure 4.19 Screening of natural products effect on GTP hydrolysis by MreB-Bs.

### 4.4.3 Screening of natural products effect on GTP hydrolysis by FtsZ-Bs

The effects of the natural products on GTP hydrolysis by FtsZ-Bs were screened by the malachite green assay in the presence and absence of the natural products (as described in section 3.4). The free phosphate concentrations of 8.38, $7.22,13.25,12.78$, and $12.72 \mu \mathrm{M}$ were obtained from release of phosphate from GTP in the reation containing apigenin, baicalein, luteolin, $\alpha$-mangostin, and naringenin, respectively, (Figure 4.19) compared to $12.56 \mu \mathrm{M}$ in the reaction absence of the natural products. The $p$-value indicated that the amount of phosphate liberated from GTP by FtsZ-Bs in the presence of luteolin, $\alpha$-mangostin, and naringenin showed no statistically significant difference compared to the control without the natural products. It suggested that luteolin, $\alpha$-mangostin, and naringenin have no effect on

GTP hydrolysis property of FtsZ-Bs. On the other hand, apigenin and baicalein significant decreased the GTPase activity of FtsZ-Bs to $33.3 \%$ and $42.5 \%$, respectively.


Figure 4.20 Screening of natural products effect on GTP hydrolysis by FtsZ-Bs

GTP hydrolysis occurs during the polymerization by one FtsZ molecule providing the GTP-binding site to a second molecule, and then the T7 loop provides the catalytic residues to cleave GTP into GDP and $\mathrm{P}_{\mathrm{i}}$. The results from the malachite green released phosphate assay indicated FtsZ-Bs has interaction with apigenin and baicalein, but the site of interaction is unclear.

Possibly, the position may be anywhere on the protein that affects GTP binding or interferes with the polymerization mechanism, as proposed in Figure 4.21. This includes any possible allosteric sites on the molecule, such as the GTP binding
site at the N -terminal subdomain, T 7 loop and $\mathrm{H} 6-\mathrm{H} 7$ loop on H 7 helix, and binding site at the C-terminal subdomain.


Figure 4.21 Proposed mechanism of straight-to-curved conformational switch. FtsZ bound with GDP is a soluble monomer in the R (Relaxed) state (A). When FtsZ is bound to GTP, FtsZ is assembled (B). The intermolecular interactions between bound GTP and the T7 loop of the second molecule induce a structural change of the molecule from the R to the T (Tense) state to form the straight protofilament (C). In the straight protofilament, the catalytic residues in the upper subunit hydrolyze GTP to GDP and an intermediate state of FtsZ with GDP in the straight protofilament is formed (D). After releasing phosphate, the FtsZ molecule returns to the R state, and the straight protofilament changes to a curved protofilament. Finally, the curved protofilament is disassembled to monomeric FtsZ (E) (Matsui, Han et al., 2013).

## CHAPTER V

## CONCLUSION AND FUTURE PERSPECTIVE

DNA encoding mreB and ftsZ of B. subtilis were cloned into the pSY5 plasmids and expressed in E. coli BL21(DE3). The proteins were purified by $\mathrm{Ni}^{2+}-\mathrm{NTA}$ with stepwise elution. Batch production of the proteins yielded approximately 1 mg of MreB-Bs, and approximately 2 mg of the FtsZ-Bs from 1 L of E. coli BL21(DE3) culture.

MreB-Bs was able to hydrolyze both ATP and GTP over a broad pH range (5.5 - 8.0). The optimum pH values for ATP and GTP hydrolysis were at pH 7.0 and pH 6.5, respectively. The results demonstrated that MreB-Bs has the optimum pH for nucleotide hydrolysis that is different from the reported optimum pH for protein polymerization (Mayer and Amann, 2009). Apigenin, baicalein, luteolin, $\alpha$ mangostin, and naringenin showed no effect on both ATP and GTP hydrolysis of MreB-Bs.

FtsZ-Bs was unable to hydrolyze ATP, but hydrolyzed GTP over a broad pH range ( $5.5-8.0$ ), with the optimum pH at 6.5 . This work found that apigenin and baicalein inhibited GTP hydrolysis by FtsZ-Bs. Apigenin decrease GTPase activity of FtsZ-Bs to $33.3 \%$ while baicalein decreased the GTPase activity of FtsZ-Bs by 42.5\%.

Together, the results suggest that MreB-Bs adopts a unique structure that is not different between the nucleotide-free and the nucleotides-bound states bound to (GTP,

GDP, ATP, or ADP), suggesting an optimum structure for the polymerization of MreB. However, the structures of MreB-Bs alone and with nucleotides are not yet determined. Thus, structural investigation of MreB-Bs is needed.

Evidently, FtsZ is one molecular targets of apigenin and baicalein. We speculated that the interactions between apigenin and baicalein and FtsZ may interfere with the polymerization of the protein (Z-ring formation) and/or disrupt interaction between FtsZ and MreB. It consequently has effects on cell wall synthesis during cell division. However the mechanism of the inhibition remains unclear. We now plan to perform further biochemical and structural studies to verify and investigate the interaction and mechanism of the two natural products on the GTPase activity of FtsZ-Bs and FtsZ from other species.


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## APPENDIX A

## SEQUENCE

## Gene sequence of $m r e B-B s$

NCBI Reference Sequence: NC_000964.3
atgtttggaattggtgctagagaccttggtatagatcttggaactgcgaatacgcttgttttgtaaaaggaaaaggaattgttgt gagagagccgtcagttgtcgctttgcagacggatacgaaatcgattgtcgctgtcggaaatgatgcgaaaaatatgattgga cggacaccgggcaacgtggtggctcttcgcccgatgaaagacggcgttatcgctgattatgaaacaacggcgacgatgat gaaatattacatcaatcaggccataaaaaataaaggcatgtttgccagaaaaccatatgtaatggtatgtgtcccatcaggca ttacagctgttgaagaacgcgctgttatcgatgcgacaagacaggcgggagcgcgtgacgcgtatccgattgaagagcctt ttgccgcagcaatcggagccaatctgccagtttgggaaccgactggaagcatggttgttgatatcgggggcggtacgaca gaagttgcgattatttccetcggaggcatcgtaacgtctcagtcaatccgtgtagccggtgatgagatggatgacgegattat caactacatcagaaaaacgtacaatctgatgatcggtgaccgtacggctgaagcgattaaaatggaaatcggatctgcaga agctcctgaagaatccgacaacatggaaatccgcggccgcgatttgctcacaggtttgccgaaaacaattgaaattacagg aaaagagatttctaacgctctacgcgacactgtatctacaattgtcgaagcagtgaagagcacactcgaaaaaacaccgcct gagcttgcagcagatatcatggacagaggtatagtgttaaccggcggcggagcgctttgcgcaatttggacaaagtcatc agcgaagaaacaaaaatgccggtccttatcgccgaagatccgettgattgtgtagcgatcggaacagggaaagcactgga gcacatccatctttcaaagggaaaactagataa

## Protein sequence of MreB-Bs

mfgigardlgidlgtantlvfvkgkgivvrepsvvalqtdtksivavgndaknmigrtpgnvvalrpmkdgviadyett atmmkyyinqaiknkgmfarkpyvmvcvpsgitaveeravidatrqagardaypieepfaaaiganlpvweptgs mvvdigggttevaiislggivtsqsirvagdemddaiinyirktynlmigdrtaeaikmeigsaeapeesdnmeirgrd lltglpktieitgkeisnalrdtvstiveavkstlektppelaadimdrgivltgggallrnldkviseetkmpvliaedpldc vaigtgkalehihlfkgktr

## Gene sequence of ftsZ-Bs

NCBI Reference Sequence: NC_000964.3
atgttggagttcgaaacaaacatagacggcttagcatcaattaaagtaatcggagtaggaggcggcggtaacaacgccgtt aaccgaatgattgaaaatgaagtgcaaggagtagagtatatcgcggtaaacacggacgctcaagctcttaacctgtcaaaa gcagaagtgaaaatgcaaatcggcgcaaagctgactagaggattgggagcaggtgcgaatccggaagtcgggaaaaaa gccgctgaagaaagcaaagagcagattgaagaagcacttaaaggtgctgacatggtattcgtgacagctggtatgggcgg cggaacaggaacaggtgccgcaccggttcgcacaaatcgcgaaagacttaggcgcattaacagtcggcgttgtgacaa gaccgtttaccttcgaaggacgcaaaagacagcttcaggctgcaggcggaatctcggcaatgaaagaagcggtggatac actgatcgtgatcccgaacgaccgtatccttgaaattgttgataaaaacacaccgatgcttgaagcattccgcgaagcggat aacgtacttcgccaaggggttcaaggtatttctgacttgattgctacacctggtcttatcaaccttgactttgctgatgtgaaaac aatcatgtcaaacaaaggatctgctttgatgggtatcggtattgctactggggaaaatcgcgcggcagaggcagcaaaaaa agcaatttccagccegcttcttgaagcggccattgacggtgcgcaaggcgtcctcatgaacatcactggaggaacaaacct cagcctatatgaggttcaggaagcagcagacattgtcgcttcggcgtctgatcaagacgtaaacatgatttcggttctgttatt aatgaaaatctaaaagatgagattgtggtgacagtgattgcaaccggctttatcgaacaagagaaggacgtgacgaagcct cagcgtccaagcttaaatcaaagcatcaaaacacacaatcaaagtgttccgaagcgtgagccaaaacgtgaggaacctca gcagcagaacacagtaagccgtcatacttcacagccggctgatgatacgcttgacatcccgacattcttaagaaaccgtaat aaacgeggctaa

## Protein sequence of FtsZ-Bs

mlefetnidglasikvigvggggnnavnrmienevqgveyiavntdaqalnlskaevkmqigakltrglgaganpev gkkaaeeskeqieealkgadmvfvtagmgggtgtgaapviaqiakdlgaltvgvvtrpftfegrkrqlqaaggisamk eavdtlivipndrileivdkntpmleafreadnvlrqgvqgisdliatpglinldfadvktimsnkgsalmgigiatgenr aaeaakkaissplleaaidgaqgvlmnitggtnlslyevqeaadivasasdqdvnmifgsvinenlkdeivvtviatgfi eqekdvtkpqrpslnqsikthnqsvpkrepkreepqqqntvsrhtsqpaddtldiptflrnrnkrg

## Sequencing result

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## >141009-07_A10_pTG19-ftsZBs_M13F-pUC.ab1 1691

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## APPENDIX B

## PHOSPHATE RELEASE ASSAY

## Standard curve for free phosphate concentration

Table B1 Free phosphate concentration standard.


Table B2 Effect of pH on ATP hydrolysis of the proteins.

| pH | MreB | FtsZ | Lysozyme | BSA |
| :---: | :---: | :---: | :---: | :---: |
| pH 5.5 | 6.80 | -0.50 | 0.20 | 0.50 |
| pH 6.0 | 16.50 | -1.70 | 0.35 | 0.15 |
| pH 6.5 | 14.90 | -1.32 | -1.00 | 0.30 |
| pH 7.0 | 31.11 | -2.12 | -1.12 | -0.22 |
| pH 7.5 | 33.70 | -1.31 | -0.92 | 0.21 |
| pH 8.0 | 33.24 | 0.25 | -0.51 | -0.3 |

Not: $\mathrm{N}=10$

Table B3 Effect of pH on GTP hydrolysis of the proteins.

| pH | MreB | FtsZ | Lysozyme | BSA |
| :---: | :---: | :---: | :---: | :---: |
| pH 5.5 | 20.64 | 9.00 | 0.71 | 0.32 |
| pH 6.0 | 28.64 | 13.71 | -0.90 | 0.61 |
| pH 6.5 | 44.43 | 17.81 | -1.30 | -1.51 |
| pH 7.0 | 39.81 ายาลัยแาค11.15 ${ }^{\text {a }}$ |  | -0.25 | -1.09 |
| pH 7.5 | 25.77 | 9.76 | -0.91 | 1.05 |
| pH 8.0 | 25.26 | 11.42 | 0.40 | -1.43 |

Not: $\mathrm{N}=10$

## T-value Equation

$$
t=\frac{\bar{X}_{1}-\bar{X}_{2}}{\sqrt{\left(\frac{\left(N_{1}-1\right) s_{1}^{2}+\left(N_{2}-1\right) s_{2}^{2}}{N_{1}+N_{2}-2}\right)\left(\frac{1}{N_{1}}+\frac{1}{N_{2}}\right)}}
$$

Table B4 Screening of natural compounds effect on ATP hydrolysis by MreB-Bs.

| Sample | $\left[\mathrm{PO}_{4}{ }^{3-}\right](\boldsymbol{\mu M})$ | Standard deviation (SD) |
| :--- | :---: | :---: |
| MreB + ATP | 32.22 | 2.73 |
| MreB + ATP + Apigenin | 31.51 | 1.11 |
| MreB + ATP + Baicalein | 32.76 | 0.66 |
| MreB + ATP + Luteolin | 30.57 | 1.63 |
| MreB + ATP + Mangostin | 32.04 | 0.97 |
| MreB + ATP + Naringenin | 29.34 | 0.58 |
| BSA + ATP | 1.73 | 0.96 |
| Lysozyme + ATP | 0.03 | 2.19 |

Not: $\mathrm{N}=10$

## Statistics

MreB + ATP + Apigenin

| MreB + ATP + Baicalein | The result is not significant at $\mathrm{p}<0.05$ |
| :--- | :--- |
| MreB + ATP + Luteolin | The result is not significant at $\mathrm{p}<0.05$ |
| The t-value is -0.33284. The p -value is 0.377986 |  |
| MreB + ATP + Mangostin | The result is not significant at $\mathrm{p}<0.05$ |
|  | The t -value is 0.10761 . The p-value is 0.459744 |
|  | The result is not significant at $\mathrm{p}<0.05$ |

The result is not significant at $\mathrm{p}<0.05$

The result is not significant at $\mathrm{p}<0.05$
The t -value is -0.33284 . The p -value is 0.377986
The result is not significant at $\mathrm{p}<0.05$

The result is not significant at $\mathrm{p}<0.05$

Table B5 Screening of natural compounds effect on GTP hydrolysis by MreB-Bs.

| Sample | $\left[\mathbf{P O}_{4}{ }^{\mathbf{3}}\right](\boldsymbol{\mu M})$ | Standard deviation (SD) |
| :--- | :---: | :---: |
| MreB + GTP | 38.37 | 2.42 |
| MreB + GTP + Apigenin | 36.38 | 1.31 |
| MreB + GTP + Baicalein | 35.51 | 0.95 |
| MreB + GTP + Luteolin | 37.12 | 0.95 |
| MreB + GTP + Mangostin | 35.74 | 1.33 |
| MreB + GTP + Naringenin | 38.93 | 1.05 |
| BSA + GTP | -1.50 | 0.39 |
| Lysozyme + GTP | -1.30 | 1.36 |

Not: $\mathrm{N}=10$

## Statistics

| MreB + GTP + Apigenin | The t-value is 1.5872 . The p-value is 0.09383 |
| :--- | :--- |
| MreB + GTP + Baicalein | The result is not significant at $\mathrm{p}<0.05$ |
| MreB + GTP + Luteolin | The result is not significant at $\mathrm{p}<0.05$ |
| MreB + GTP + Mangostin is 0.88195 . The p-value is 0.213807 |  |
|  | The result is not significant at $\mathrm{p}<0.05$ |
| MreB + GTP + Naringenin | The result is not significant at $\mathrm{p}<0.05$ |
|  | The t -value is -0.12525. The p -value is 0.453186 |

Table B6 Screening of natural compounds effect on GTP hydrolysis by FtsZ-Bs.

| Sample | $\left[\mathbf{P O}_{\mathbf{4}}{ }^{3}\right](\boldsymbol{\mu M})$ | Standard deviation (SD) |
| :--- | :---: | :---: |
| FtsZ + GTP* | 12.56 | 1.19 |
| FtsZ + GTP + Apigenin* | 8.38 | 0.67 |
| FtsZ + GTP + Baicalein* | 7.22 | 1.21 |
| FtsZ + GTP + Luteolin | 13.25 | 0.58 |
| FtsZ + GTP + Mangostin | 12.78 | 0.92 |
| FtsZ + GTP + Naringenin | 12.72 | 0.90 |
| BSA + GTP | -1.50 | 0.39 |
| Lysozyme + GTP | -1.30 | 1.36 |
| Not: N=10 |  |  |

## Statistics

| FtsZ + GTP + Apigenin | The t -value is 5.29673 . The p -value is 0.00305 |
| :--- | :--- |
| FtsZ + GTP + Baicalein | The result is significant at $\mathrm{p}<0.01$ <br> The t -value is 5.43484 . The p -value is 0.002781 <br> The result is significant at $\mathrm{p}<0.01$ |
| FtsZ + GTP + Luteolin | The t -value is -0.89731. The p-value is 0.21014 |
| The result is not significant at $\mathrm{p}<0.05$ |  |
| FtsZ + GTP + Mangostin | The t -value is -0.24936. The p -value is 0.407682 <br> The result is not significant at $\mathrm{p}<0.05$ |
| FtsZ + GTP + Naringenin | The t -value is -0.18151. The p -value is 0.432396 |

## APPENDIX C

## CHEMICAL PREPARATIONS

## $10 \mathrm{~N} \mathbf{N a O H}(200 \mathrm{ml})$

Dissolve 80 g NaOH in a final volume of $200 \mathrm{ml} \mathrm{dH}_{2} \mathrm{O}$.

## 1 M Tris-HCl (500 ml)

1. Dissolve 60.55 g Tris base in 300 ml of $\mathrm{dH}_{2} \mathrm{O}$.
2. Adjust the pH to the desired value with concentrated HCl .
3. Bring up the volume to 500 ml with $\mathrm{dH}_{2} \mathrm{O}$.

## 1 M Sodium acetate ( $\mathbf{2 0 0} \mathbf{~ m l}$ )

1. Dissolve 27.22 g sodium acetate in 100 ml of $\mathrm{dH}_{2} \mathrm{O}$.
2. Add 6 ml of glacial acetic acid.
3. Adjust the pH to the desired value with 10 NaOH .
4. Bring up the volume to 200 ml with $\mathrm{dH}_{2} \mathrm{O}$.

### 0.5 M MES ( $\mathbf{2 0 0} \mathbf{~ m l}$ )

1. Dissolve 19.52 g HEPES (free acid) in 100 ml of $\mathrm{dH}_{2} \mathrm{O}$.
2. Adjust the pH to the desired value with 10 N NaOH .
3. Bring up the volume to 200 ml with $\mathrm{dH}_{2} \mathrm{O}$.

## 1 M HEPES (200 ml)

1. Dissolve 41.66 g HEPES (free acid) in 100 ml of $\mathrm{dH}_{2} \mathrm{O}$.
2. Adjust the pH to the desired value with 10 N NaOH .
3. Bring up the volume to 200 ml with $\mathrm{dH}_{2} \mathrm{O}$.

### 0.5 M EDTA, pH 8 ( $\mathbf{5 0 0} \mathbf{~ m l}$ )

1. Resuspend $93.05 \mathrm{~g} \mathrm{Na}_{2} \cdot E D T A \cdot 2 \mathrm{H}_{2} \mathrm{O}$ (disodium dihydrate) in about 400 ml of $\mathrm{dH}_{2} \mathrm{O}$.
2. Add about 9 g solid NaOH .
3. Once all the NaOH dissolves, slowly adjust the pH with 10 N NaOH .
4. Bring up the volume to 500 ml with dH 2 O .

Note: EDTA will not completely dissolve until the pH reaches 8 .

## 50X TAE buffer (1 L)

Tris base
Glacial acetic acid
0.5 M EDTA ( pH 8.0 ) $\quad 100 \mathrm{ml}$

Bring up the volume to 1 L .

## 6X DNA loading sample buffer ( $\mathbf{1 0} \mathbf{~ m l}$ )

Glycerol 3 g
Bromophenol blue $\quad 0.025 \mathrm{~g}$
Xylene cyanol FF $\quad 0.025 \mathrm{~g}$
Bring up the volume to 10 ml and store at $4^{\circ} \mathrm{C}$.

## SDS-PAGE preparation (30\% gel)

Acrylamide gel solution (100 ml)

| acrylamide | 29.4 g |
| :--- | :--- |
| bis-acrylamide | 0.6 g |

Separating gel ( $15 \%$ gel)

| $\mathrm{dH}_{2} \mathrm{O}$ | 2.9 ml |
| :--- | :--- |
| 2 M Tris- $\mathrm{HCl}, \mathrm{pH} 8.8$ | 2 ml |
| $10 \%$ SDS | 0.1 ml |
| $30 \%$ acrylamide gel solution | 5 ml |
| $10 \%\left(\mathrm{NH}_{4}\right)_{2} \mathrm{~S}_{2} \mathrm{O}_{8}$ | $50 \mu \mathrm{l}$ |
| TEMED | $5 \mu \mathrm{l}$ |

Stacking gel (4\% gel)

| $\mathrm{dH}_{2} \mathrm{O}$ | 6.1 ml |
| :--- | ---: |
| 0.5 M Tris- $\mathrm{HCl}, \mathrm{pH} 6.8$ | 2.5 ml |

$10 \%$ SDS
0.1 ml
$30 \%$ acrylamide gel solution
1.3 ml
$10 \%\left(\mathrm{NH}_{4}\right)_{2} \mathrm{~S}_{2} \mathrm{O}_{8}$
$50 \mu 1$
TEMED
$10 \mu 1$

## 10X Running buffer ( $\mathbf{1} \mathbf{L}$ )

Tris base $\quad 30 \mathrm{~g}$
Glycine $\quad 144 \mathrm{~g}$
SDS $\quad 10 \mathrm{~g}$
Bring up the volume to 1 L .

## 5X Sample buffer

| SDS | 1.0 g |
| :--- | :--- |
| Glycerol | 5.0 ml |
| Bromophenol blue | 25 mg |
| Tris base | 242 mg |
| HCl | 0.35 ml (adjust the pH to 6.8 ) |
| 2-Mercapoethanol | 1.0 ml |
| Bring up the volume to | 10 ml and store at $4^{\circ} \mathrm{C}$. |

## Coomassie blue stain (1 L)

Methanol
500 ml
Acetic acid 100 ml

Coomassie blue 0.5 g
$\mathrm{dH}_{2} \mathrm{O}$
400 ml
Mix on stir plate until all coomassie blue is dissolved.


## De-stain (1 L)

| Methanol | 400 ml |
| :--- | :--- |
| Acetic acid | 100 ml |

Bring up the volume to 1 L .

10\% SDS ( 100 ml )
10 g SDS into 100 ml , heat to $68^{\circ} \mathrm{C}$ for solubility, $\mathrm{pH} \sim 6.6$.

## $100 \mathrm{mg} / \mathrm{ml}$ ampicillin ( 10 ml )

1. Weigh 1 g of ampicillin.
2. Bring up the volume to 10 ml and filter sterilizes $(0.22 \mu \mathrm{~m})$, store at $-20^{\circ} \mathrm{C}$.

## 1 M IPTG ( 10 ml )

1. Weigh 2.38 g of $\operatorname{IPTG}(\mathrm{MW}=238.3 \mathrm{~g} / \mathrm{mol})$.
2. Bring up the volume to 10 ml and filter sterilizes $(0.22 \mu \mathrm{~m})$, store at $-20^{\circ} \mathrm{C}$.

## $20 \mathrm{mg} / \mathrm{ml}$ X-gal (1 ml)

1. Weigh 20 mg of X -gal.
2. Bring up the volume to 1 ml with $100 \%$ DMF (dimethylformamide), store the stock solution at $-20^{\circ} \mathrm{C}$ in the dark. Discard the stock solution if the color changes significantly.

## LB broth ( $\mathbf{1}$ L)

Tryptone
Yeast extract 5 g
$\mathrm{NaCl} \quad 5 \mathrm{~g}$

Dissolve components in 1 L of $\mathrm{dH}_{2} \mathrm{O}$ and sterilize by autoclaving at 15 psi , at $121{ }^{\circ} \mathrm{C}$ for 15 minutes.

## LB agar ( $\mathbf{2 0 0} \mathbf{~ m l}$ )

Tryptone 2 g
Yeast extract $\quad 1 \mathrm{~g}$
$\mathrm{NaCl} \quad 1 \mathrm{~g}$
Agar $\quad 4 \mathrm{~g}$
Dissolve components in 200 ml of $\mathrm{dH}_{2} \mathrm{O}$ and sterilize by autoclaving at 15 psi , at $121^{\circ} \mathrm{C}$ for 15 minutes.

## Blue-white selection LB agar plate

1. Melt LB agar, and allow media to cool to $55-60^{\circ} \mathrm{C}$. Add ampicillin to final concentration of $100 \mu \mathrm{~g} / \mathrm{ml}$.
2. Gently swirl the flask to mix the ampicillin into the agar.
3. Pour a thin layer of LB agar $\sim 10 \mathrm{ml}$ into each plate.
4. Let each plate cool until its solid $\sim 20$ minutes.
5. Spread $40 \mu \mathrm{l}$ of IPTG on top of the plate with spreader, let the plates dry in laminar flow.
6. Spread $40 \mu 1$ of X-gal on top of the plate with a hockey stick spreader. This should take 30 minutes or so if the plate is dry.

## $\mathrm{Ni}-\mathrm{NTA}$ resin regeneration

1. Put all of the resin into a big column.
2. Wash with 3 CV water.
3. Wash with $3-5 \mathrm{CV}$ of 0.5 M NaOH .
4. Wash with water 5 CV make sure pH turn back to $6-7$.
5. Wash with 6 M guanidine hydrochloride and 25 mM imidazole, 2 CV each.
6. Wash with 5 CV water immediately.
7. Wash with 2 CV 100 mM EDTA.
8. Wash with more than 5 CV water.
9. Re-charge the resin with $2-3 \mathrm{CV} 100 \mathrm{mM} \mathrm{NiSO} 44$ and keep resin in buffer for a couple hours with shaking.
10. Wash the resin with more than 10 CV water.
11. Wash with 2 CV 20\% ethanol.
12. Wash with 2 CV 50\% ethanol.
13. Wash with 2 CV 70\% ethanol.
14. Wash with 2 CV 50\% ethanol.
15. Wash with 2 CV 20\% ethanol.

Stock the resin in 20\% ethanol in 4 degree.

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## PUBLICATION

Sunaree Choknud and Sakesit Chumnarnsilpa (2016). Screening of natural products effect on GTP hydrolysis of Bacillus subtilis FtsZ. The 11 ${ }^{\text {th }}$ international Symposium of the Protein Society of Thailand. August 3-5, Bangkok, Thailand: 215-221.

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