

Graduate Student Recruitment and Training Support

Report for

One Ajahn, One Project

May 2003 through April 2004

Academic Year 2546

Associate Professor Dr. Kenneth J. Haller
School of Chemistry
Institute of Science
Suranaree University of Technology
Nakhon Ratchasima 30000

This is how your work will appear to the public on the World Wide Web and in the printed book of abstracts.

Precipitation of arsenate with copper(II) in the presence of cationic polyelectrolyte

Kenneth J. Haller¹, Preeyaporn Pookrod¹, and John F Scamehorn². (1) School of Chemistry, Suranaree University of Technology, Nakhon Ratchasima 30000, Thailand, Fax: (66-2) 280-1010, haller@ccs.sut.ac.th, (2) Institute for Applied Surfactant Research, University of Oklahoma

Polyelectrolyte-enhanced ultrafiltration (PEUF) can achieve 99.95% removal of arsenate from solutions with initial arsenate concentration of 100 ppb, thereby easily meeting the new 10 ppb drinking water standard. For the process to be economically competitive, the polymer must be recovered from the retentate for recycle to the PEUF feed. One promising method for recovering the polyelectrolyte is addition of divalent metal ion (eg. Cu^{2+}) to precipitate the arsenate ions. The adsorption of poly(diallyldimethylammonium chloride) (QUAT) on $\text{NaCu}_6(\text{AsO}_4)_4\text{Cl}\cdot 4\text{H}_2\text{O}$ particles increases with increasing polymer concentration, decreasing ionic strength, and decreasing copper to arsenate ratio. The average particle size decreases with increasing polymer concentration, increasing salt concentration, decreasing temperature, and decreasing copper to arsenate concentration ratio. The sedimentation rate of the crystals increases with increasing polymer concentration, increasing electrolyte concentration, and increasing temperature. The viscosity of the supernatant solution increases with increasing polymer concentration, decreasing ionic strength, decreasing temperature, and decreasing copper to arsenate concentration ratio.

not yet rated

Abstract ID#: 715984

Password: 282347

Program Selection: Division of Colloid and Surface Chemistry

Topic Selection: Environmental Colloids